

**Cryogenics based chilling, energy supply
and services for deeper or hotter mines.**

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CanMIND GLENCORE



Interrelations

Glencore

- Funding
- Engineering Critique
- Real Mine Design

CanMIND Associates

- Principle Investigator
- Physics Engineering
- Concept Development

Highview Power

- Liquid Air Energy Storage
- Pilot plant
- 5 MW plant

CEMI

- Project Management
- Business Acumen

Dearman Engine Co.

- Engine Development
- Techno-economic Analysis
- In Kind Contribution

UDMN

- Funding
- Industry Network
- Commercialisation

Project Elements

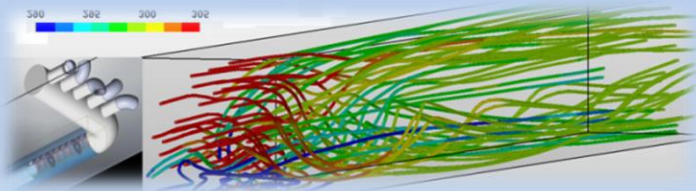
**Large UG Equipment
Powered by Dearman Engines**



**UG Cryogenic
Piping and Storage**

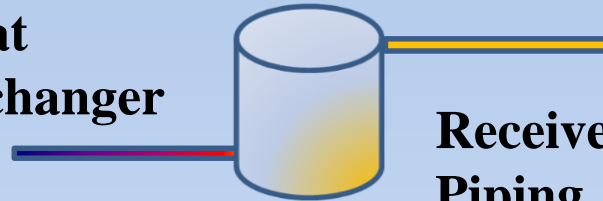


**Cryogenic Chilling
CFD Modelling**



Compressed Air System Design

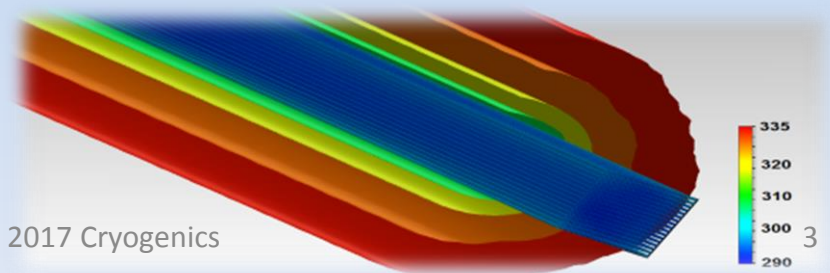
**Heat
Exchanger**



**Receiver size
Piping
Modularity design**

Rapid Response Chilling on Demand

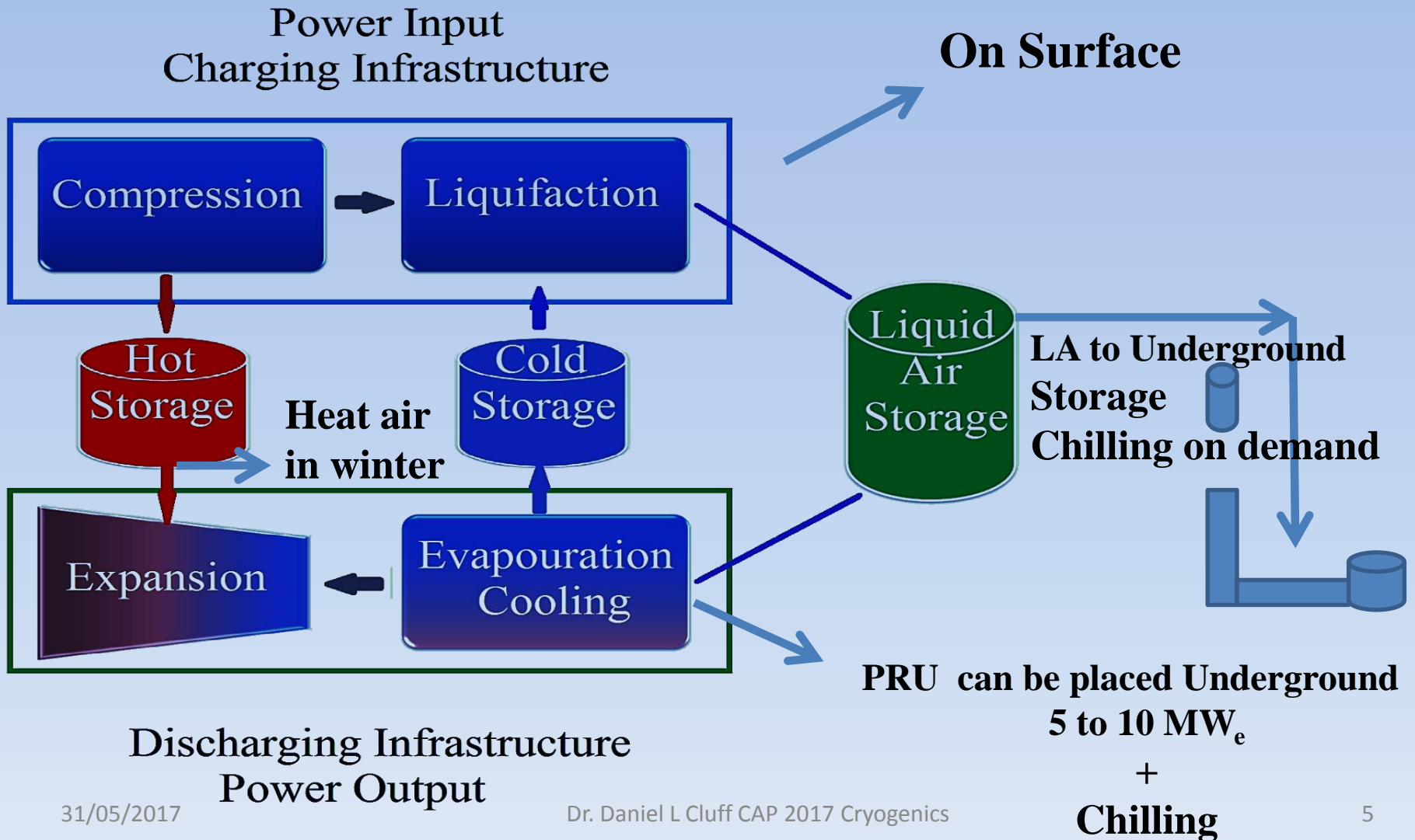
PRU Design



Why LAES

- LAES technology provides plant size economy of scale
- Energy storage is an emerging technological niche
- The cryogenic liquids are produced on the surface in a standard cryogenic liquefaction plant.
- Cryogenic liquid is piped to the depth required
 - Depends on mine design decision
 - Sent to a central location and chill air in downcast shaft
 - Sent to individual levels to chill on demand

