

# Detectors for Measurement of Charged Particles in High Energy/Nuclear Physics

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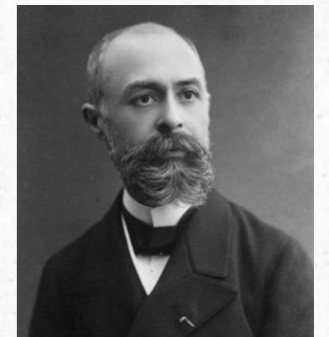
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# Radiation and Nuclear Physics

- History
  - Discovery of radiation and radioactive material
    - Wilhelm Conrad Röntgen
    - Antoine Henri Becquerel
    - Maria Salomea Skłodowska-Curie (Marie Curie)
    - Pierre Curie
  - Hierarchy of material is recognized
    - Atom
    - Nucleus and electron
      - Stable and unstable nucleus
    - proton and neutron
    - quark and lepton

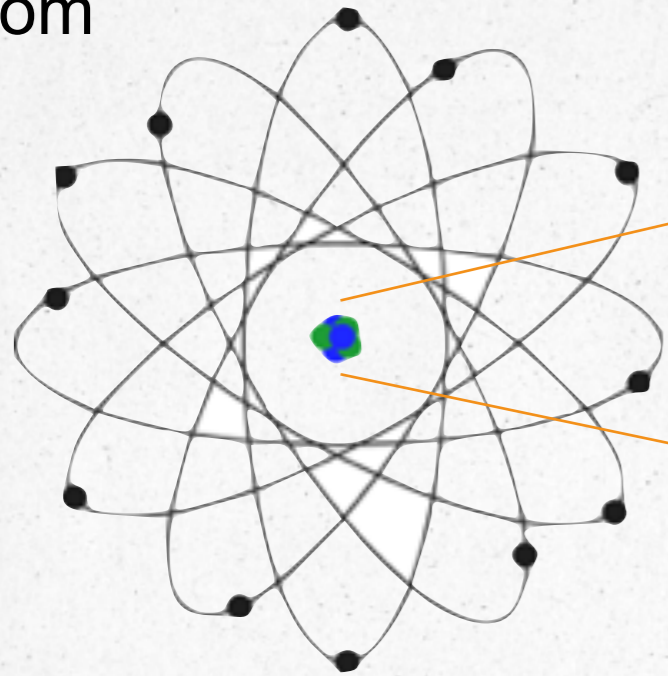


Photos: wikimedia commons



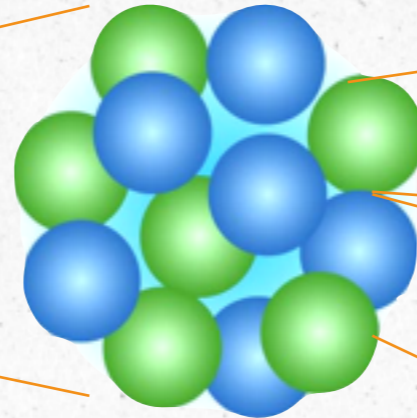
# Atom, Nucleus, Elementally Particle

Atom

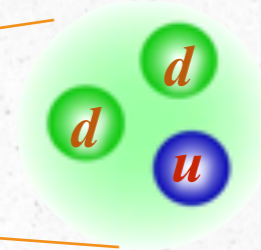


electron

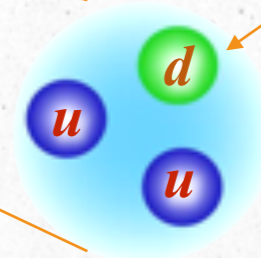
Nucleus



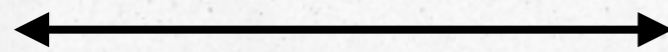
Neutron



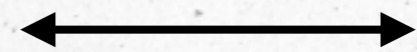
Quark



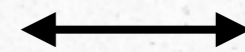
Proton



$\sim 10^{-10}$  [m]



$\sim 10^{-14}$  [m]



$\sim 10^{-15}$  [m]



# Radiation

- $\alpha$ -ray
  - ${}^4\text{He}$  nucleus

Charged particle

- $\beta$ -ray
  - electron and positron (anti-particle of electron)

- X-ray,  $\gamma$ -ray

- high-energy electromagnetic radiation

- X-ray: from transition of electron between different orbits

- $\gamma$ -ray: from nucleus

- neutron

- nucleus consists from proton and neutron



# Particle Physics

- Elementally particle physics and nuclear physics
  - Physicist wants to know phenomena of nature
  - Physics in small size
    - for example, how to probe inside of atom
    - Discovery of electron (J.J. Thomson)
      - Atom is NOT elementally particle
      - Existence of nucleus (positive charged particle)
      - What is structure of atom?



# Models of Atom Structure

In 1904, J.J. Thomson's model

Plum pudding model :

+/- particles are distributed evenly



Photo: Wikimedia commons

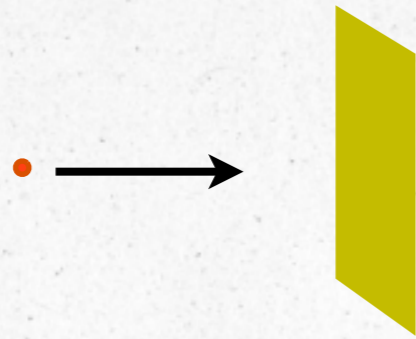
In 1903, Hantaro Nagaoka

Saturn-like model:

+ particle is solidified in the center  
and electrons are turning around



# Rutherford scattering

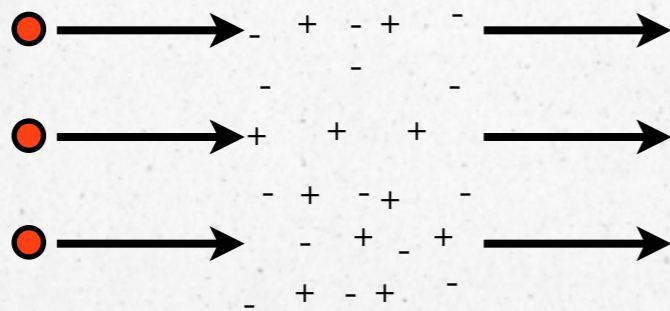


Hits  $\alpha$ -ray ( ${}^4\text{He}$  nucleus) to a thin gold film

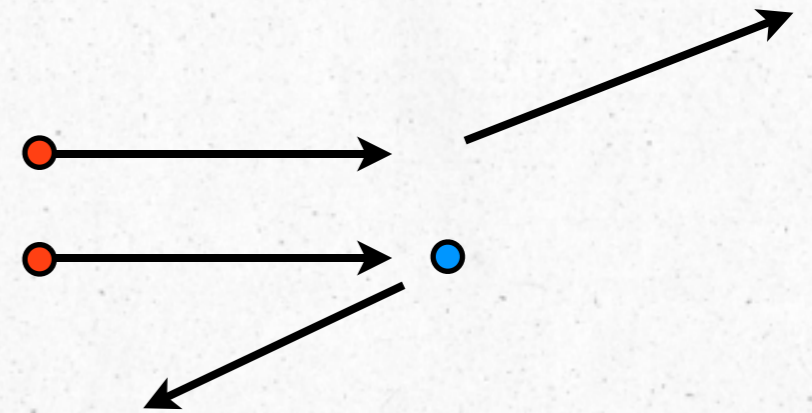
In 1911, Ernest Rutherford's Lab group  
Assistant, Johannes Wilhelm Geiger  
Students: Ernest Marsden

## Plum Pudding Model

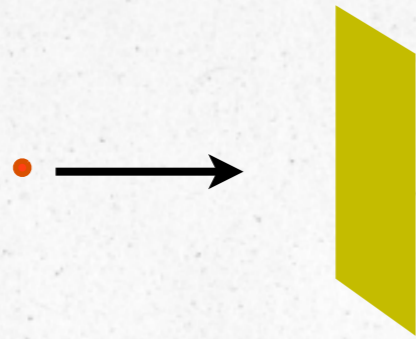
$\alpha$ -ray pass thorough the film



## Saturn-like Model



# Rutherford scattering

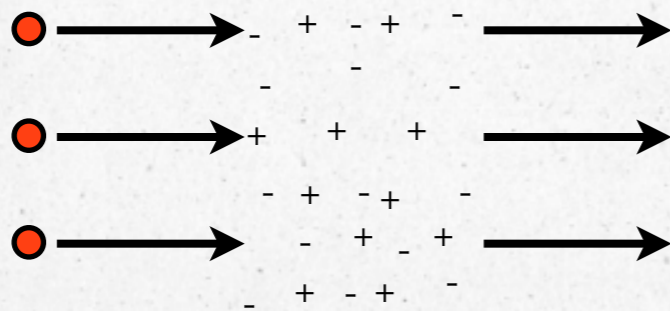


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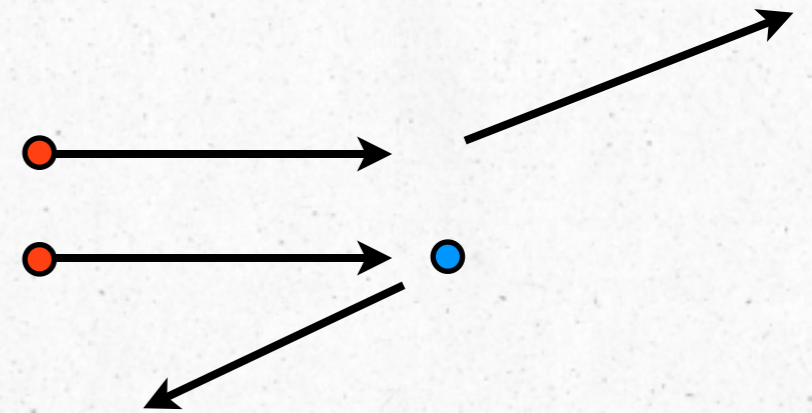
## Plum Pudding Model

$\alpha$ -ray pass thorough the film



Experiment support this result

## Saturn-like Model





# To Know Structures in Small World

- Collide elementary particle or nucleus to a target
  - Particle beam and fixed target
  - Collider
- After collision
  - Scattering of particles
  - Emission from inside of target particle
  - Particles newly created
- Detectors to measure and identify
  - Momentum (mass  $\times$  velocity)
  - Velocity (Time of flight and flight length)
  - Particle species



# Units

- In field of elementals particle/Nuclear physics
  - Energy: eV (electron volt)
    - 1 [eV] =  $1.6 \times 10^{-19}$  [J]
  - Length: fm (femto meter)
    - 1 [fm] =  $10^{-15}$  [m] =  $10^{-13}$  [cm]
  - Cross section: b (barn)
    - 1 [barm] = 100 [fm<sup>2</sup>] =  $10^{-28}$  [m<sup>2</sup>]
- Example
  - Proton
    - mass: 0.938 GeV/c<sup>2</sup>, electrical radius ~0.9 fm
      - E= mc<sup>2</sup>, c: light velocity
  - Electron
    - mass: 0.511 MeV/c<sup>2</sup> (~ 1/1836 of proton mass)



# Measurement Principle of Charged Particles



# Passage of Particles Through Matter

- Charged particle
  - Energy loss with electrons in matter
  - Bremsstrahlung radiation
  - Synchrotron radiation
  - Cherenkov radiation
  - Knock-on electron ( $\delta$ -ray)
- X-ray,  $\gamma$ -ray
  - Photoelectric effect
  - Compton scattering
  - Pair production of electron and positron
- Neutron
  - Interaction with nucleus and emission of proton or  $\gamma$ -ray



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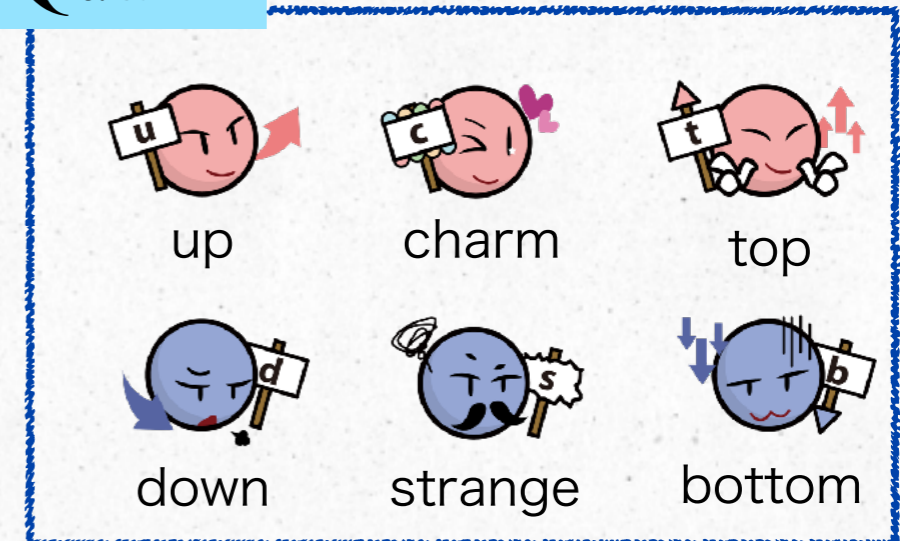


# Charged Particles

- Nucleus
- Nucleon: component of nucleus
  - proton (positive charge) and neutron (neutral)
- Family of Hadrons
- Hadron
  - Particle consisted of quarks
  - Meson (quark and anti-quark)
  - Baryon (three quarks)
- Electron, muon, and tau

