



Contribution ID: 97

Type: Oral

## An innovative application of magnetic field for CO<sub>2</sub> hydrogenation reaction on Fe and Cu supported MCM-41 catalyst

*Monday 28 November 2016 13:50 (15 minutes)*

The external magnetic field was applied in a packed-bed reactor based on the concepts of green and sustainable production of alternative fuels through CO<sub>2</sub> hydrogenation reaction. Accordingly, the roles of magnetic flux density and magnetic field direction on the performance of Fe-Cu/MCM-41 catalyst with intrinsic magnetic property were investigated and compared to that of without magnetic field. It was found that magnetic field strongly affected the activity of catalyst, both CO<sub>2</sub> conversion and product selectivity. Over 10Cu–10Fe/MCM-41 catalyst, magnetic field remarkably promoted CO<sub>2</sub> conversion, especially in the north-to-south (N-S) direction (1.8 times higher than that of without magnetic field at 260°C). With increasing magnetic flux density, CO<sub>2</sub> conversion was increased followed the order of 27.7 mT > 20.8 mT > 0 mT in each magnetic field direction. Moreover, under the magnetic field conditions which gave the highest CO<sub>2</sub> conversion, it was more favorable for CH<sub>3</sub>OH formation. CH<sub>3</sub>OH space time yield with magnetic field was 1.5 times higher than that of without magnetic field. The improvement of catalytic activity by the magnetic field application was described by mean of the reduction of apparent activation energy ( $E_a$ ). With magnetic field, the apparent activation energy was decreased for approximately 1.18 times compared to that of without magnetic field. This outstanding performance was attributed to the fact that magnetic field could facilitate the adsorption ability of reactant gases on magnetized catalyst surfaces, leading to the increase of catalytic CO<sub>2</sub> hydrogenation and selective conversion to CH<sub>3</sub>OH, and lowering of the activation energy.

**Author:** Ms KIATPHUENGORN, Sirapassorn (Nanomaterials for Energy and Catalysis Laboratory, National Nanotechnology Center (NANOTEC), National Science and Technology Development Agency (NSTDA))

**Presenter:** Ms KIATPHUENGORN, Sirapassorn (Nanomaterials for Energy and Catalysis Laboratory, National Nanotechnology Center (NANOTEC), National Science and Technology Development Agency (NSTDA))

**Session Classification:** Heron 2

**Track Classification:** Environmental nanotechnology