LAES Simplified Schematic



350 kW 2.5 MWH Pilot Plant

Highview Power Slough UK



Six 3600 tpd O₂ plants in China

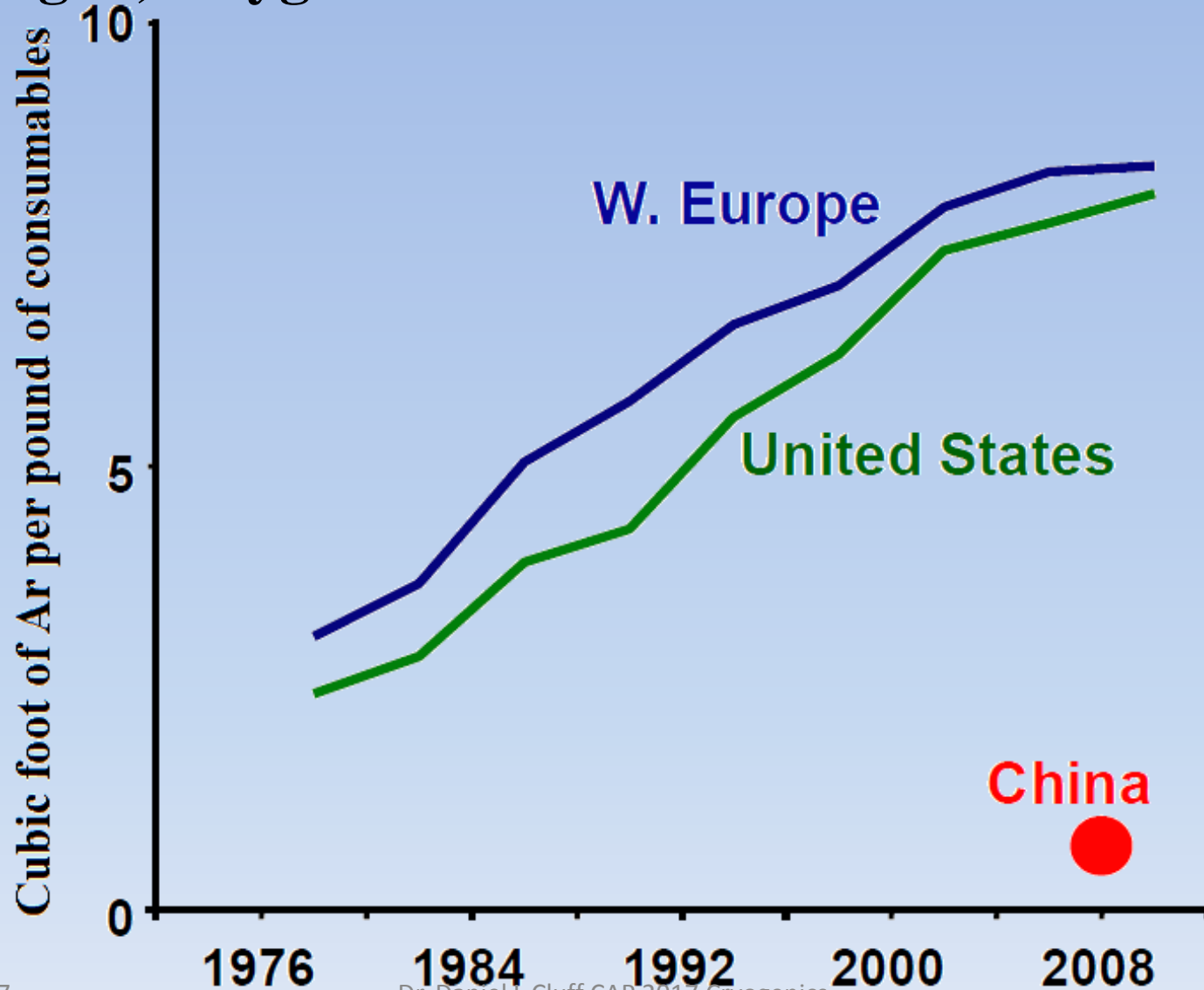


LAES Process

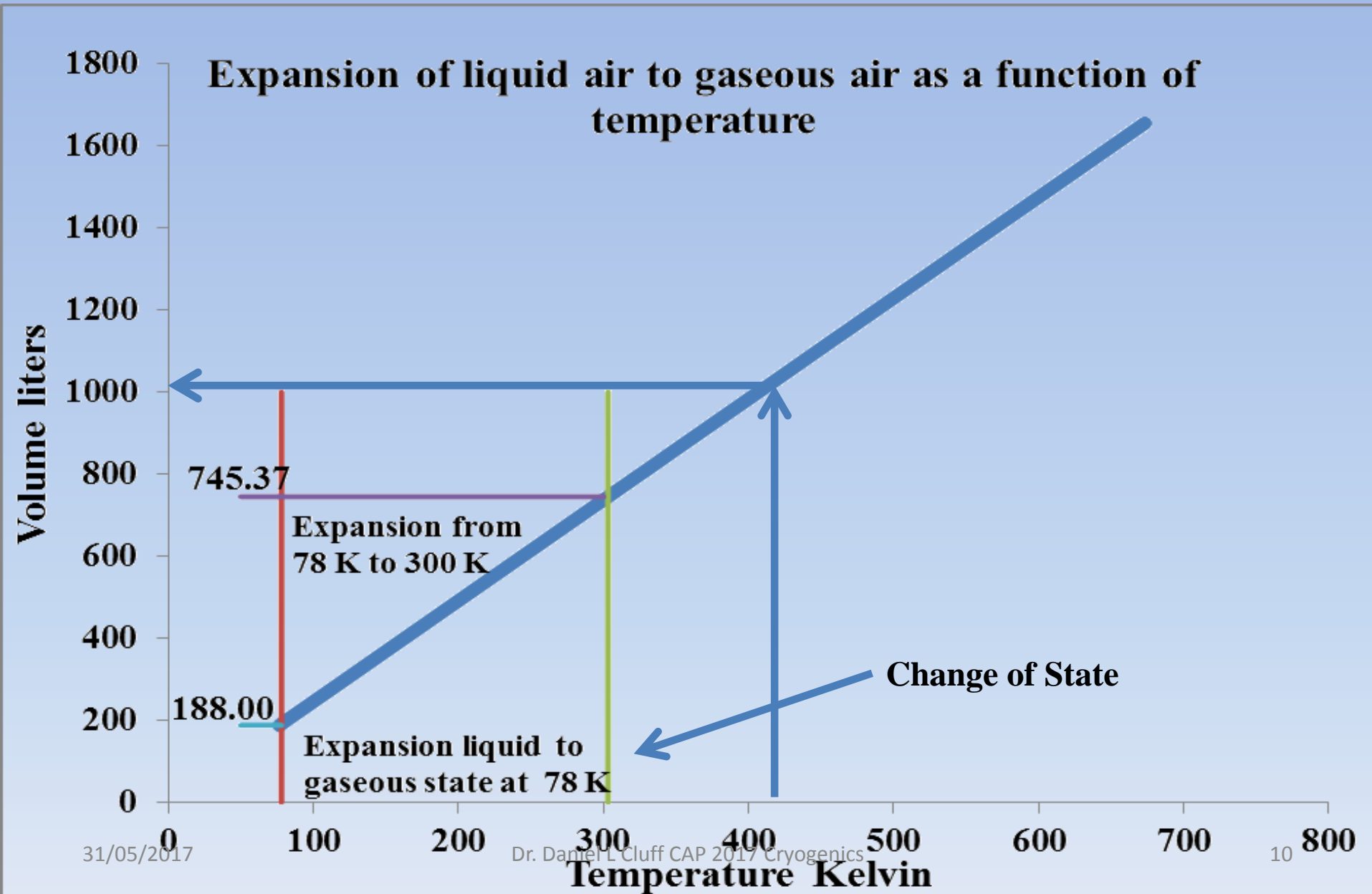
- A Liquid Air Energy Storage (LAES) system is comprised of a **charging system**, an **energy storage section** and a **discharging system**.
- Standard industrial air liquefaction plant
 - the electrical grid or a renewable energy project supply the electrical energy.
- Air drawn from the ambient environment.
 - The process creates liquid air a cryogenic liquid at temperatures near -196°C (78 K).
- The liquid air is stored in a low pressure insulated tank.
 - Easily accessed energy storage repository
 - Low risk to the environment
- When power is required
 - liquid air is pumped to a high pressure and evaporated through a turbine system.
- Capable of providing the pressure necessary
 - to power a piston engine or turbine resulting in useful work
 - to generate electricity or drive a cryogenically powered vehicle.

Ancillary Economics Includes

Argon, Oxygen Markets Both Continue to Grow



700+ ℓ Gaseous Air Per 1 ℓ Liquid Air



A Basic Calculation to Illustrate the Heat Absorbed on Change of State

The heat absorbed per kg of liquid air:

Ambient $T_a = 29.85^\circ\text{C}$ ($273.15 + 29.85 = 303\text{K}$)

Cryogenic $T_c = 78\text{ K}$

Latent Heat of vaporisation $L_v = 205\text{ kJ/kg}$

Step 1: The mass “ $m\text{ kg}$ ” absorbs ΔQ_L ,
Becomes a gas at or near T_c

$\Delta Q_L = mL_v = (m\text{ kg})(205\text{ kJ/kg}) = (m\text{ kg})205\text{ kJ/kg}$

Change of state is approximately **180ℓ** (gaseous) / **1ℓ** (liquid)

A Basic Calculation to Illustrate the Heat Absorbed on Expansion

Step 2: Very cold air (≈ 80 K) warms up to T_a

Expansion with heat absorbed ΔQ_a .

$$\Delta T_g = T_a - T_c = 303 - 78 = 225 \text{ K}$$

change in gas temperature

$$\Delta Q_a = mC_p\Delta T_g = (m \text{ kg})(1.005 \text{ kJ/kg-K})(225 \text{ K}) = 226.13 \text{ kJ/kg}$$

Heat absorbed due to change in gas temperature

$$\Delta Q_T = \Delta Q_L + \Delta Q_a = mL_v + mC_p\Delta T_g = (m \text{ kg})(451.13) \text{ kJ/kg}$$