Table of elementary particles

## Quark



## Lepton

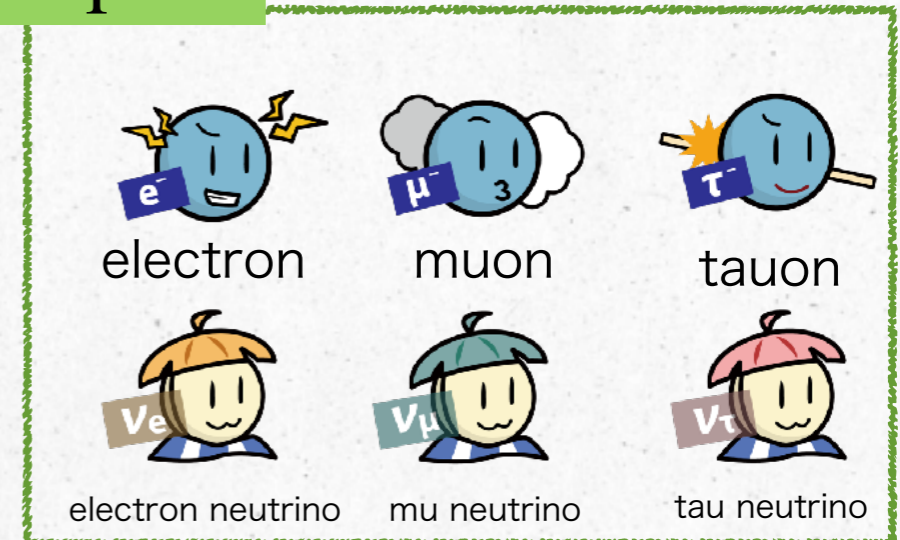
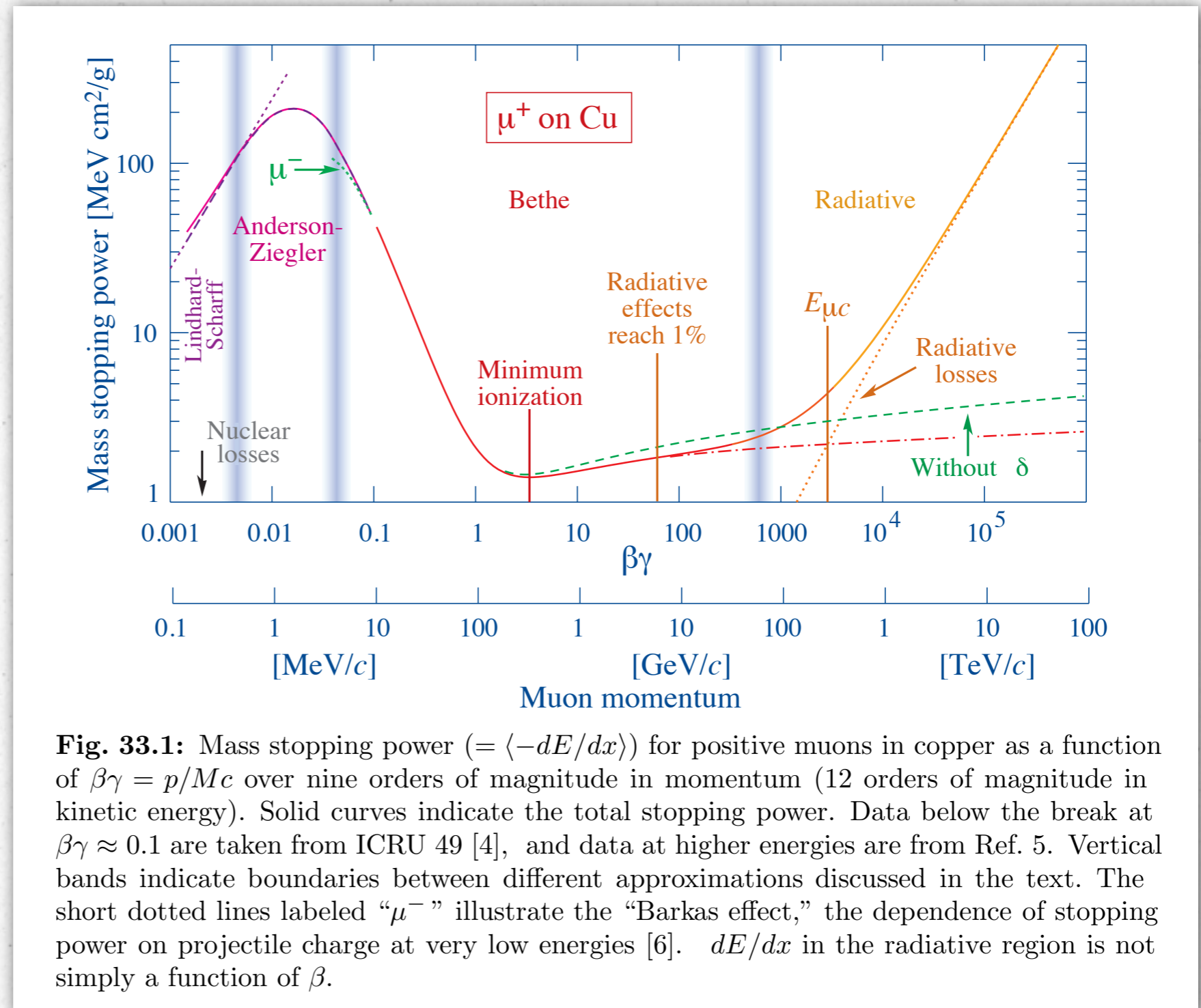


Illustration: Yuki Akimoto, <https://higgstan.com/>



# Interaction of Charged Particle with Matter

- Several processes
- Energy dependences



Ref.: M. Tanabashi et al. (Particle Data Group), Phys. Rev. D 98, 030001 (2018).  
 available on the PDG WWW page (<http://pdg.lbl.gov>)



# Energy Loss

- When the charged particles pass through the matter, its energy loses gradually by ionizing electrons
- Bethe-Bloch equation

$$\left\langle -\frac{dE}{dx} \right\rangle = K z^2 \frac{Z}{A} \frac{1}{\beta^2} \left[ \frac{1}{2} \ln \frac{2m_e c^2 \beta^2 \gamma^2 W_{\max}}{I^2} - \beta^2 - \frac{\delta(\beta\gamma)}{2} \right]$$

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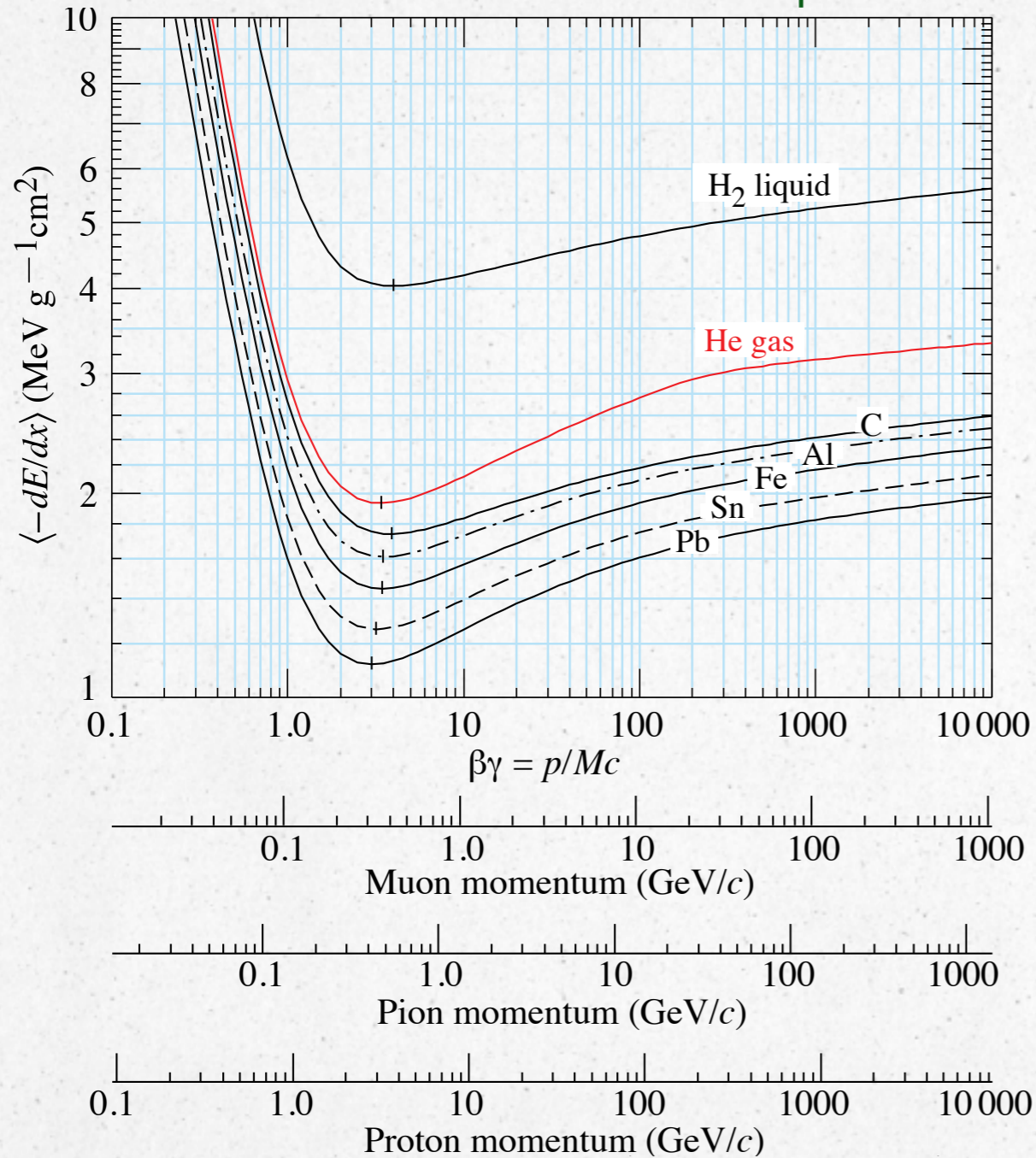
Capital characters: parameters related Matter  
lower characters: related incident particle

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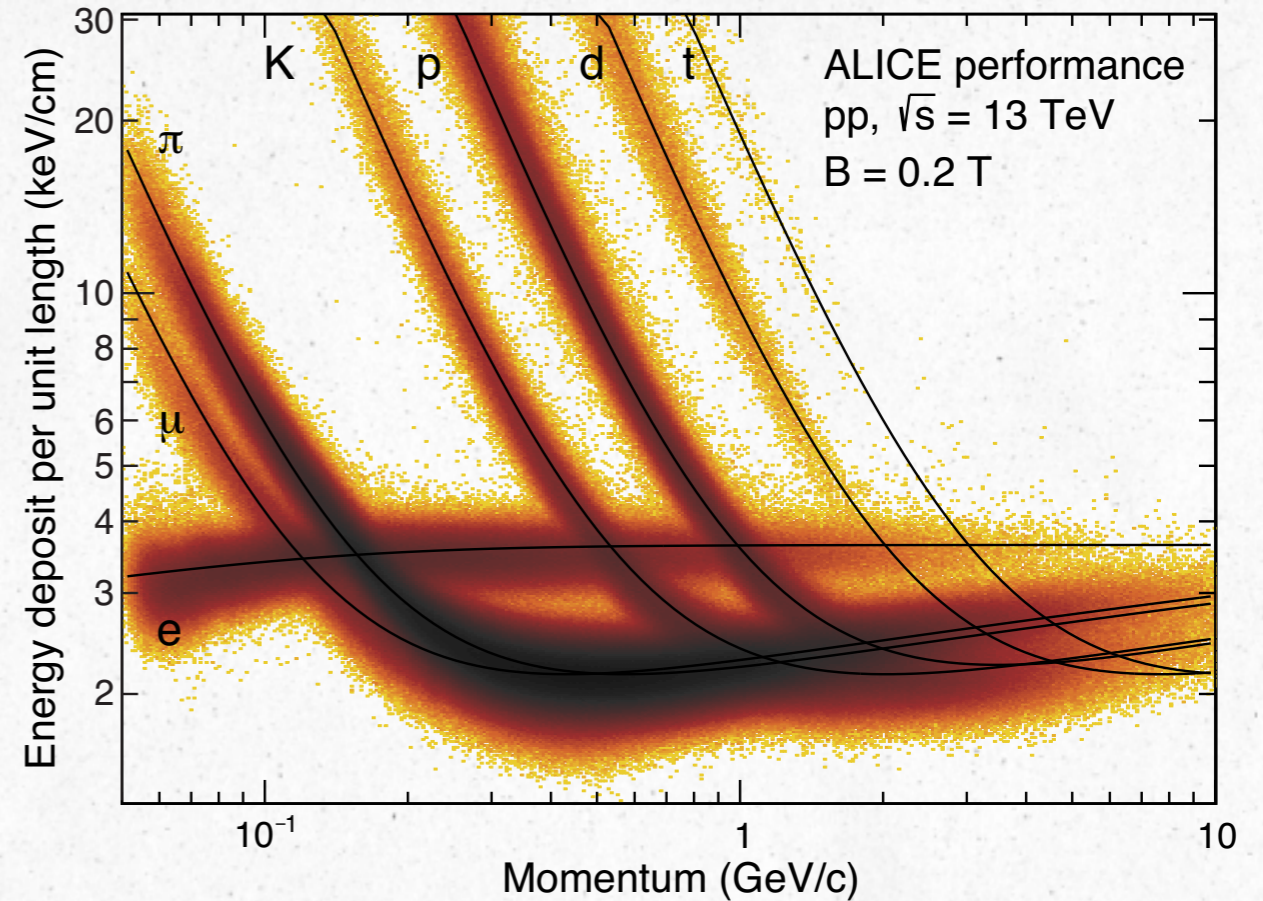
# Energy Loss

Calculation from the equation



**Figure 33.2:** Mean energy loss rate in liquid (bubble chamber) hydrogen, gaseous helium, carbon, aluminum, iron, tin, and lead. Radiative effects, relevant for muons and pions, are not included. These become significant for muons in iron for  $\beta\gamma \gtrsim 1000$ , and at lower momenta for muons in higher- $Z$  absorbers. See Fig. 33.23.

Measurement results of an experiment



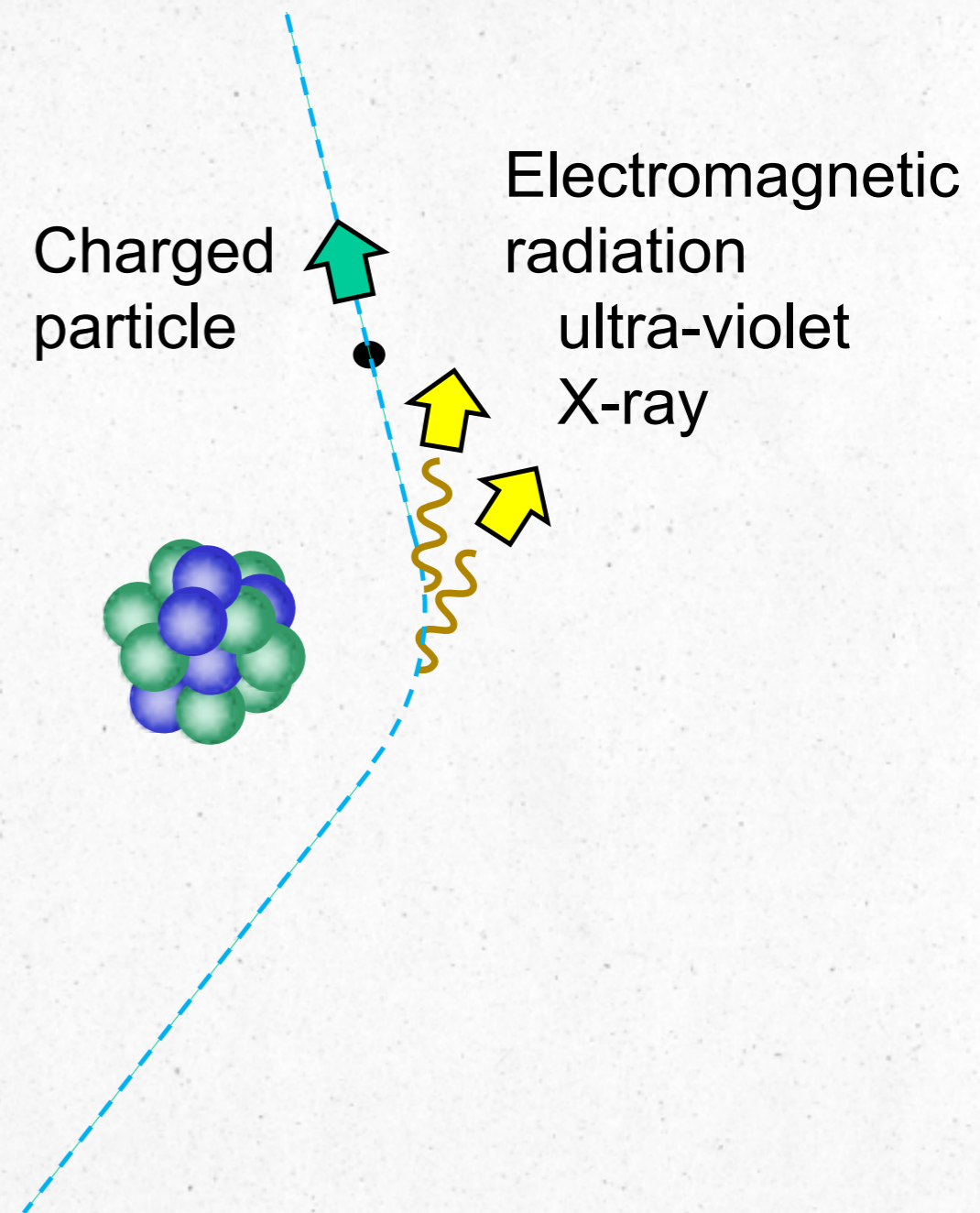
**Figure 34.15:** Energy deposit versus momentum measured in the ALICE TPC [111].