Total heat absorbed due to change of state and expansion

For 1 MW_r Chilling

So let $\Delta Q_T = 1 \text{ MJ}$

The total heat absorbed by the ambient air

The mass of liquid air required is **2.217 kg**

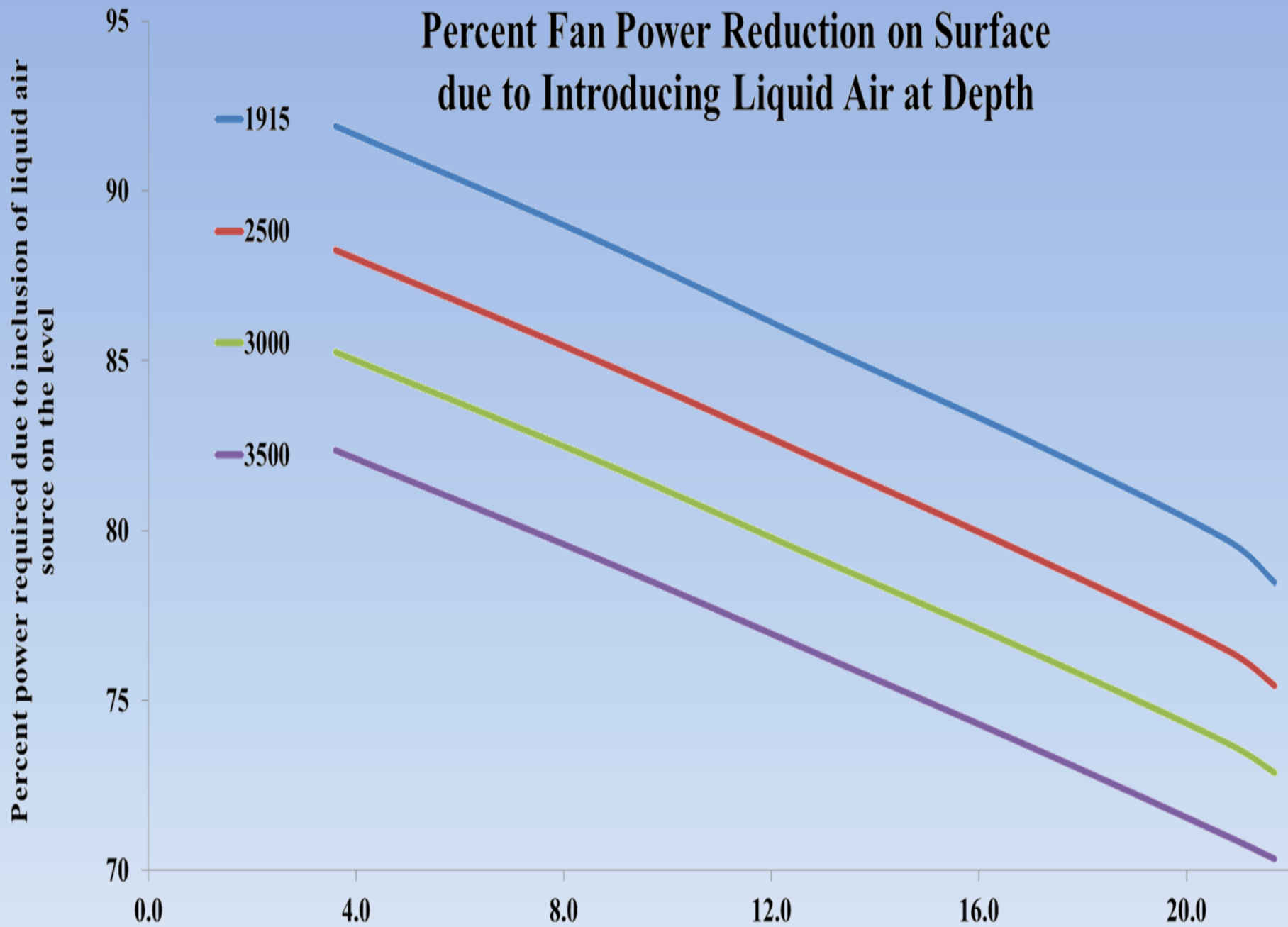
So a liquid flow **2.217 kg/s** will provide **1 MW_r** chilling.

The density of liquid air is about **870 kg/m³**

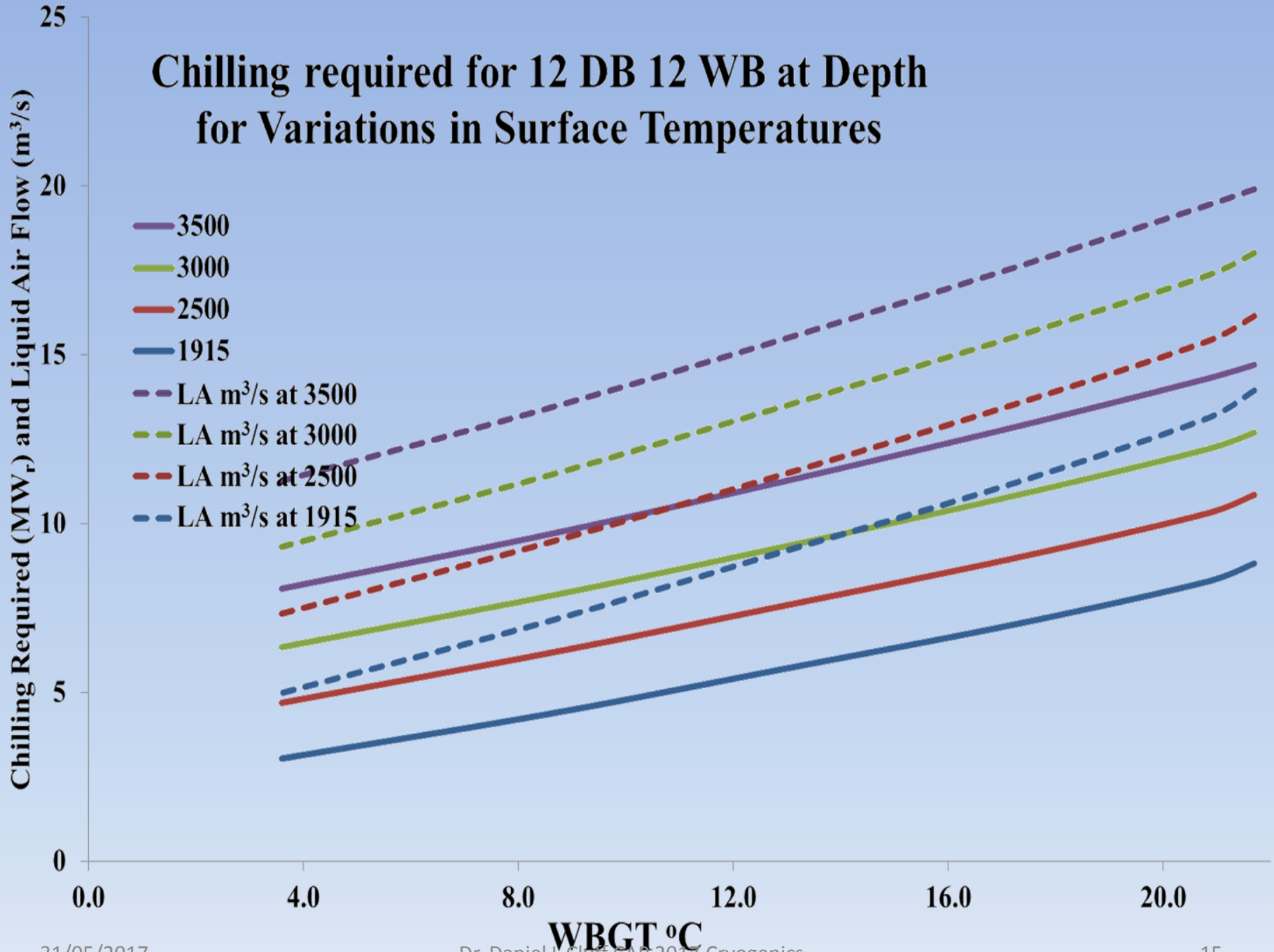
So a flow of about **2.55 ℓ_(liquid)** provides **1MW_r**

Final gaseous volume **1899.42 ℓ_(gaseous)** or **1.9 m³**

Percent Fan Power Reduction on Surface due to Introducing Liquid Air at Depth



Chilling required for 12 DB 12 WB at Depth for Variations in Surface Temperatures

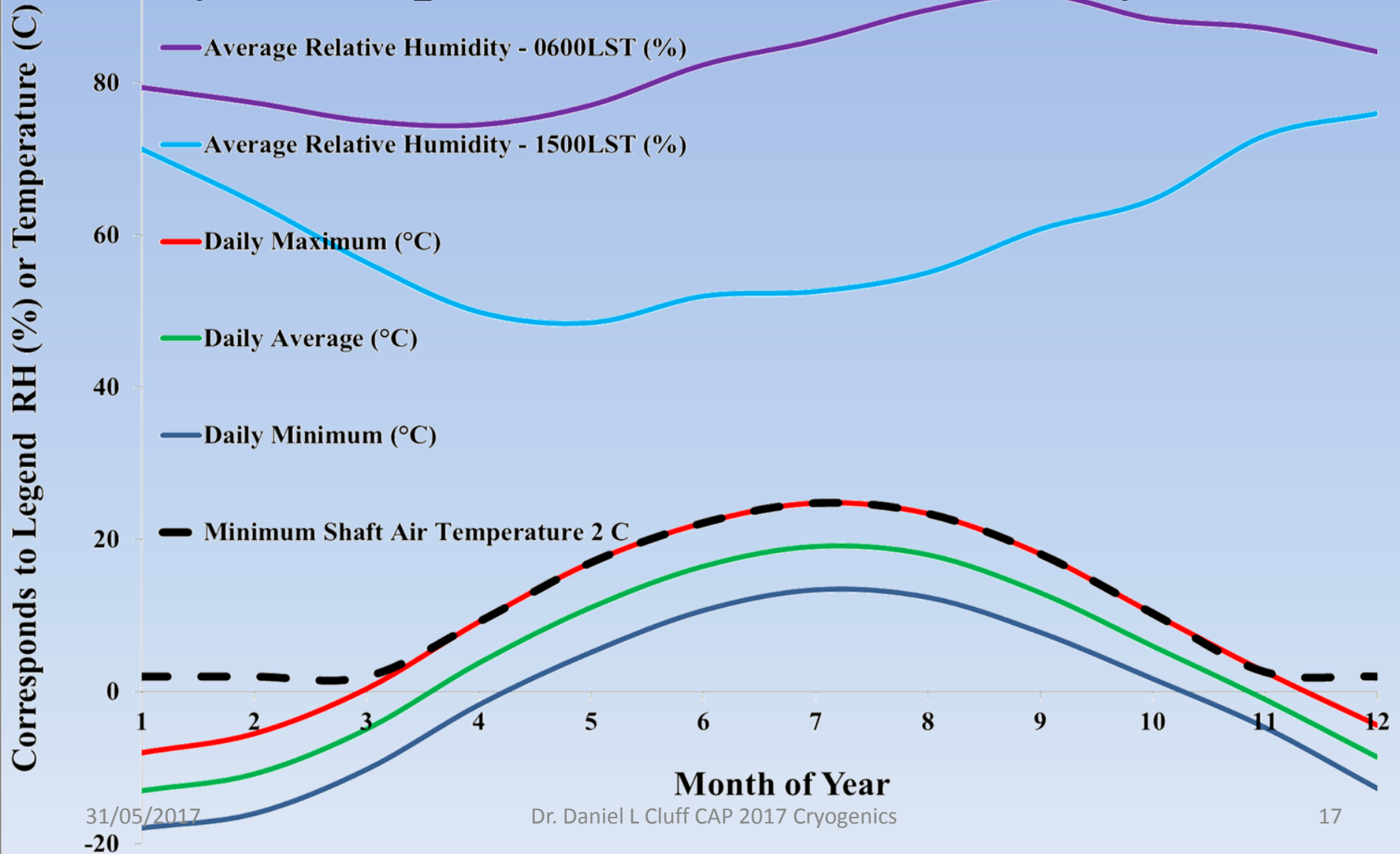


Psychrometrics to Liquid Air

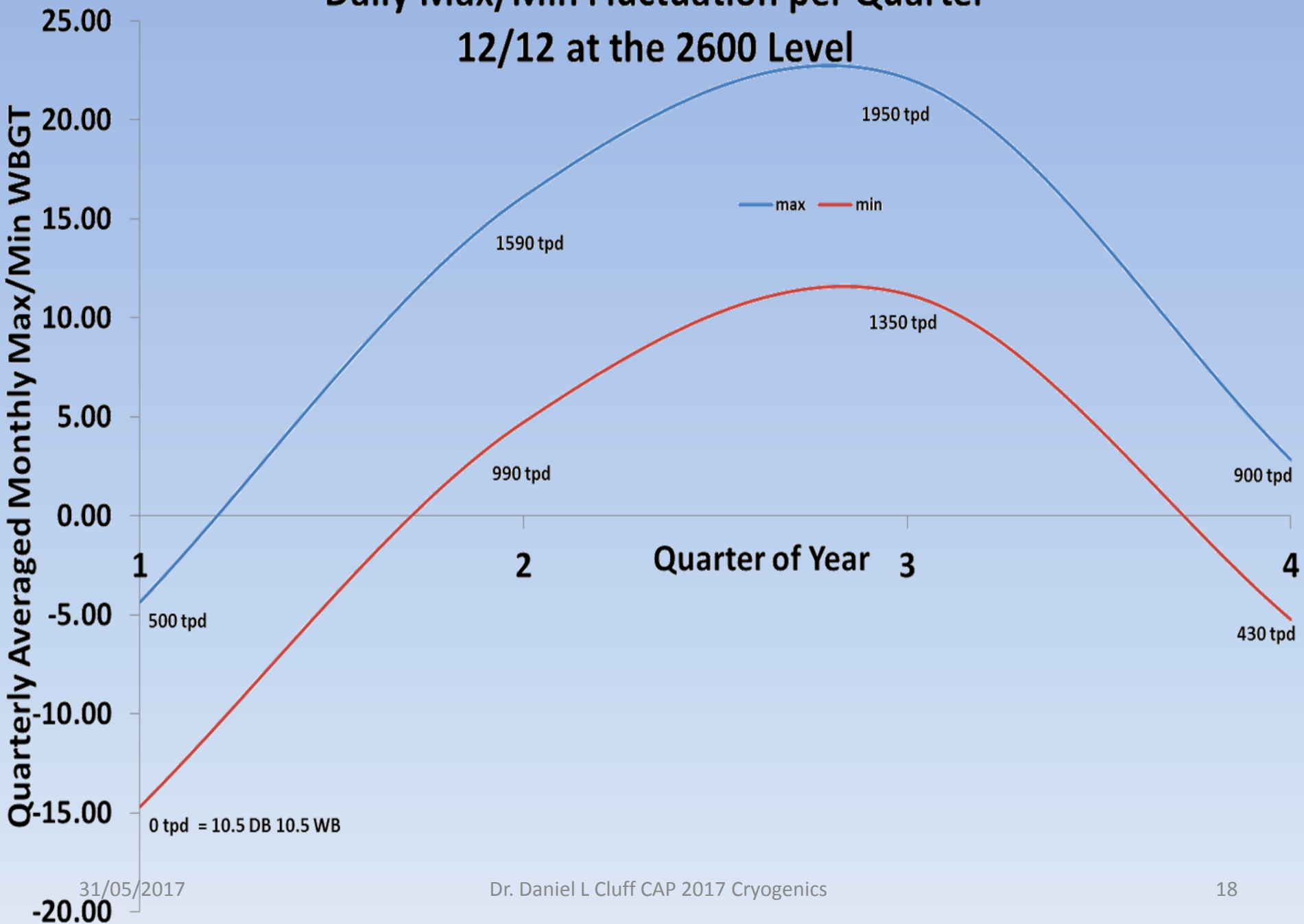
		Surface		Depth		Duty Factor Cooling		
Temperature	Celsius	Farenheit	AC effect	Target	Depth (m)			
DB	28	82.4	46.5983	12	-1915		Revised	
WB	19	66.2	33.71	12	Air Flow required at depth	166 m3/s	152.041	
WBGT	21.7	71.06	37.5765	12	Resulting mass Flow	254.936 kg/s	215.387	
Humidity Ratio	kg-w/kg-a	0.01072	0.02130	0.00697	Surface demand	211.823 m3/s	196.082	
Relative Humidity	%	43.3007	40.255	100	Auto-compression Heat		Other Heat	
Surface elevation (m)		402	124970	GO	Power	8.86582 MW	0	
	PSI	14.009	18.3539		Liq Air Mass flow	19.7743 kg/s	0	
Specific Enthalpy	BTU/lb	31.5508	51.5084	Surface Density	1.11018	Plant impact	1708.5 tpd	0
	kJ/kg	73.3872	119.808	Density at Depth	1.41664	Liq Air Plant Demand	1708.5 tpd	1708.5 x DF
Specific Volume	cu-ft/lb	14.5827	12.0154			Total Chilling	8.86582 MW	8.86582
	cu-m/kg	0.91036	0.7501			Liq Air Mass flow	19.7743 kg/s	19.7743
Delta H	kJ/kg	46.4213				Liq Air Volflow	13.9586 m3/s	
						Total Air Volflow	179.959 m3/s	