Ref.: M. Tanabashi et al. (Particle Data Group), Phys. Rev. D 98, 030001 (2018). available on the PDG WWW page (<http://pdg.lbl.gov>)



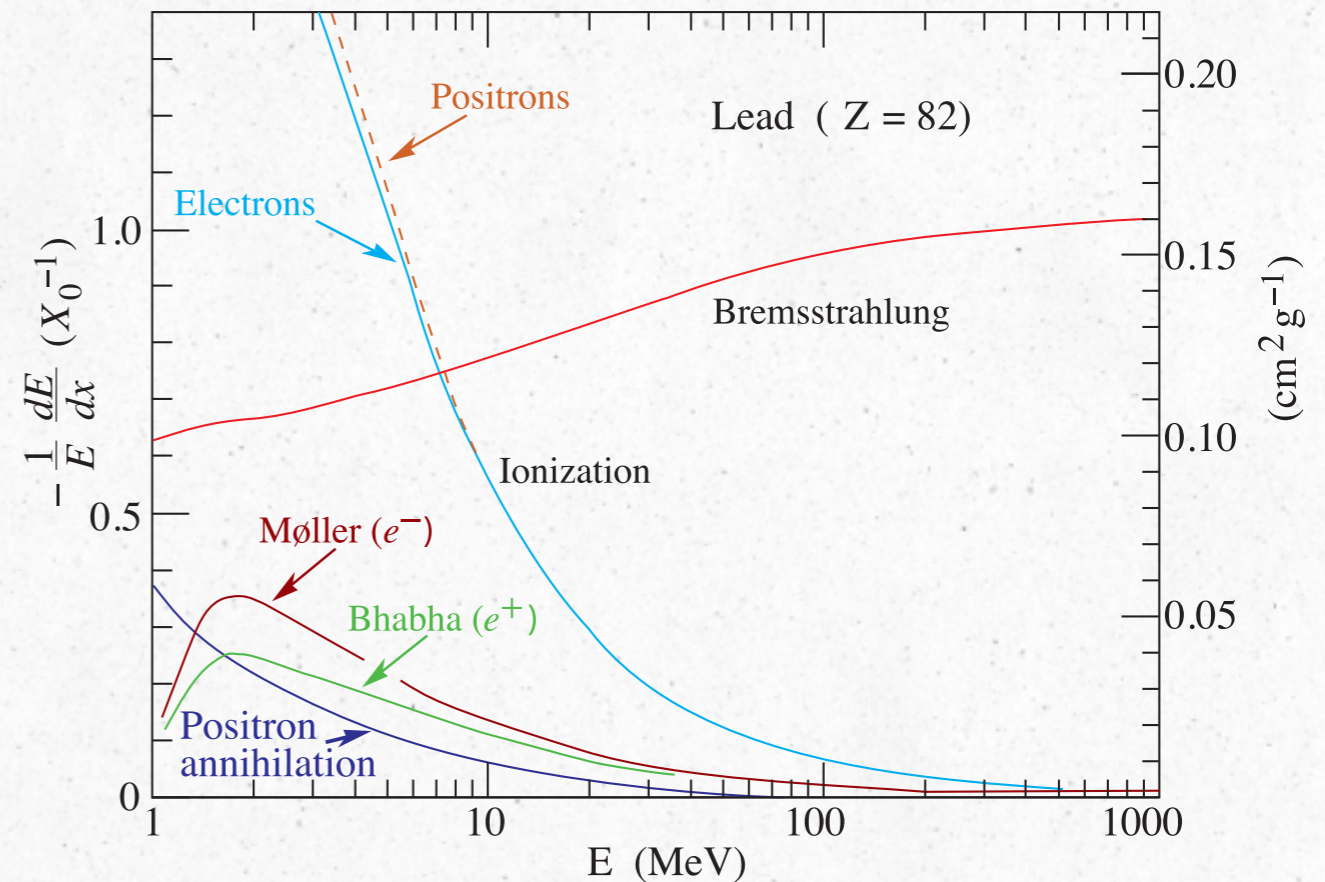
# Bremsstrahlung Radiation

- Charged particle is bended by acceleration/ deceleration from strong electro-magnetic field by nucleus
- Charged particle emits part of the energy as electromagnetic waves



# Bremsstrahlung Radiation

- Dominant process for  $e^+/e^-$  (in  $E > \sim 10$  MeV)
- One of processes in Electro-Magnetic shower



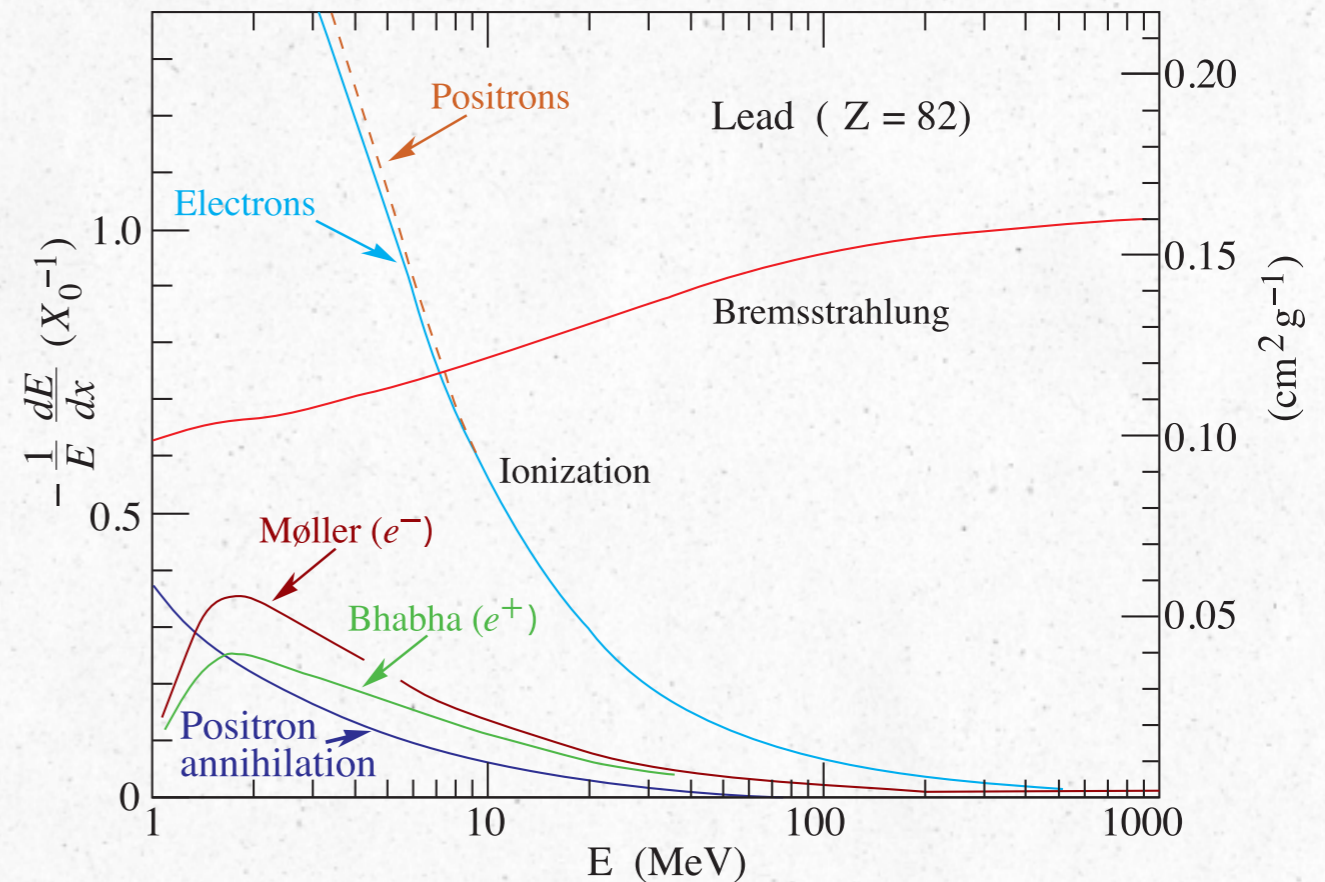
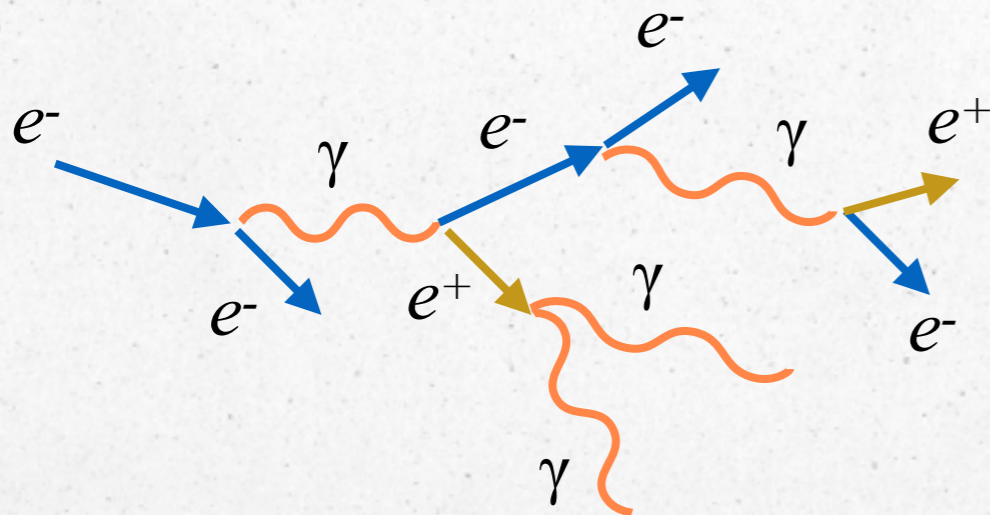
**Figure 33.11:** Fractional energy loss per radiation length in lead as a function of electron or positron energy. Electron (positron) scattering is considered as ionization when the energy loss per collision is below 0.255 MeV, and as Møller (Bhabha) scattering when it is above. Adapted from Fig. 3.2 from Messel and Crawford, *Electron-Photon Shower Distribution Function Tables for Lead, Copper, and Air Absorbers*, Pergamon Press, 1970. Messel and Crawford use  $X_0(\text{Pb}) = 5.82 \text{ g/cm}^2$ , but we have modified the figures to reflect the value given in the Table of Atomic and Nuclear Properties of Materials ( $X_0(\text{Pb}) = 6.37 \text{ g/cm}^2$ ).

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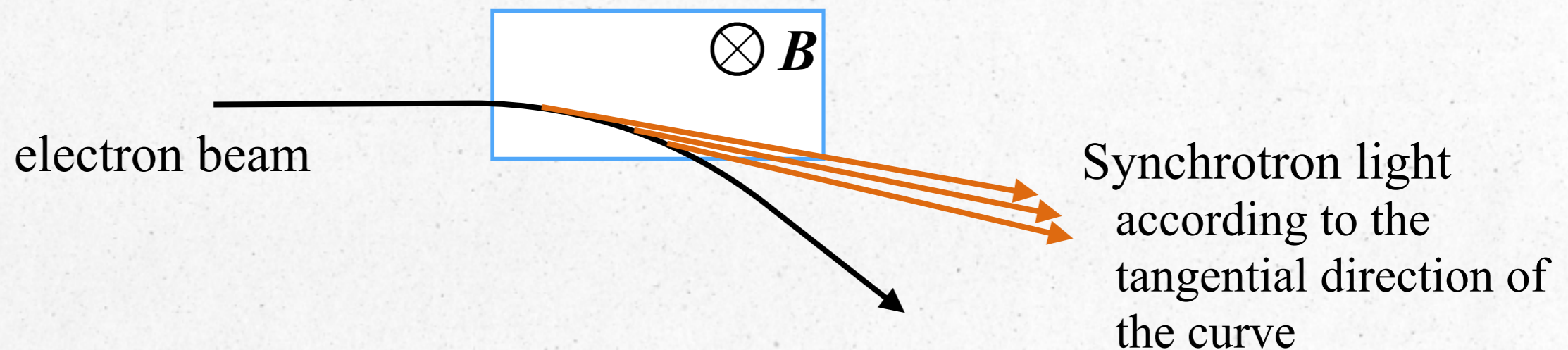
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# Synchrotron Radiation

- Light was observed from the electron's orbital plane on bending the electron by magnetic field
- As with bremsstrahlung, electromagnetic waves are generated by bending charged particles in a magnetic field
- Infrared light to X-rays
  - Tool to know material structure





# Cherenkov Radiation

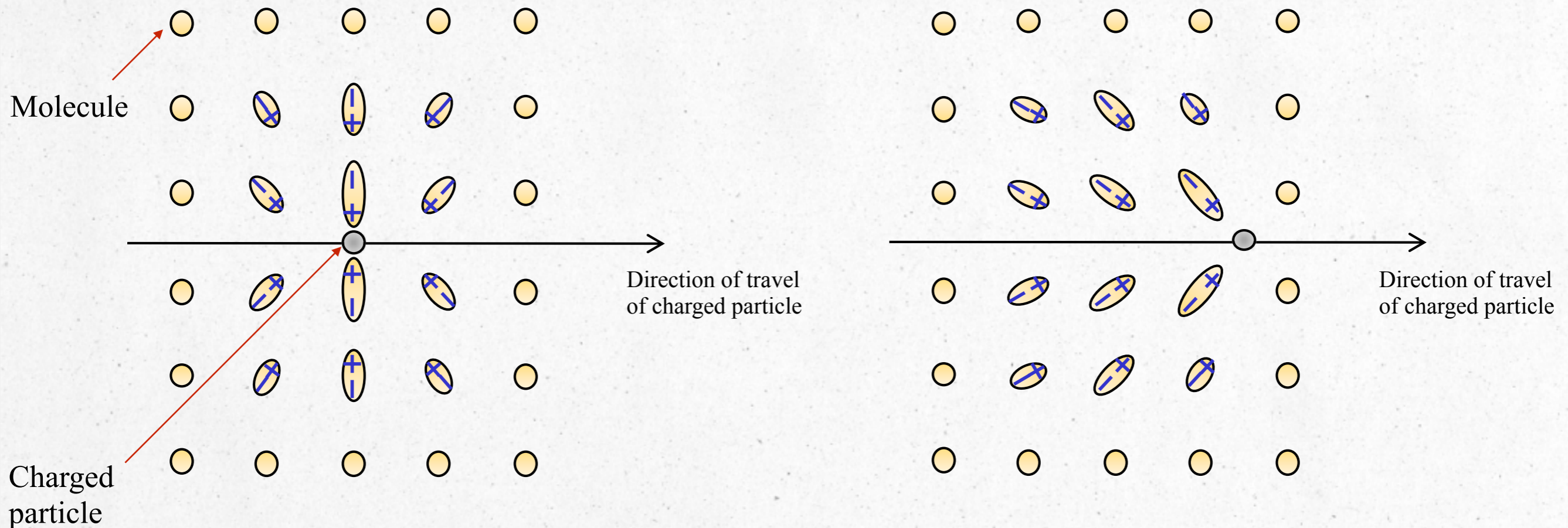
- Radiation of particle of which speed is faster than light velocity in a material
  - Velocity of light:  $c/n$ 
    - $c$ : Speed of light in vacuum
    - $n$ : Refractive index of material
- Discovered by Pavel A. Cherenkov in 1934
  - 1958 Nobel Prize in Physics