Chilling on Demand

Yearly Atmospheric Conditions Sudbury Ontario

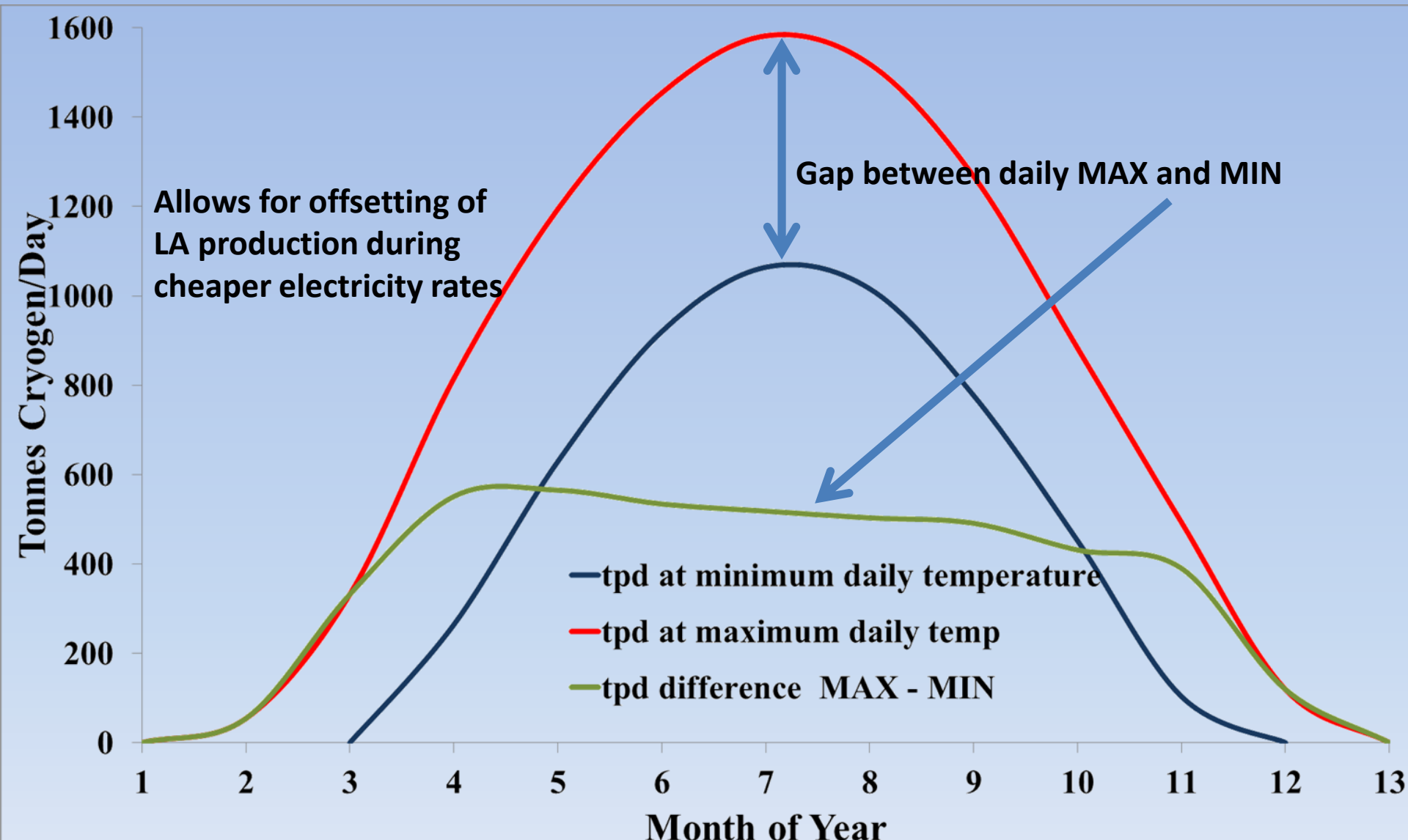


Daily Max/Min Fluctuation per Quarter 12/12 at the 2600 Level



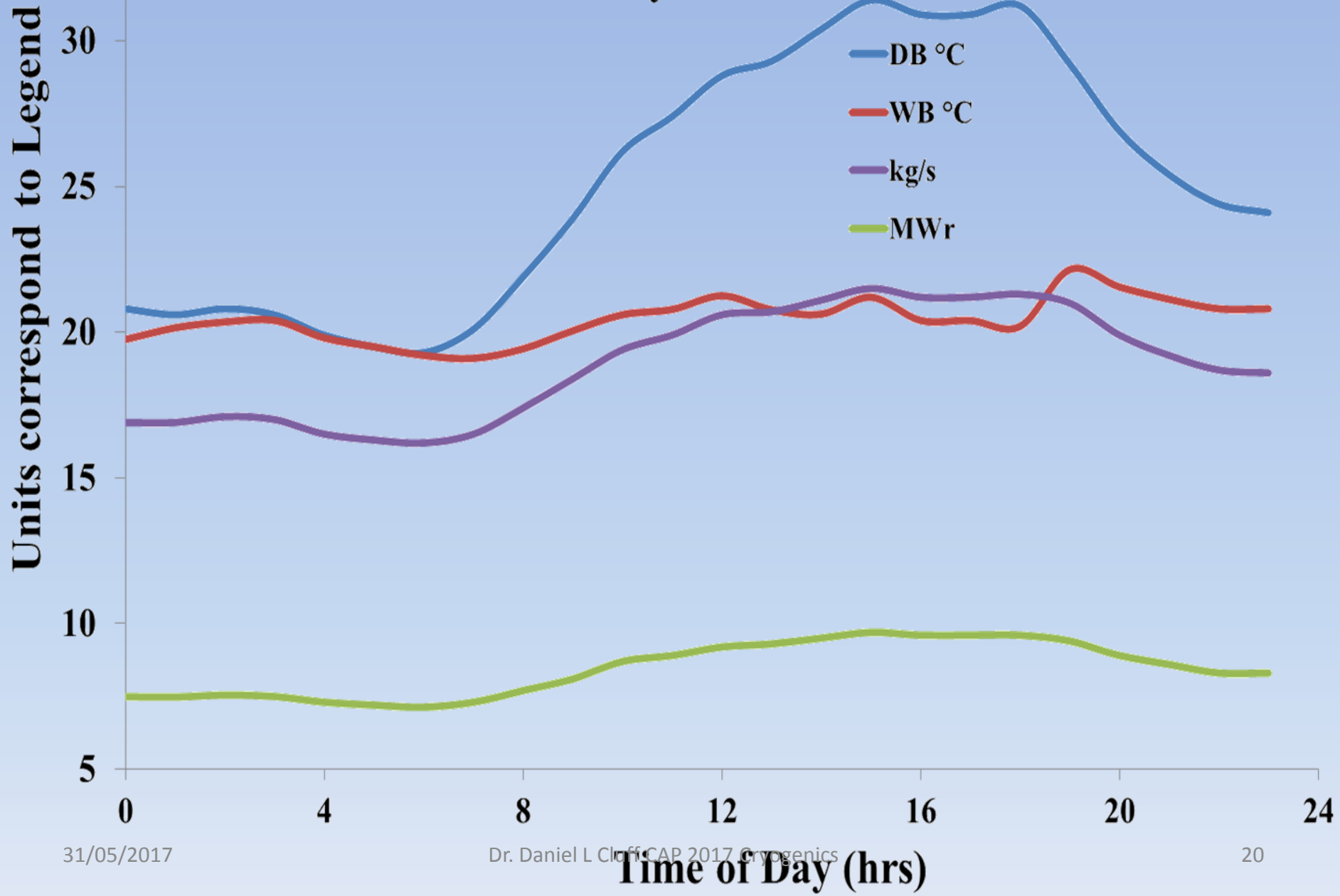
Demand for Liquid Air (tpd)

to Create a 12/12 DB/WB °C Environment at 1915 Depth

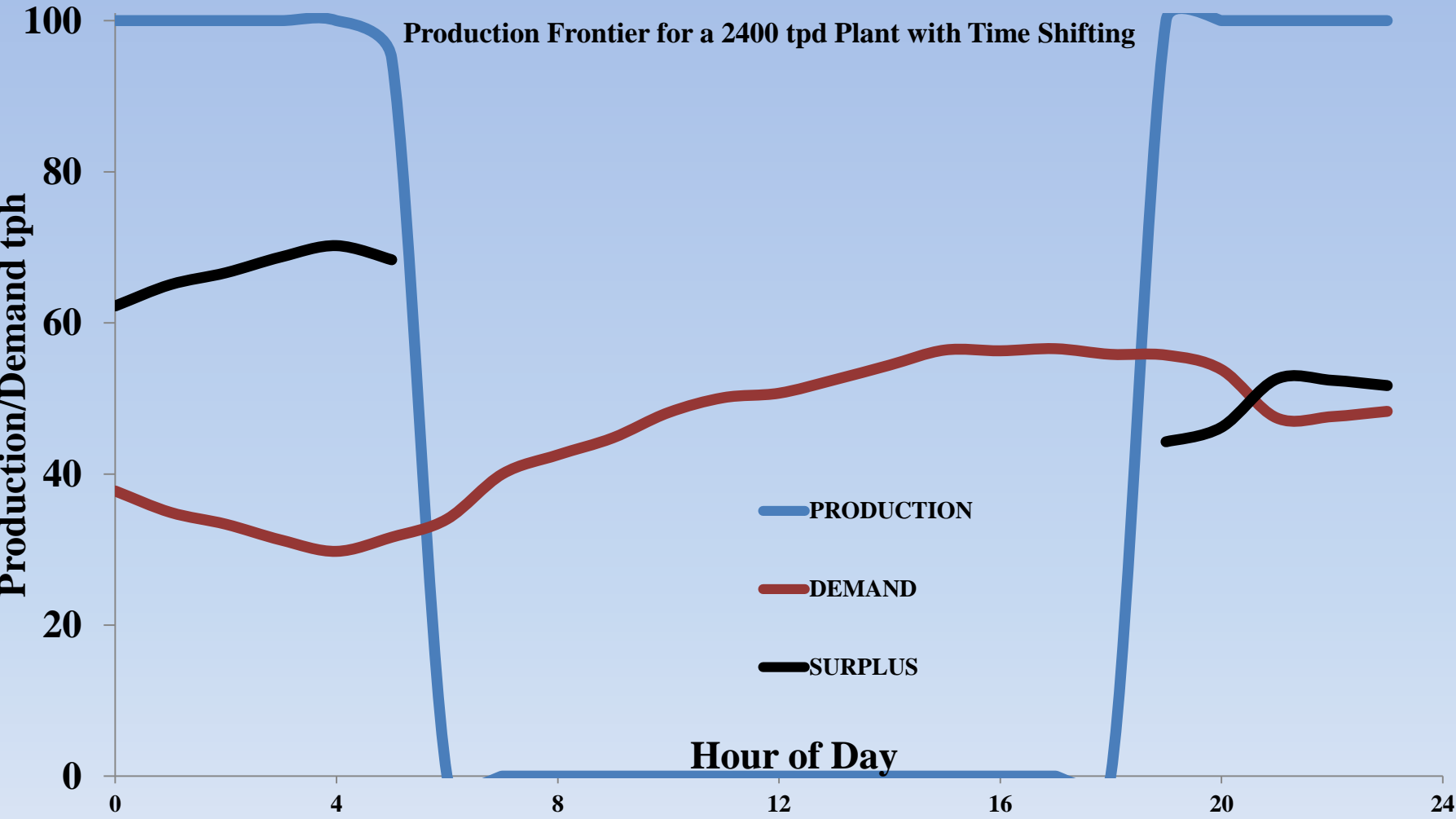


Cryogenic Chilling for Actual Temperature Fluctuations

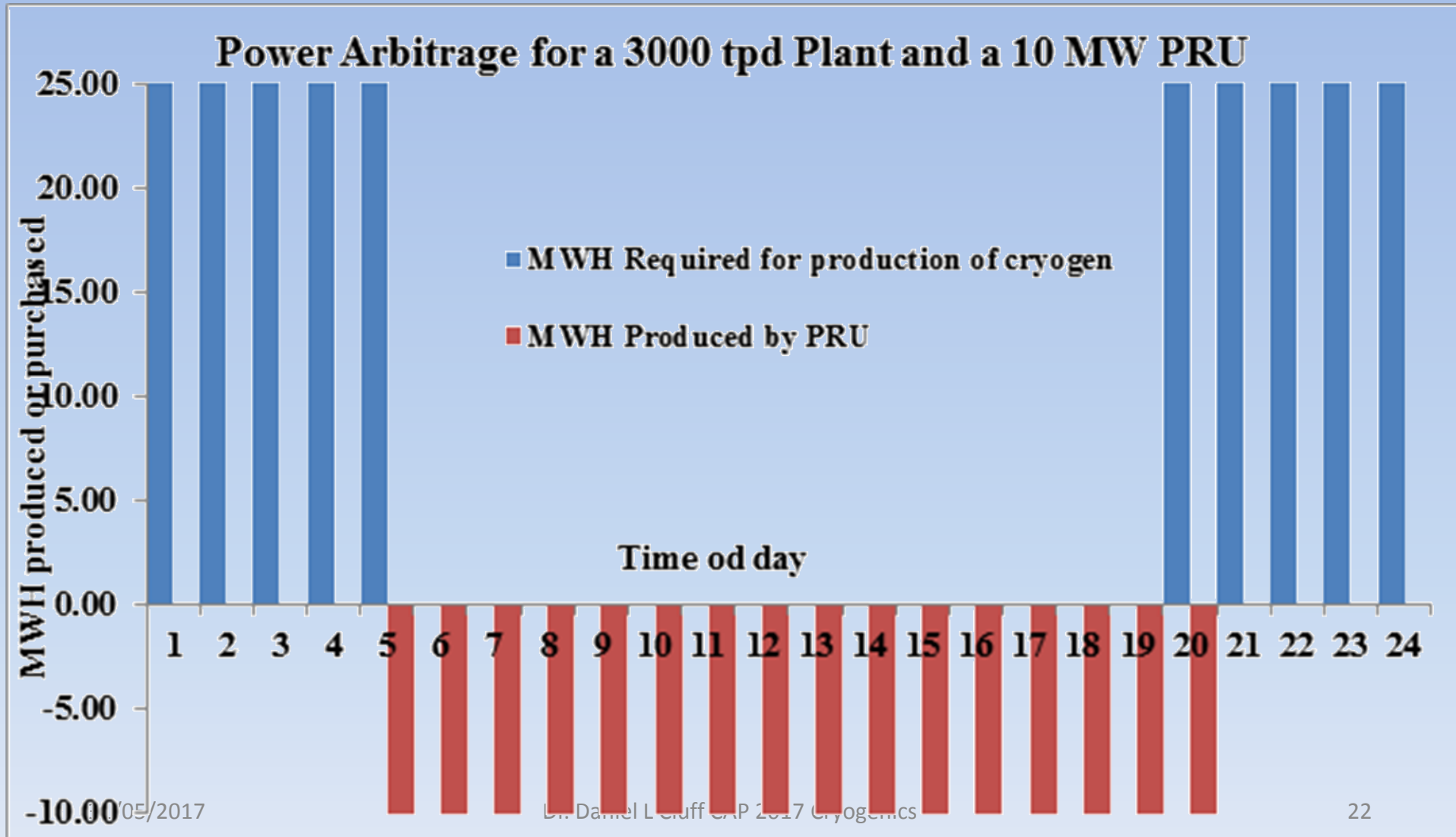
07/07/2010 One of the hottest days in 2010



A 2400 tpd plant production frontier when the surplus produced over the least expensive energy cost is redistributed to the peak time cost



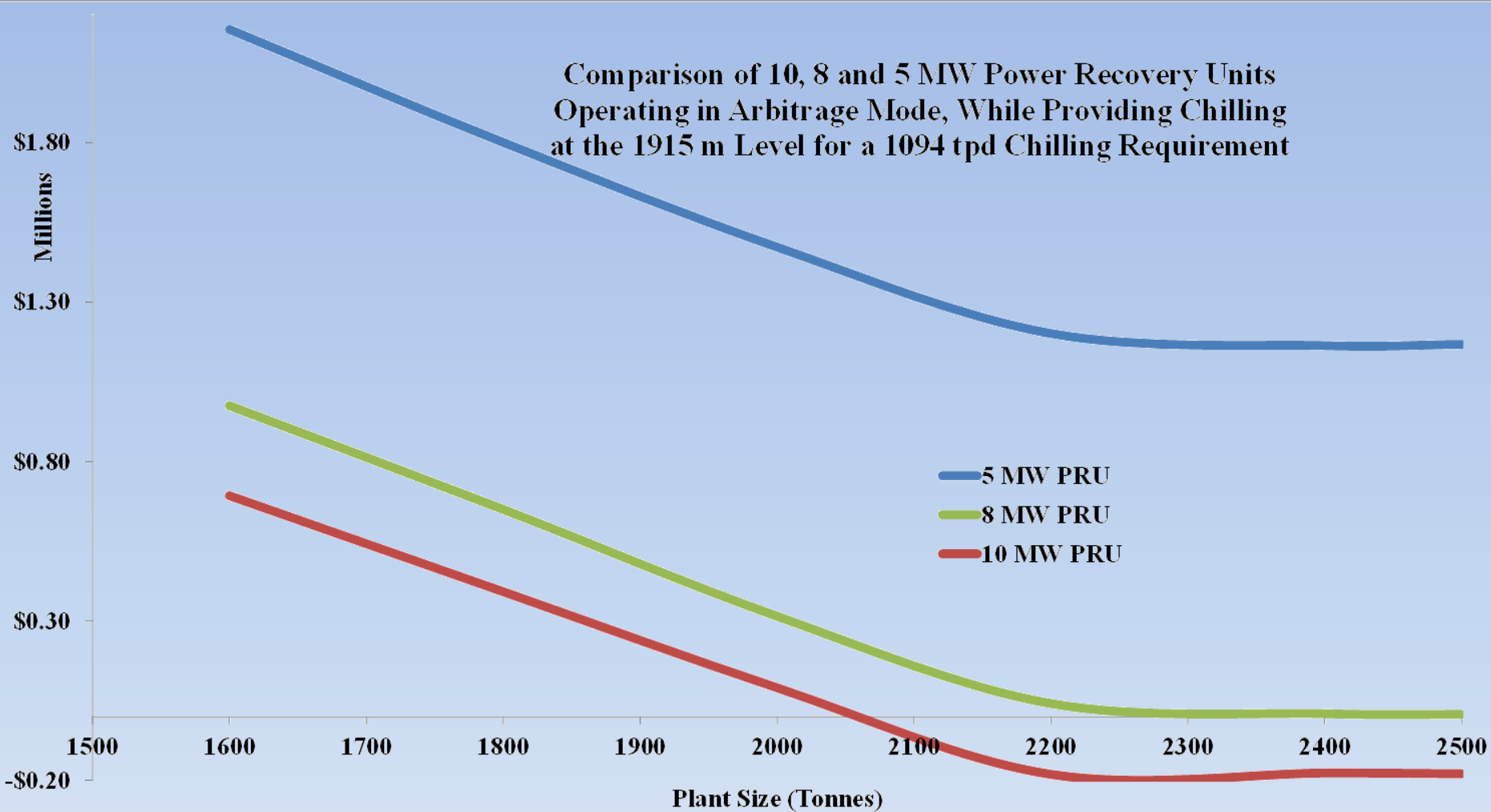
Summary of the cryogen production and power production



Summary of the cryogen production and power production implications for a 10 MW PRU

Plant Size tonnes	1600	1800	2000	2200	2400	2500
Energy Cost	\$3,170	\$2,345	\$1,520	\$775	\$787	\$781
Fan Savings	-\$1,271	-\$1,271	-\$1,271	-\$1,271	-\$1,271	-\$1,271
Final Cost	\$1,899	\$1,074	\$248	-\$496	-\$484	-\$490
YEAR at this rate	\$693,071	\$391,886	\$90,701	-\$180,987	-\$176,752	-\$178,869
MWH paid	218.84	218.84	218.84	218.91	219.04	218.98
10 MWH PRU recovered	-120.00	-120.00	-120.00	-120.00	-120.00	-120.00
FAN MWH recovered	-10.20	-10.20	-10.20	-10.20	-10.20	-10.20
TOTAL	88.64	88.64	88.64	88.71	88.84	88.77
Tonnes produced	1094.00	1094.00	1094.00	1094.33	1095.00	1094.67
Tonnes required	1094.12	1094.12	1094.12	1094.12	1094.12	1094.12
Power Generation	10	10	10	10	10	10
Operating Period	12	12	12	12	12	12
MWH Produced	120	120	120	120	120	120
Efficiency	54.83%	54.83%	54.83%	54.82%	54.78%	54.80%

Cost of Chilling When Time shifting and Using a Power Recovery Unit



Plant Configuration Costing

Plant Configuration	Waste heat	Standalone	Standalone	Waste heat
Liquefaction capacity (tonnes/day)	2500	3000	2000	1700
Power input (MW @ charge time)	21.4	25	16.6	14.3
Charge time hrs	6	6	8	8
Consumed energy MWH	128.4	150	132.8	114.4
Discharge time hrs @ 5 MW	18	18	16	16
Energy output (MWH)	90	90	80	80
Liquid air store capacity (tonnes)	570	570	510	510
Round trip efficiency	70%	60%	60%	70%
CAPEX (million \$)	46	40	33	38
PRU (turbines/generators/grid)	8.15	8.15	8.15	8.15
Storage cost	3	3	2.7	2.7
Cost per kilowatt (\$)	9285	8098	6569	7545
Cost per kilowatt-hour (\$)	516	450	411	472