# Cherenkov Radiation

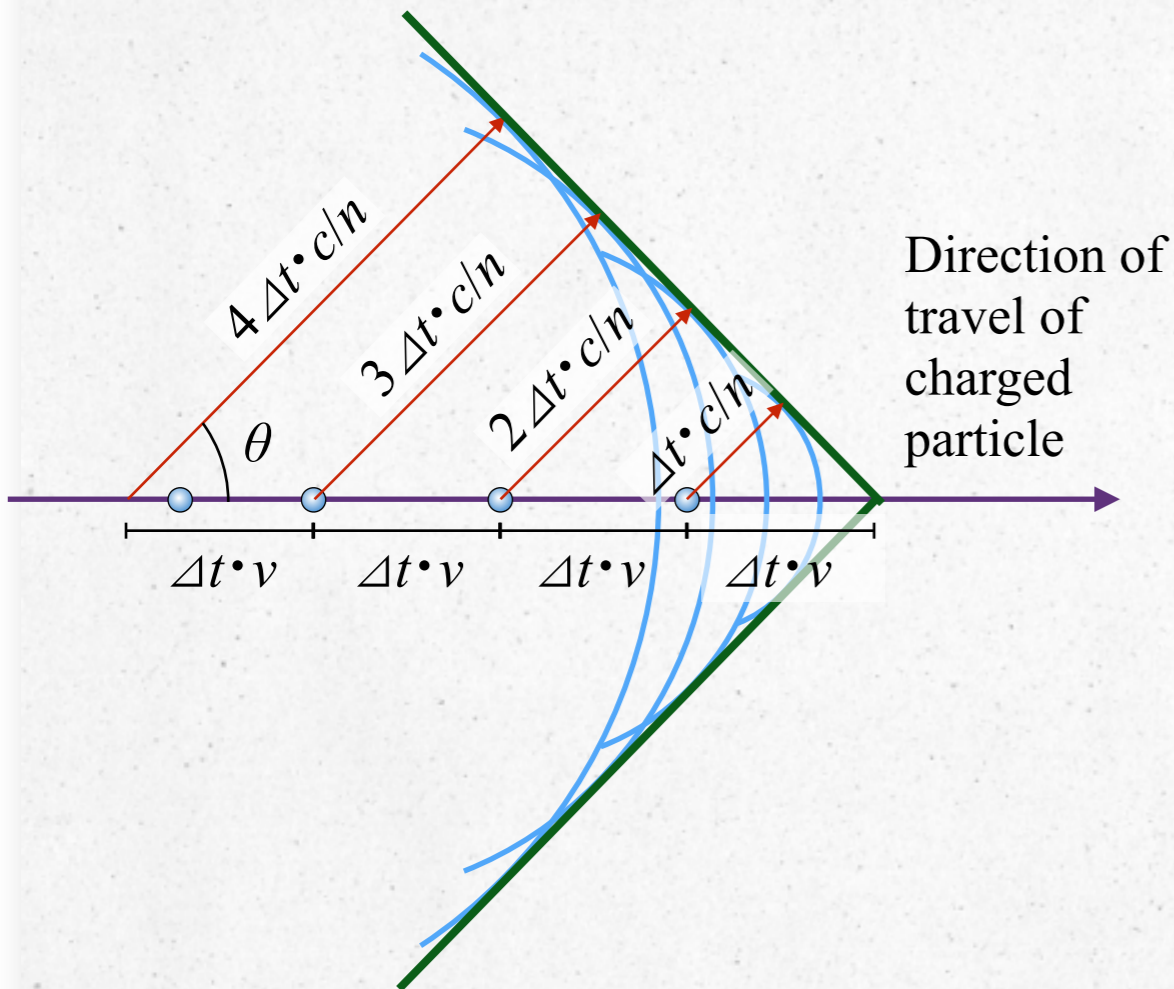
- Polarization by Coulomb force of charged particle



- Symmetric polarization by slow particles
- Even if unpolarized, no emission of electromagnetic wave

- Asymmetric polarization by fast particles
- When the molecules are unpolarized, electromagnetic wave is emitted to the outside.

# Cherenkov Radiation



- Spherical electromagnetic wave
- From each point after unpolarized
- Superimposed wave: Plain
- Huygens' principle
- Emitted at an angle of  $\theta$  with respect to the traveling direction

$$\cos \theta = \frac{\Delta t (c/n)}{\Delta t v} = \frac{1}{nv/c} = \frac{1}{n\beta}$$

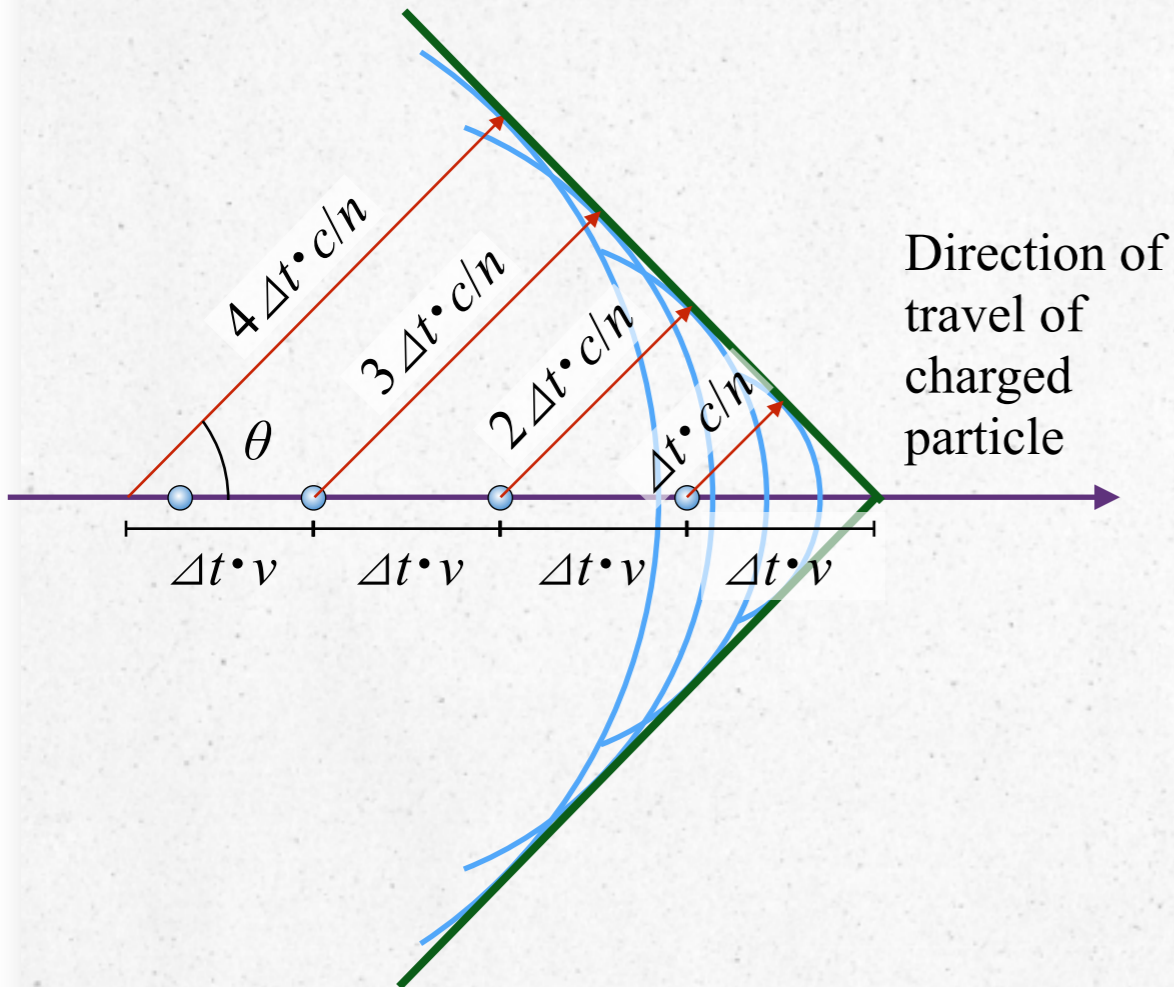
- Condition of Cherenkov radiation

$$n\beta \geq 1 \quad \Rightarrow \quad \beta \geq 1/n$$

- For same momentum
  - $\beta$  is differ with mass
  - Cherenkov light is emitted or not emitted
  - angle  $\theta$  is different from particle mass



# Cherenkov Radiation



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- For same momentum

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Particle identification

- Cherenkov light is emitted or not emitted
- angle  $\theta$  is different from particle mass



# Detectors for Charged Particle Measurement

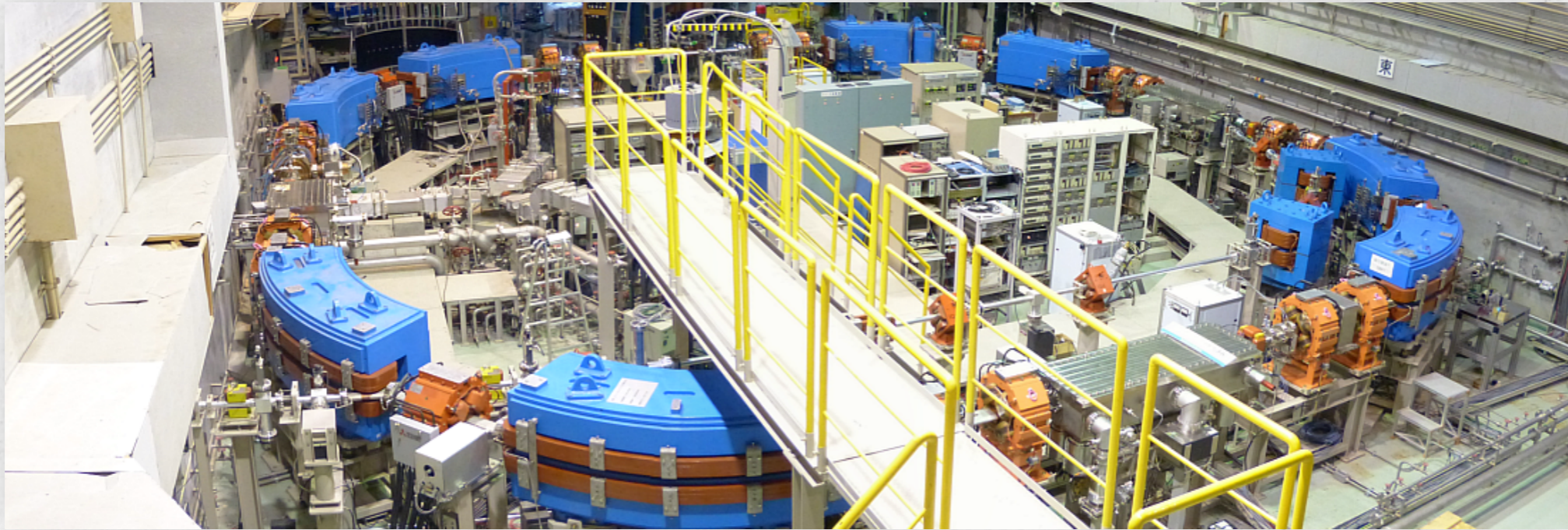


# Signales in Detector

- Charged particle makes a signal by interaction with material
  - Ionization
    - Gas (electron and ion pair)
    - Semiconductor (electron and hole) } Electrical signal
  - Scintillation
    - Electron in material is excited by Coulomb force
    - Visible light / Ultraviolet emission with de-excitation } Photon
  - Cherenkov light
- Detectors to measure and identify
  - Momentum (mass × velocity)
  - Velocity (Time of flight and flight length)
  - Particle species

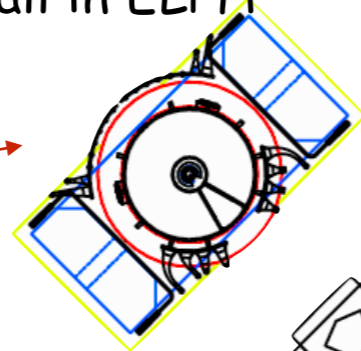


# Example: NKS2 Experiment at ELPH in Tohoku Univ.



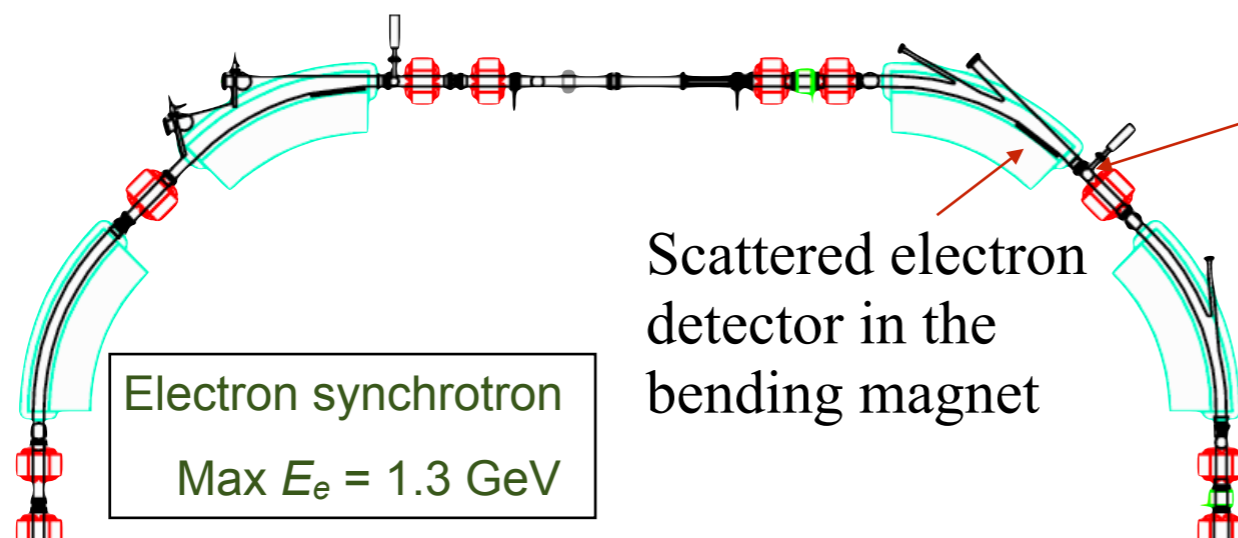
2nd experimental hall in ELPH

Magnetic spectrometer for charged particle measurement



NKS2 experiment:  
Measurement of  
hadron production  
by  $\gamma+p$  /  $\gamma+n$  reaction

Photon beam is generated  
by bremsstrahlung from a  
interaction of electron  
with carbon wire



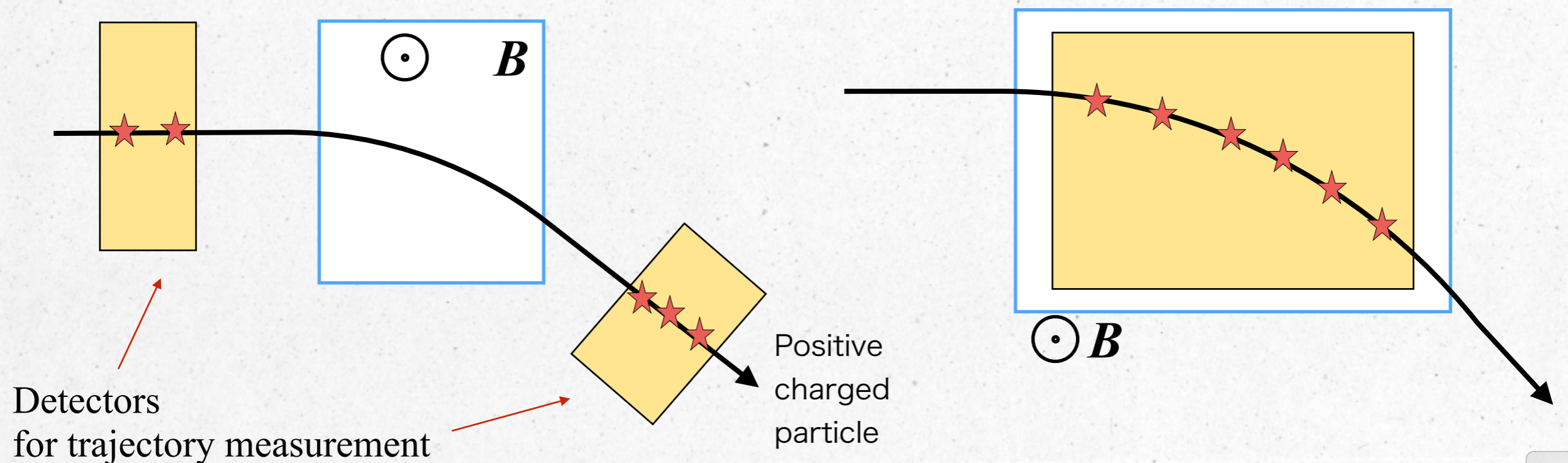
Electron synchrotron  
Max  $E_e = 1.3$  GeV

Scattered electron  
detector in the  
bending magnet

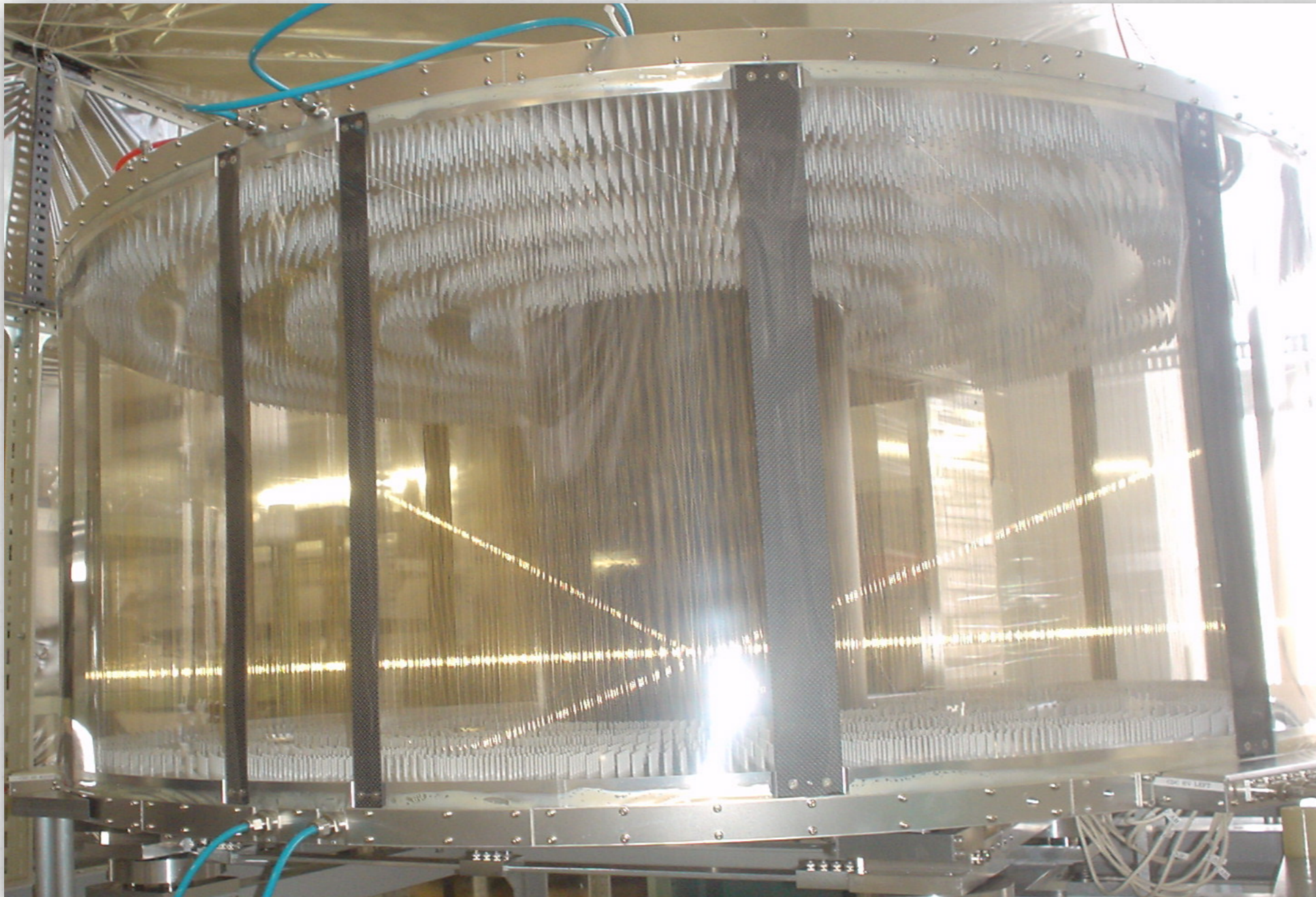


# Momentum

- Momentum can be estimated from trajectory in magnetic field
  - Lorentz force:  $F = q\mathbf{v} \times \mathbf{B}$
  - Equation of motion of circle:  $F = ma = mv^2/r$
  - $\rho = mv_t/qB = p_t/qB$ 
    - $\rho$ : radius of bending curve
    - $v_t, p_t$ : transverse component of velocity and momentum
- Trajectory reconstructed from hits in detector(s)
  - Si semiconductor detector, drift chamber, and etc.
  - Typical position resolution: several 10  $\mu\text{m}$  to a few 100  $\mu\text{m}$



# Drift Chamber of NKS2

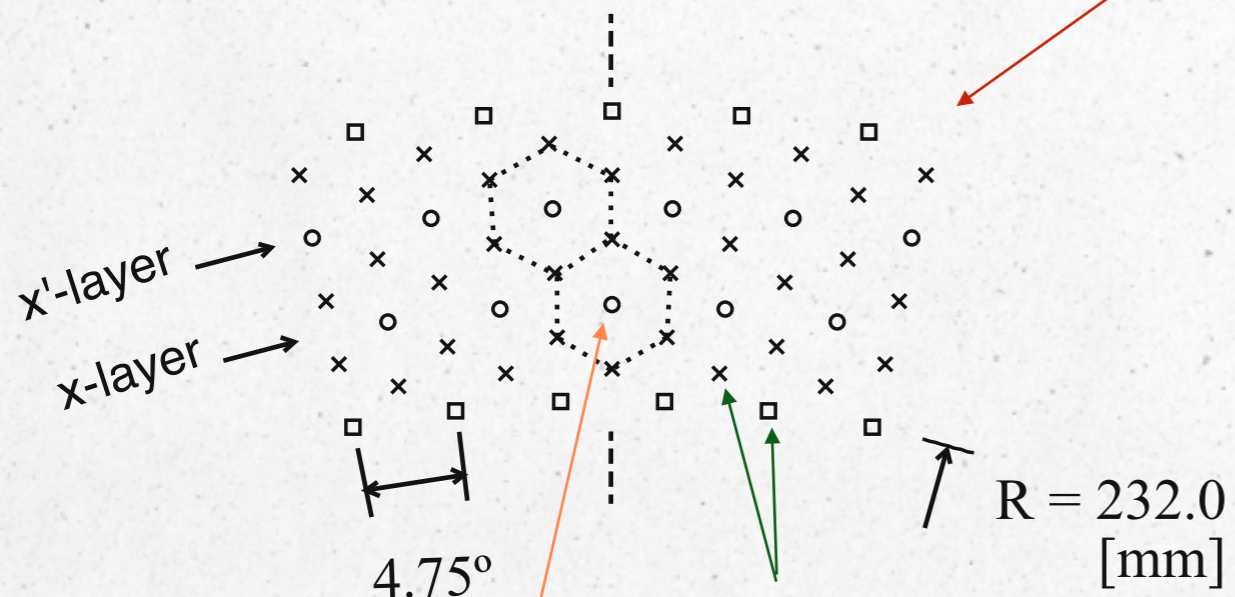
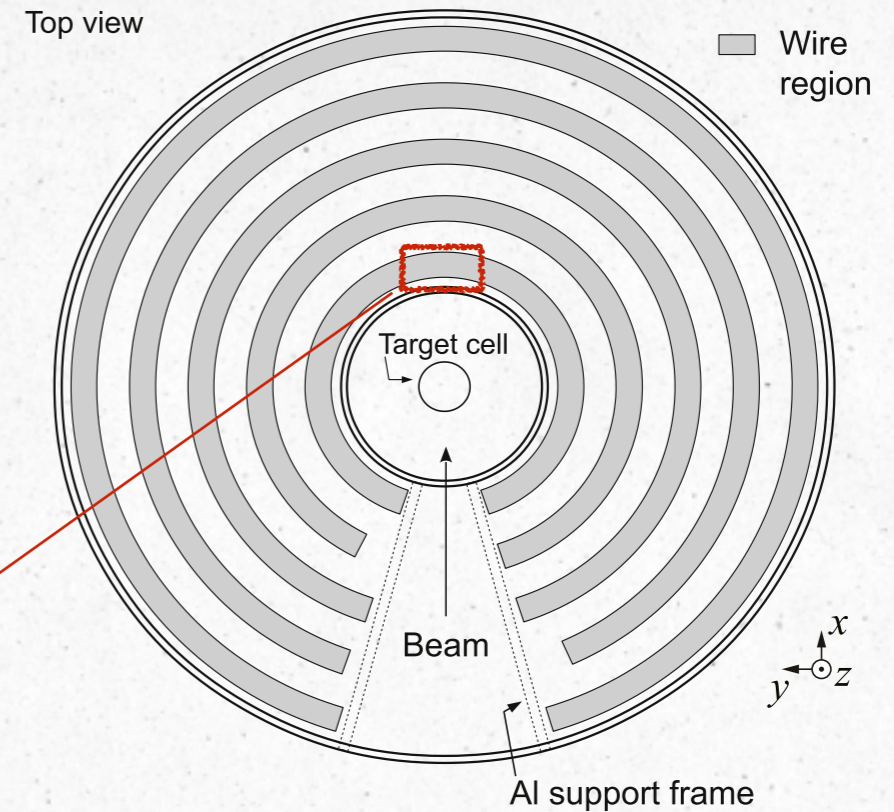
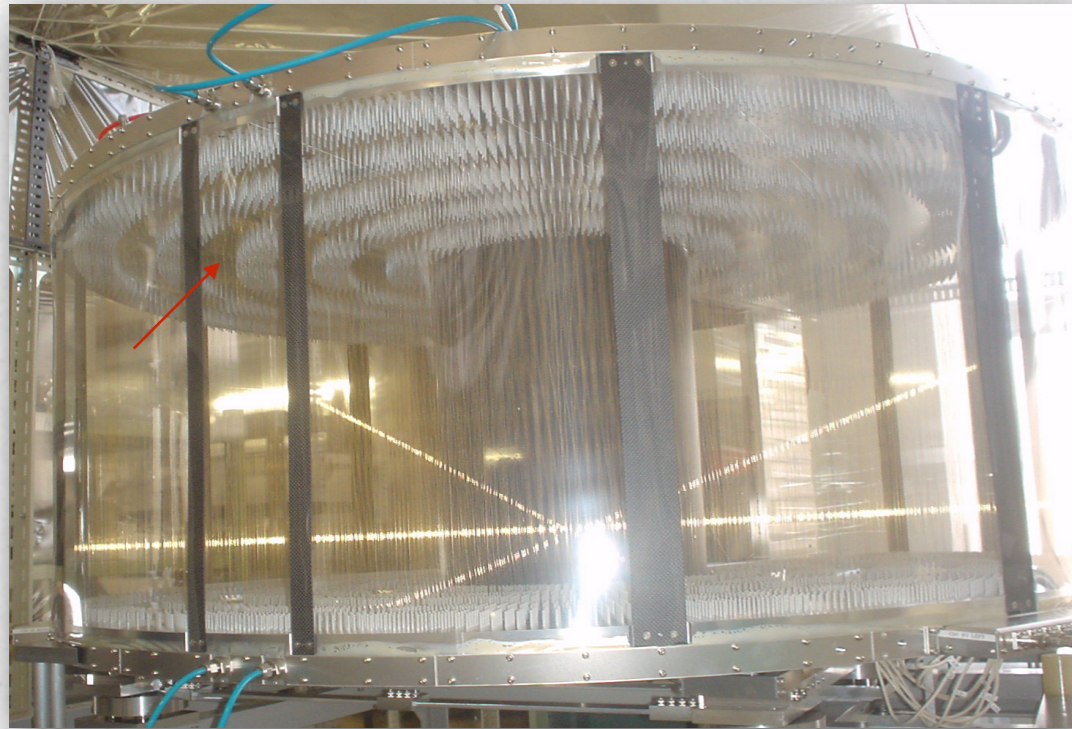