Ancillary Systems

- Chilling accounts for a major share of consumption
- There are a number of other services that can be implemented that provide a service while also simultaneously chilling as a side benefit.
 - Compressed air
 - Electricity production
 - Vehicles, pumps or fans that can be driven by liquid air

Compressed Air

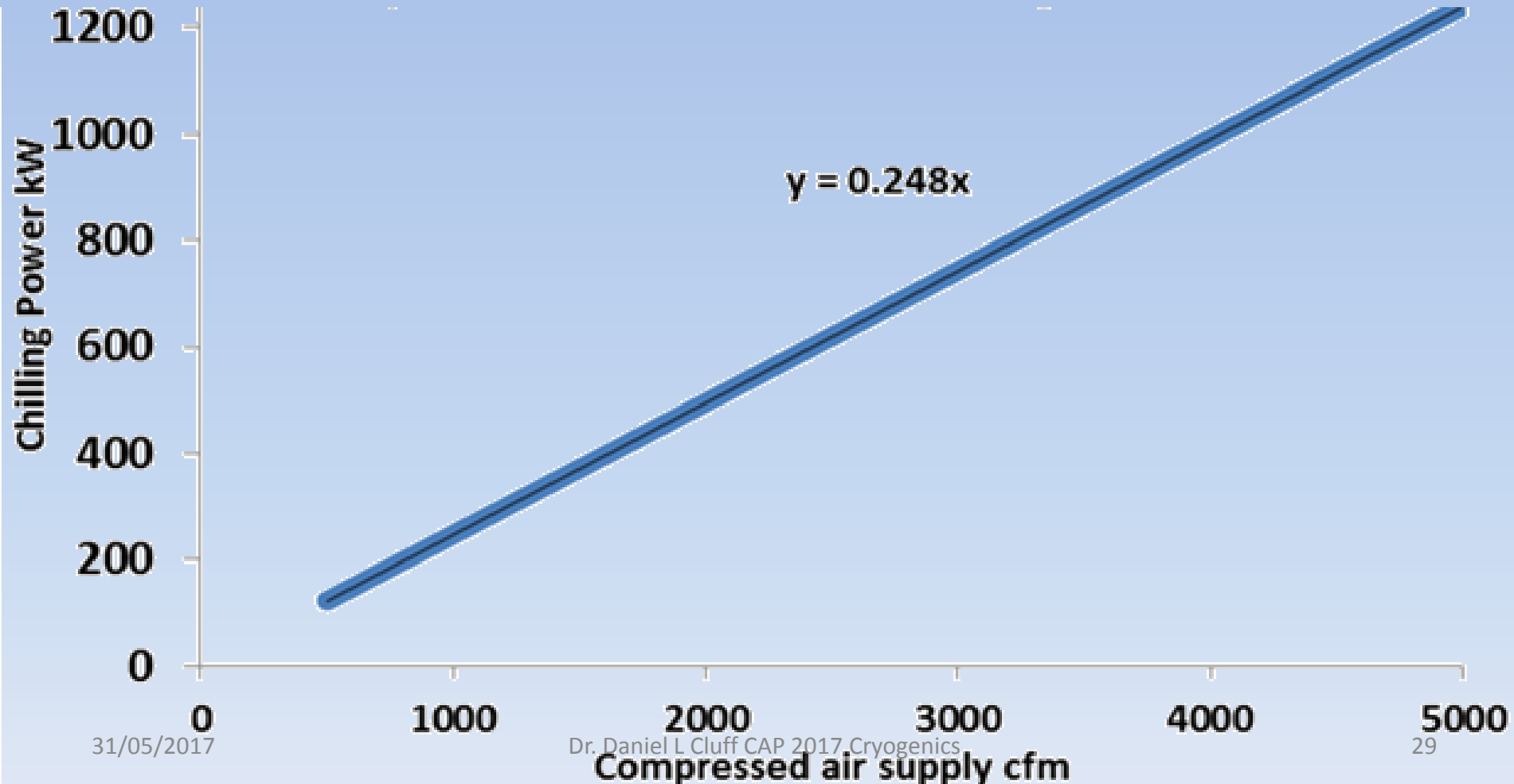
- The production of compressed air exploits the liquid nature of the cryogen, which is simply squirted into a receiver tank and allowed to reach ambient temperature – quickly!

Compressed Air Supply With Chilling Power

Compressed Air Consumption		Receiver Size	Chilling Power	<u>kW</u> ft ³ /min	Mass flow	Liq flow	Plant impact
ft ³ /min	m ³ /hr	m ³	kW		kg/s	l/s	tpd
500	850	1.79	124	0.248	0.29	0.33	25
1000	1699	3.6	248	0.248	0.58	0.67	50
2000	3398	7.28	496	0.248	1.16	1.33	100
3000	5097	11.06	744	0.248	1.74	2.00	150
5000	8495	18.54	1240	0.248	2.9	3.33	250

Assuming continuous demand the liquid air can be configured to provide a modular compressed air system which will simultaneously chill at the location the receiver tank is located akin to spot chilling.

Chilling Power as a function of the Compressed Air Supply



Recall the earlier LAES Simplified Schematic

- In the schematic it was indicated that the PRU could be placed underground.
- Part of the PRU process is to pump the liquid air to a pressure of about 1000 psi before evaporating and expanding through the turbomachinery.
- At 2000 m the pressure is about 2500 psi
- Sufficient pressure to eliminate the pumps

Recall the earlier discussion regarding the gap between the MAX and MIN daily temperature was about 500 tpd, Below for a 5 MW_e PRU

	Waste heat	Standalone	Standalone	Waste heat
Liquefaction capacity (tonnes/day)	2500	3000	2000	1700
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Storage cost	3	3	2.7	2.7
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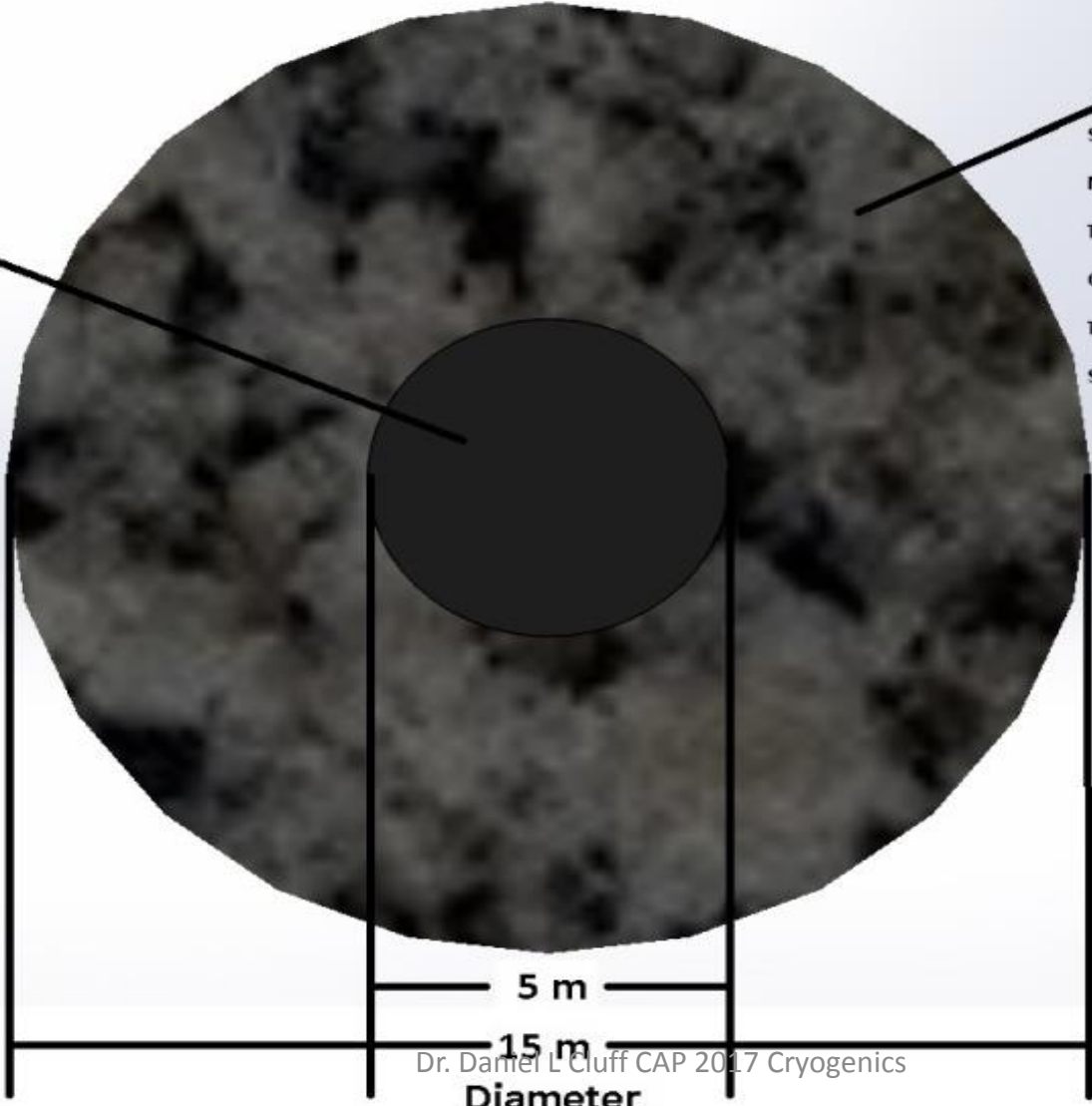
Typically the cold would be recycled to cold storage, but here it is absorbed by the air at depth