Many wires in gas volume

Wire: make electrical field and detection of signal

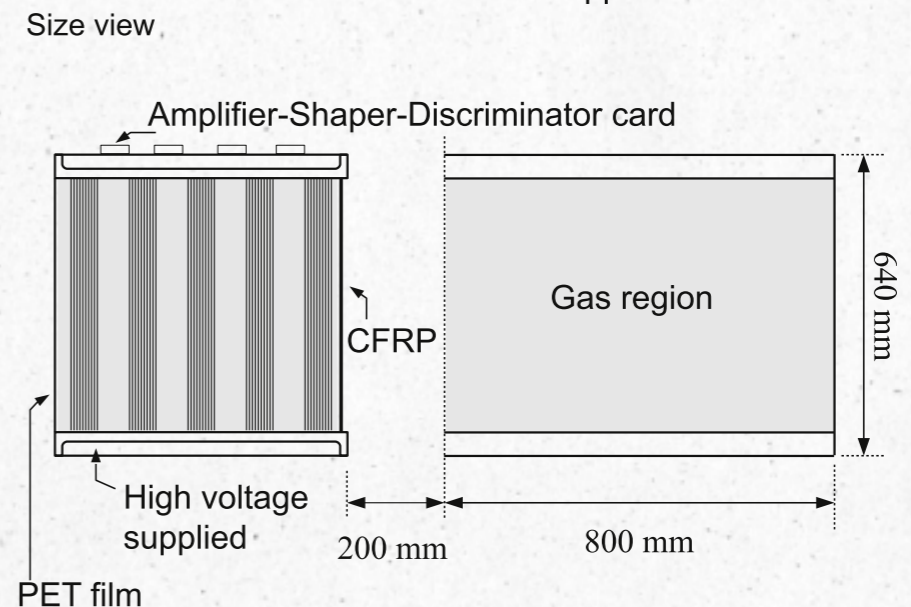


# Drift Chamber of NKS2



Wire for signal measurement

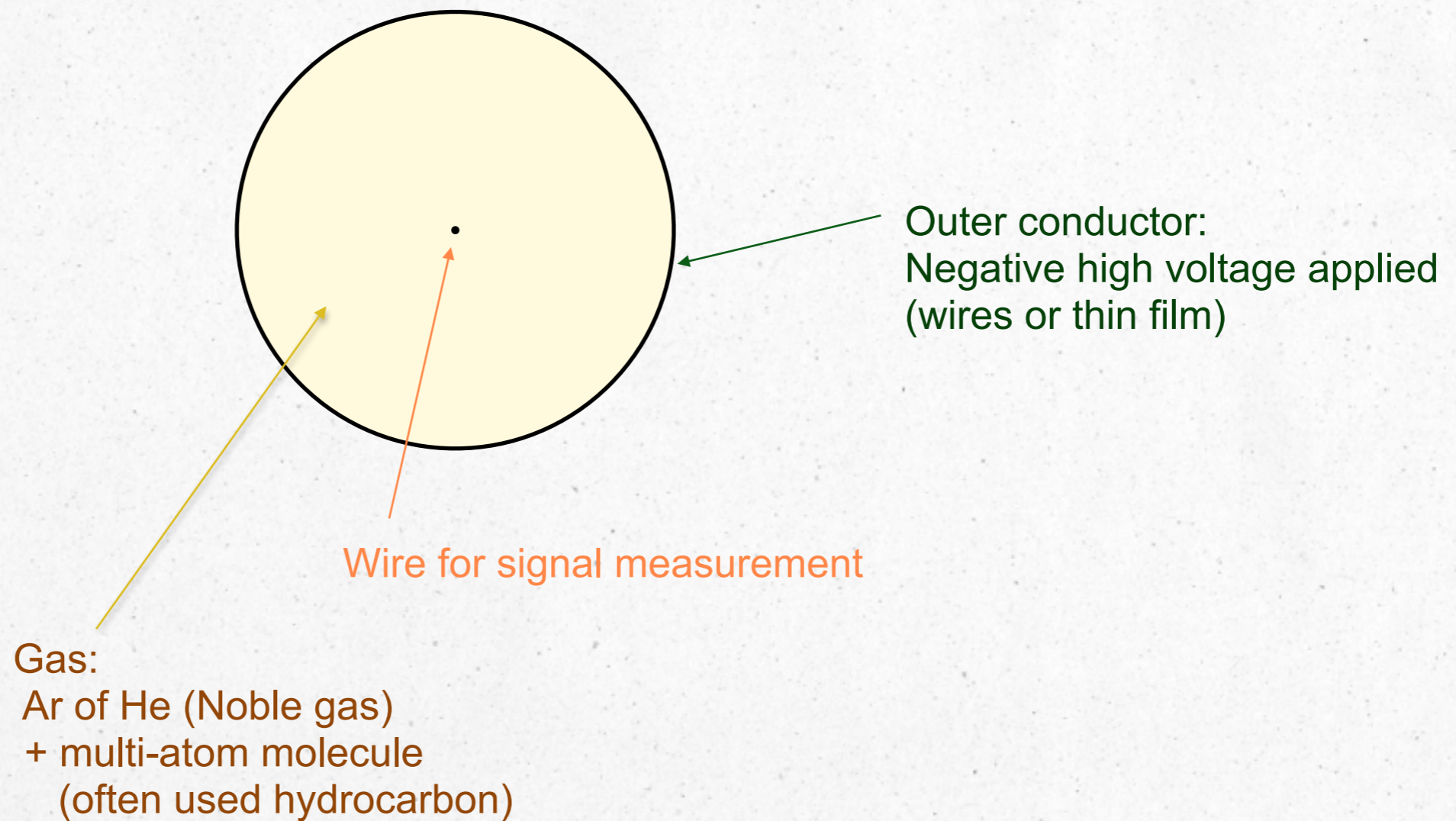
Wire with high voltage applied



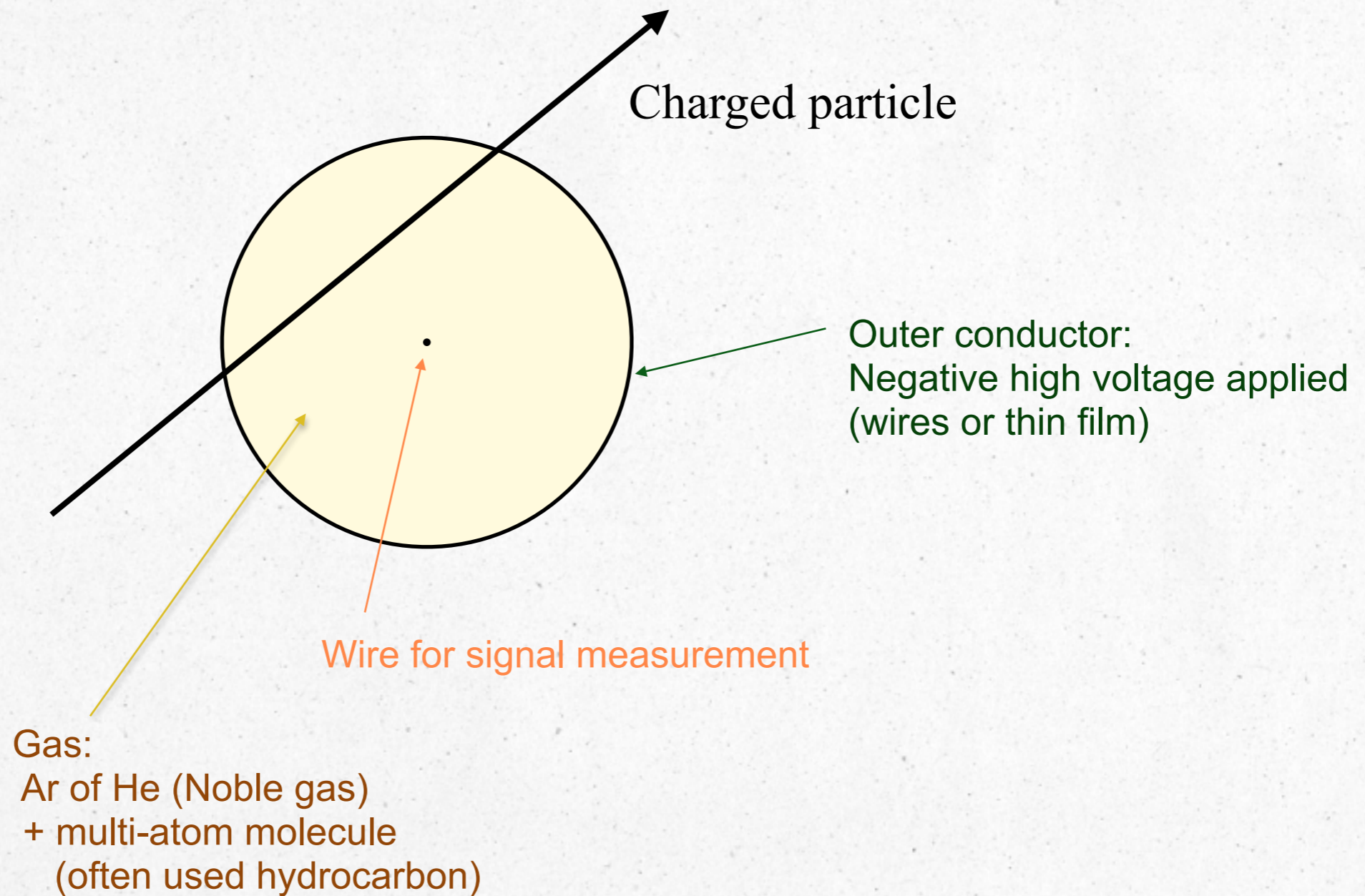
Figures: Nucl. Instrum. Methods A886 (2018) 88-103



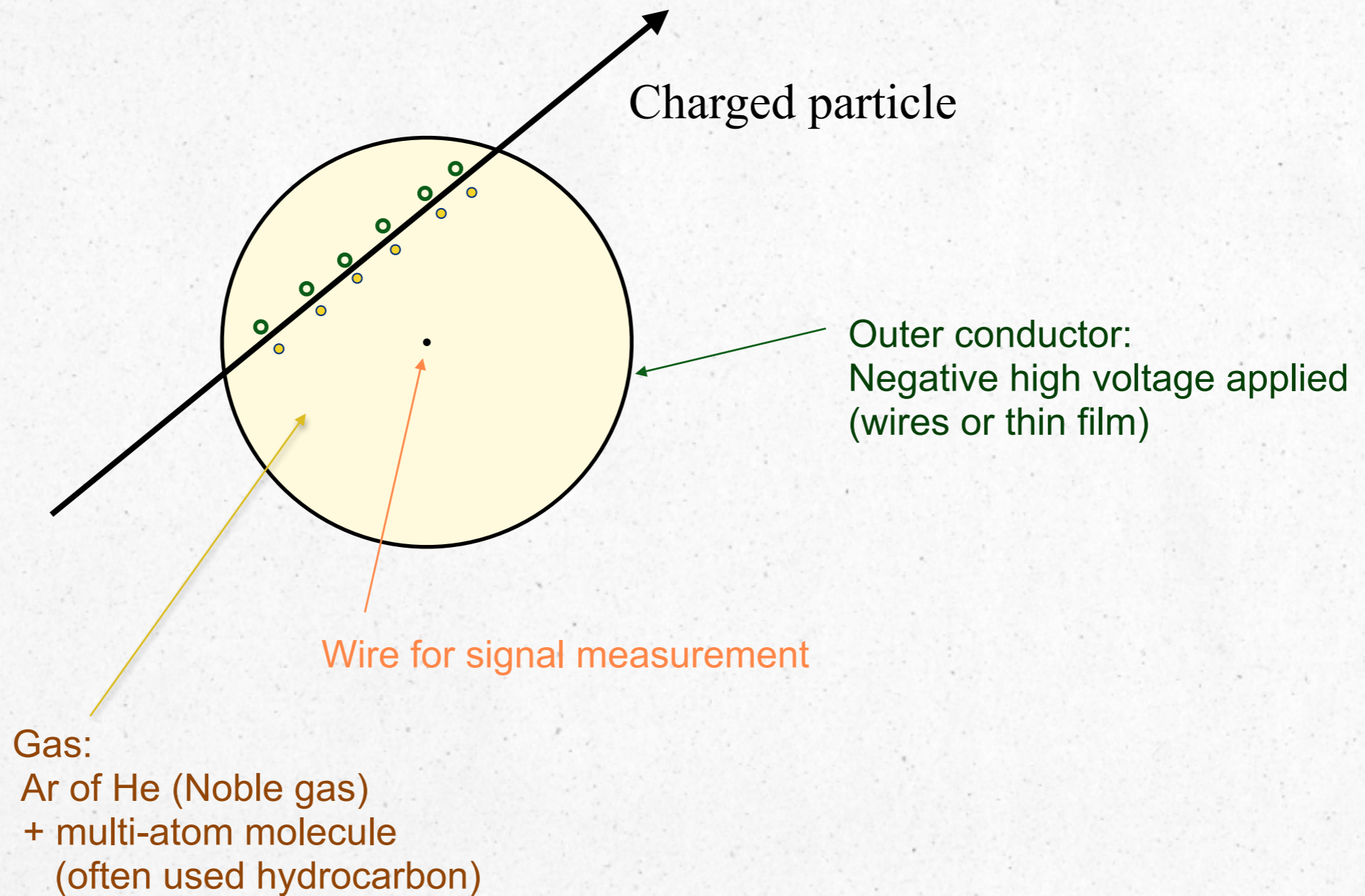
# Principle of Drift Chamber



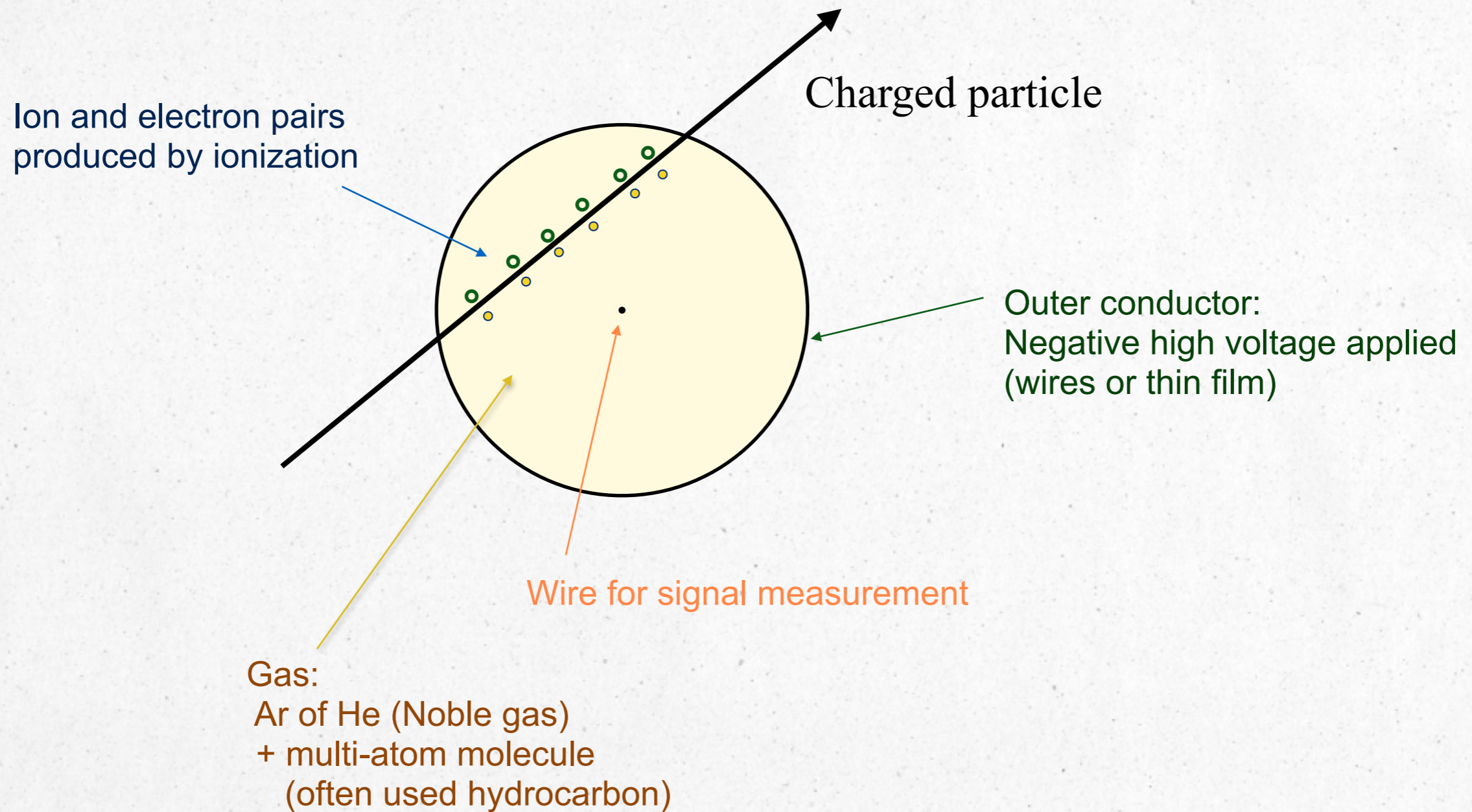
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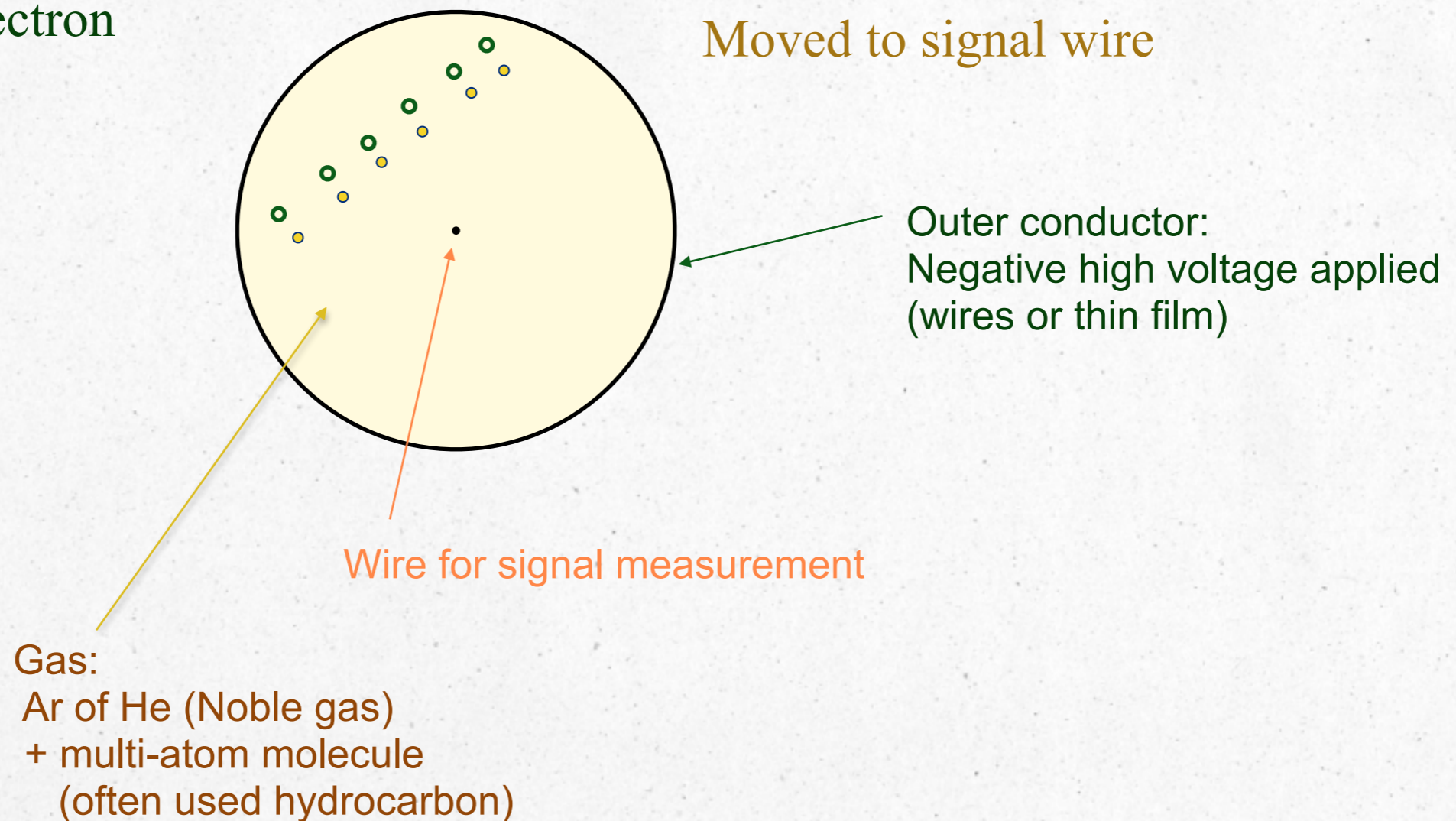
# Principle of Drift Chamber

## Ions

Moved to outer conductor  
Slower than electron

## Electrons

Moved to signal wire



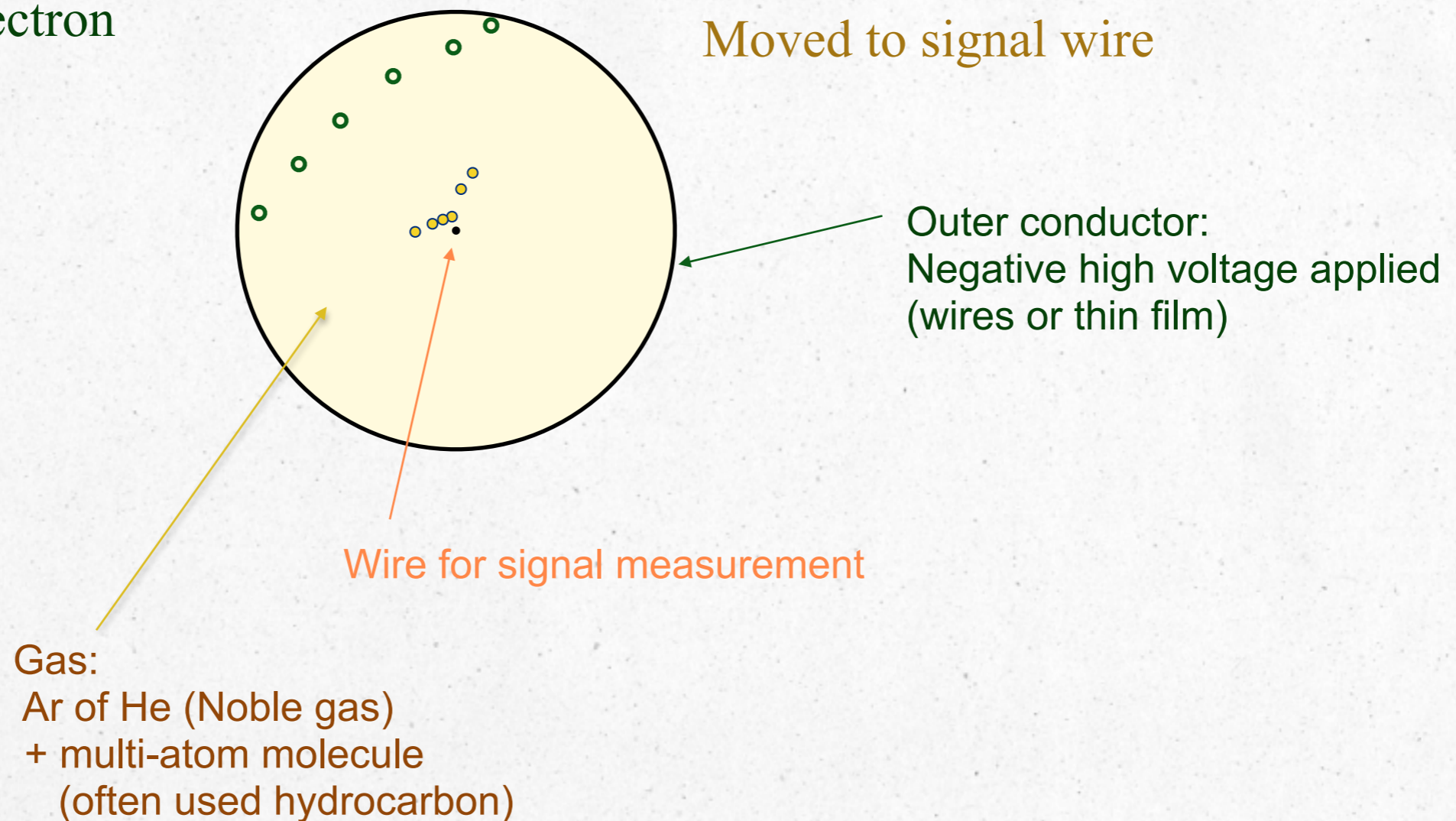
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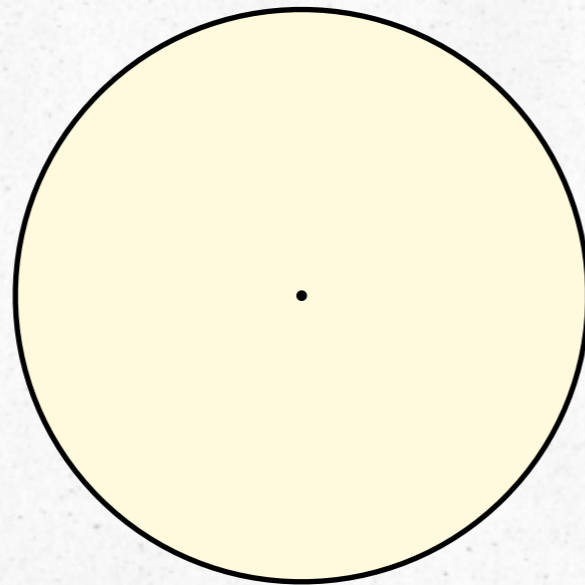
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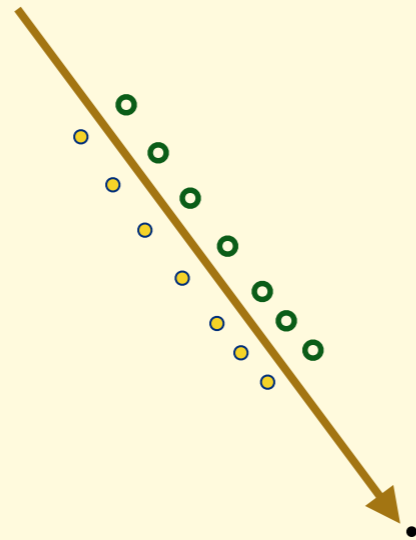




# Principle of Drift Chamber

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# Principle of Drift Chamber



Electron

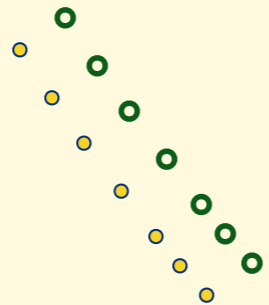
accelerated at close to wire  
due to strong electric field and  
makes electron-ion pairs

Electrons produced makes the other pairs

Repeat the processed

→ Avalanche amplification:  $\sim 10^5 - 10^6$

# Principle of Drift Chamber



Electron

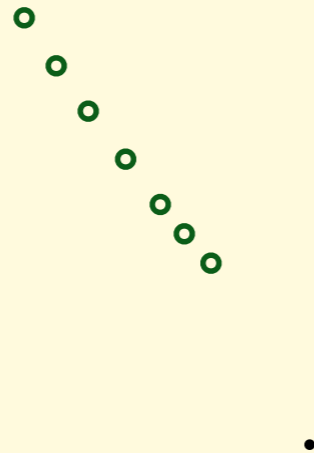
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Electrons produced makes the other pairs

Repeat the processed

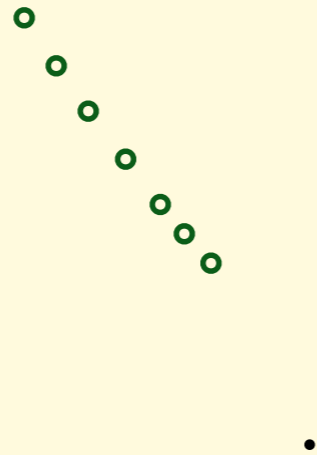
→ Avalanche amplification:  $\sim 10^5 - 10^6$

# Principle of Drift Chamber

## Ions

Still remained around wire  
after electrons were  
absorbed into the wire

→ Ions induce charge on wire



## Electron

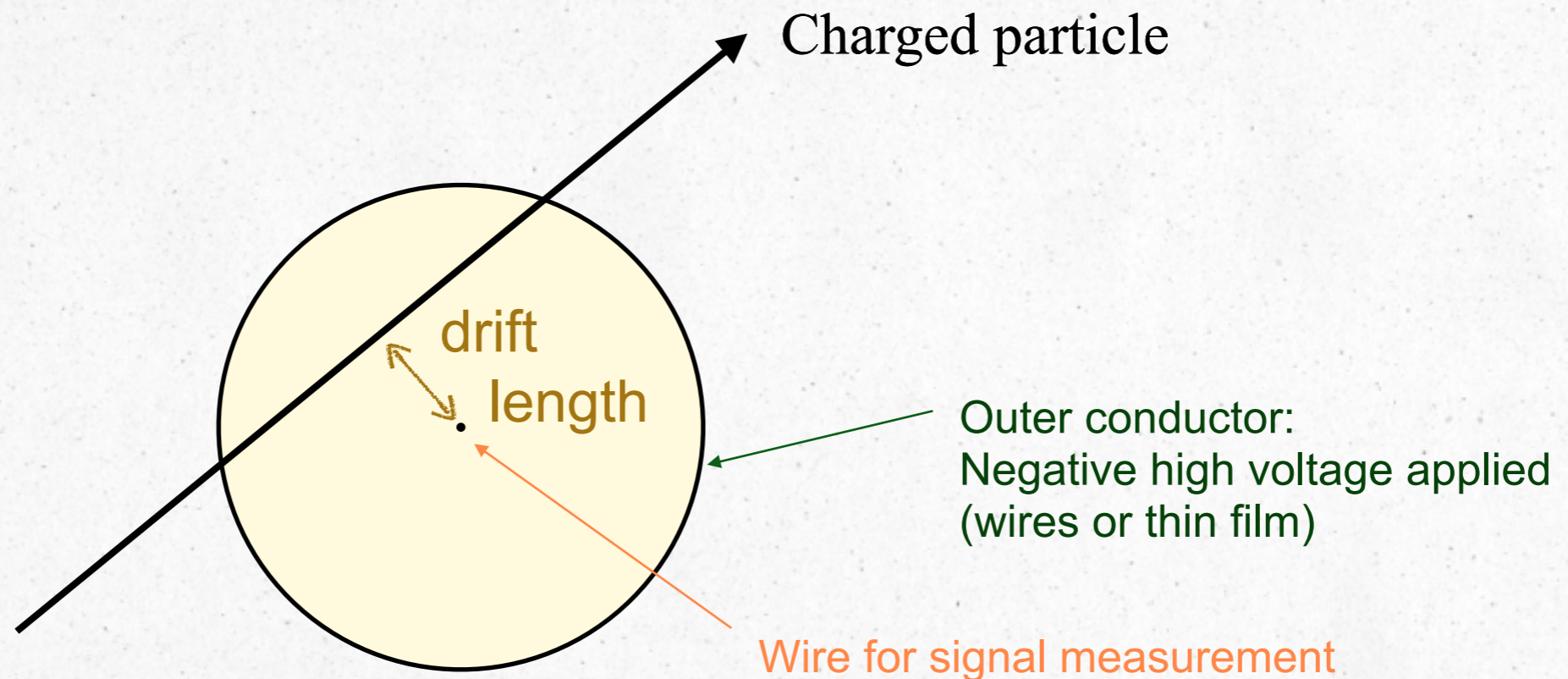
accelerated at close to wire  
due to strong electric field and  
makes electron-ion pairs

Electrons produced makes the other pairs

Repeat the processed

→ Avalanche amplification:  $\sim 10^5 - 10^6$

# Principle of Drift Chamber



Speed of electron in gas:  $\sim 5 \text{ cm}/\mu\text{s}$

Slower than charged particle we want to measure the trajectory

( $\sim$  close to  $c$ :  $30 \text{ cm/ns} = 30000 \text{ cm}/\mu\text{s}$ )

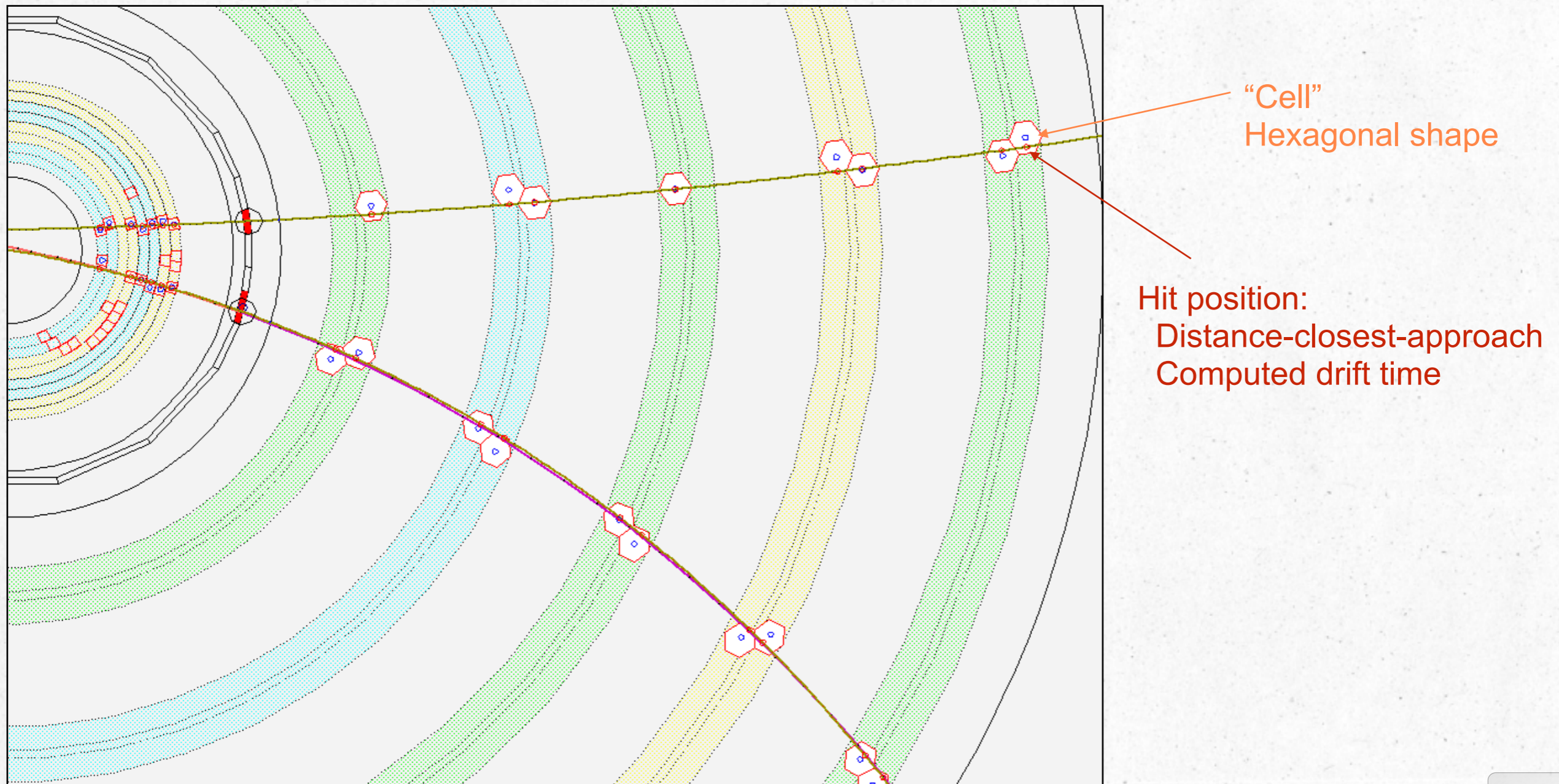
Drift time of electron

Time delay of signal respect to charged particle passage

Distance from wire to charged particle trajectory

# Example of Tracking in NKS2

- Reconstruction of trajectory from hit information
  - Resolving equation of motion in magnetic field



# Time-of-Flight: TOF

- Timing measurement of passage of charged particle
  - For velocity calculation
  - Typical timing resolution: several 10 ps to several 100 ps
- Two set of TOF detector
  - Absolute time measurement is difficult
    - Recording time is vary with condition
    - Relative time among detector is not changed





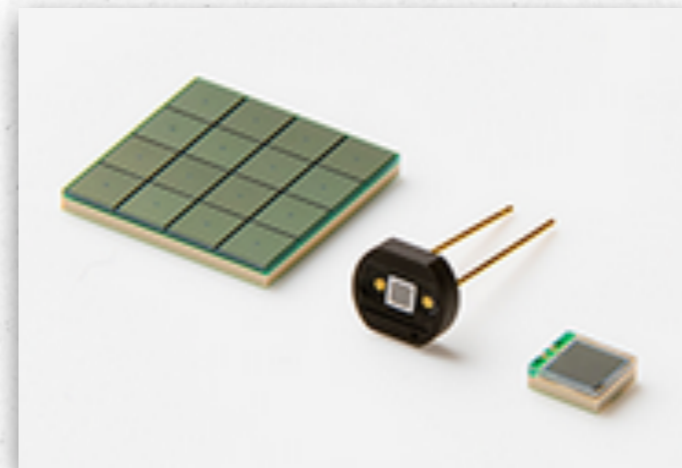
# TOF Detectors

- Plastic scintillator + photon sensor
  - Photon sensor
    - Photo-Multiplier Tube (PMT)



Plastic scintillator is surrounded by a black sheet to avoid large current flow in PMT due to external light

- Silicon Photo-Multiplier (SiPM)
  - Avalanche photo-diode operating Geiger mode
  - Work in magnetic field
  - Small size

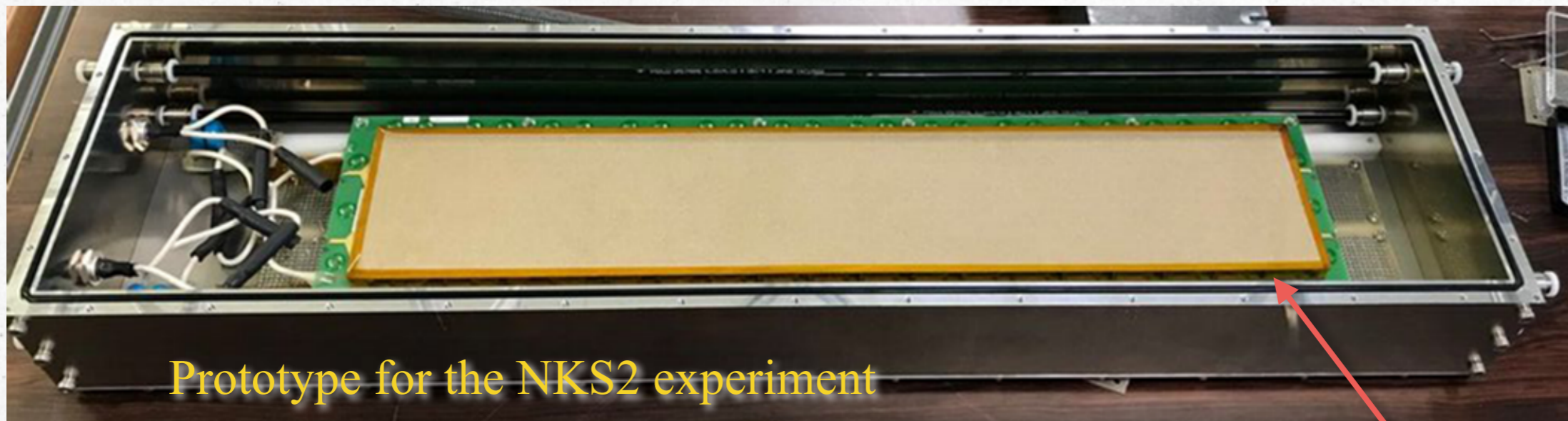


MPPC by HAMAMATSU

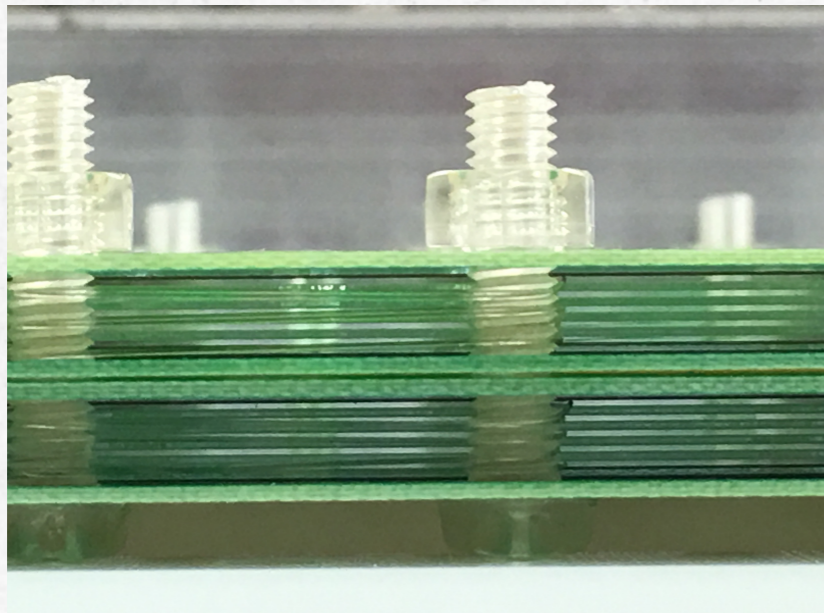


# TOF Detectors

- Multi-gap Resistive Plate Chamber
- Gas multiplier with Geiger mode



Prototype for the NKS2 experiment



Gas:  
R-134a+SF<sub>6</sub> (90:10)

0.4<sup>th</sup> mm soda-lime glass  
#2 fish line (∅ 0.23mm)  
5-gap and double-stack

47 cm × 3.5 cm strip  
Readout from both ends

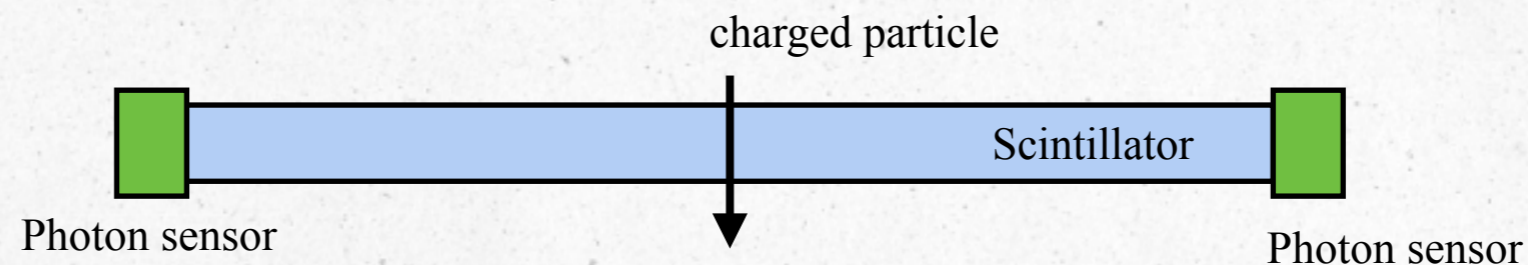
Intrinsic resolution  
~120 ps



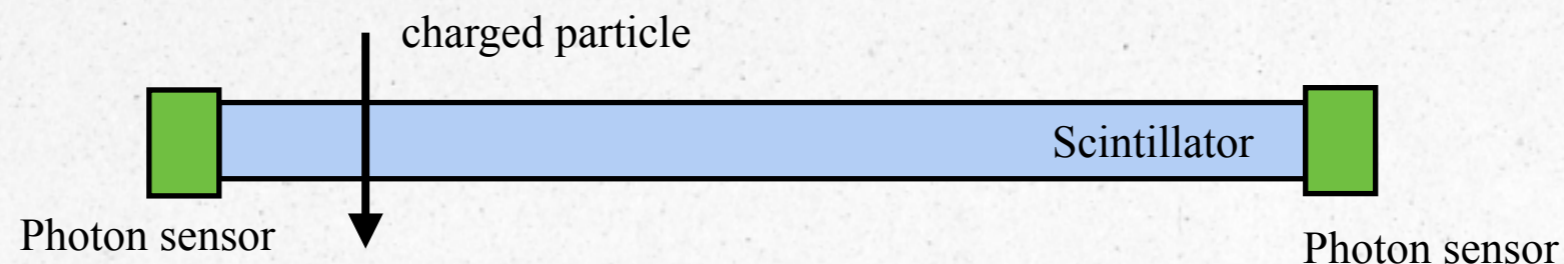
# TOF Detector as Hodoscope

- Hodos: Way or path in Greek
  - Hodoscope: Position detector
  - Array of plastic scintillation counters
- Two side read-out of signal
  - Time difference of hit signal at both edge
  - → distance from hit position to both edge

Example 1:  
Charged particle pass through  
center of scintillator



Example 2:  
Charged particle pass through  
close to edge of scintillator



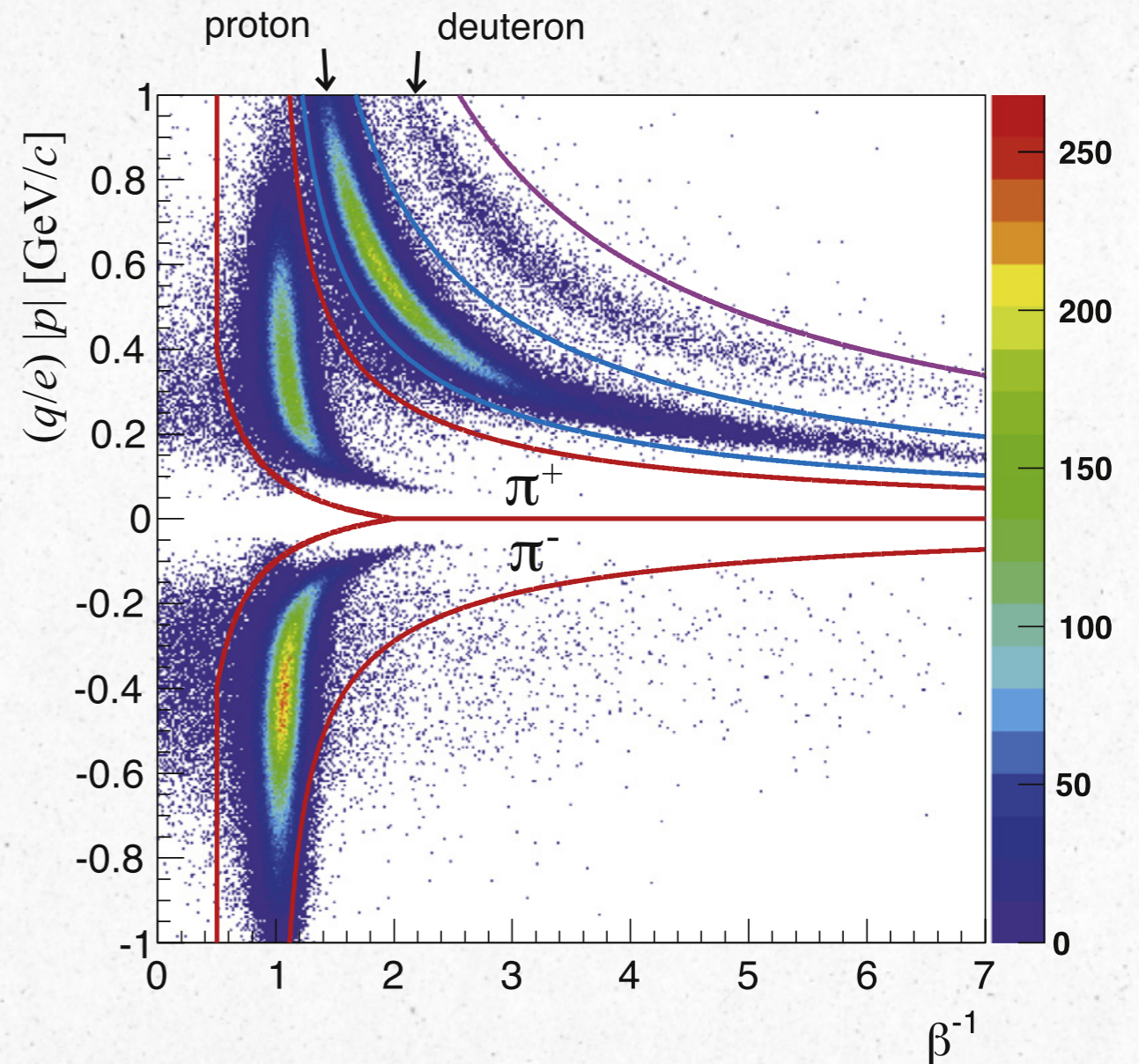
# Particle Identification

- Characteristics of particle
  - Charge sign
    - Bending direction in magnetic field
  - Mass
    - Momentum and velocity (Flight-path / TOF)

$$E^2 = (mc^2)^2 + (pc)^2$$

$$p^2 = \frac{v^2 E^2}{c^4} = \beta^2 \frac{E^2}{c^2}, \text{ where } \beta = \frac{v}{c}$$

$$(mc^2)^2 = p^2 c^2 \left( \frac{1}{\beta^2} - 1 \right)$$



Figures: Nucl. Instrum. Methods A886 (2018) 88-103



# Summary

- Topics
  - Radiation and nuclear physics
  - Charged particle measurement
    - Principle
    - Detectors
      - Example from NKS2 experiment at ELPH, Tohoku Univ.
- The technique in elementary particle/nuclear physics
  - Many applications in medical
    - Using radiation, radioactive material, and accelerator