Exploitation of the Joule Thompson Effect

- With 2500 psi available as a forcing pressure the Joule Thompson Effect, which is part of the liquefaction process, can be exploited to provide further chilling commonly referred to as free expansion or a throttling process.
- $T_2 = T_1 - U_j(P_1 - P_2)$, $P_2 = 14.5 \text{ psi}$, $T_1 = 25^\circ\text{C}$
- where U_j is the JT coefficient - ambient T_1 - contained P_1
- For a contained pressure of 880 psi, $U_j = 0.1815$
- $T_2 = 298 - 0.1815(880 - 14.5) = 140.9 \text{ K} = -132.05^\circ\text{C}$

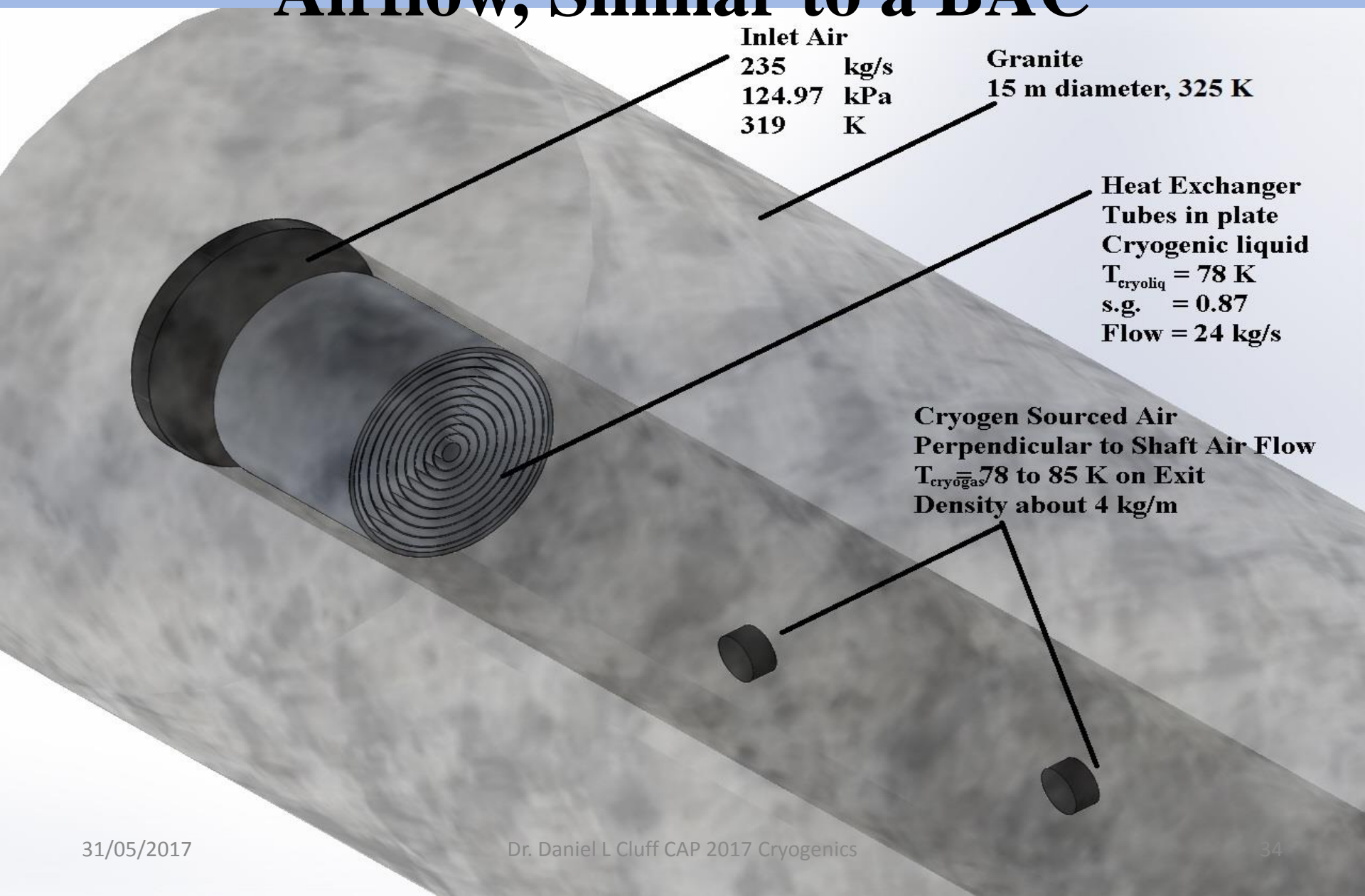
CFD Model for Chilling the Entire Airflow, Similar to a BAC

Inlet Air
235 kg/s
124.97 kPa
319 K

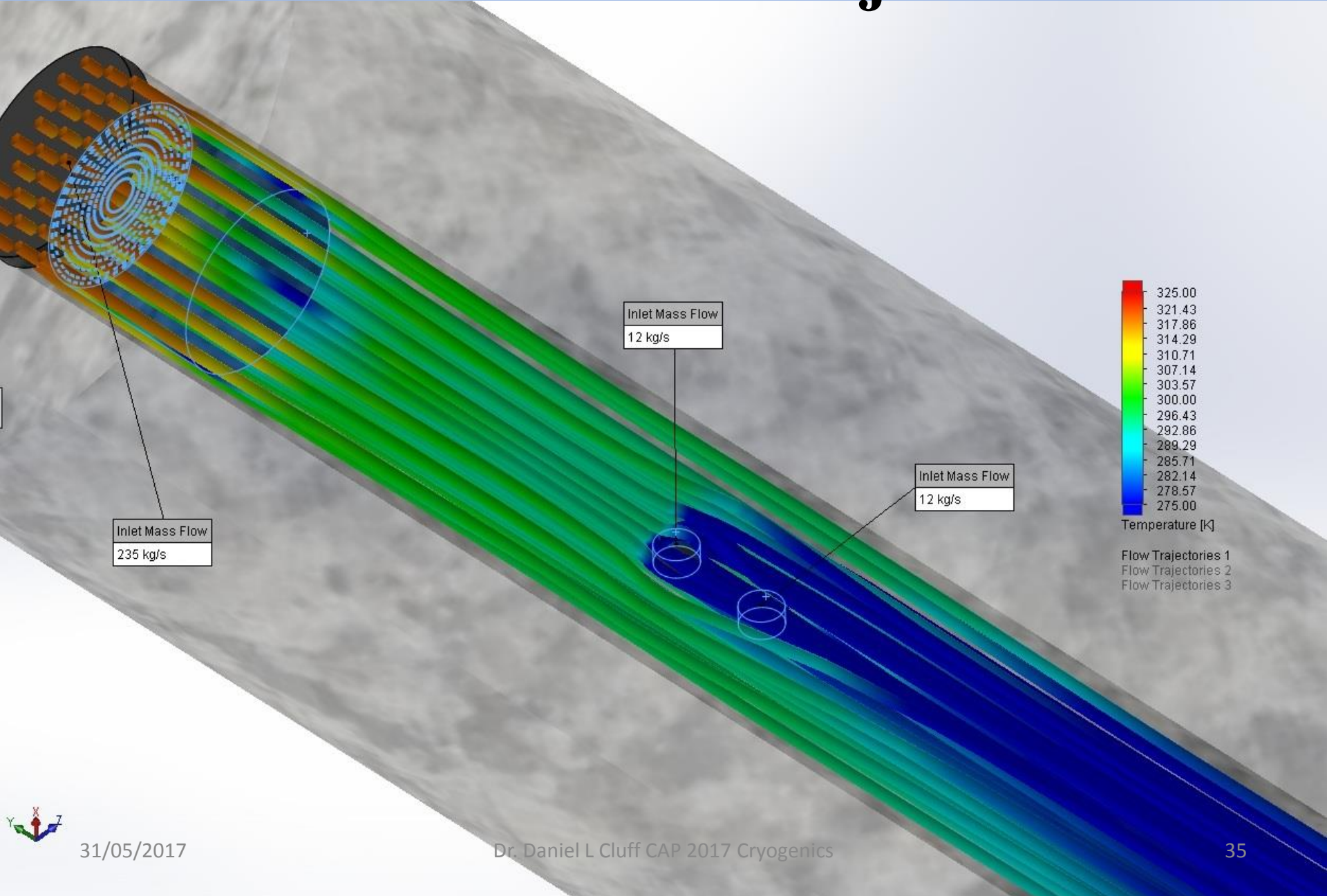


Granite		
Shear Modulus	318.9	MN/m ²
Mass Density	2700	kg/m ³
Tensile Strength	30	MN/m ²
Compressive Strength	200	MN/m ²
Thermal Conductivity	3.5	W/(m·K)
Specific Heat	1	J/(kg·K)

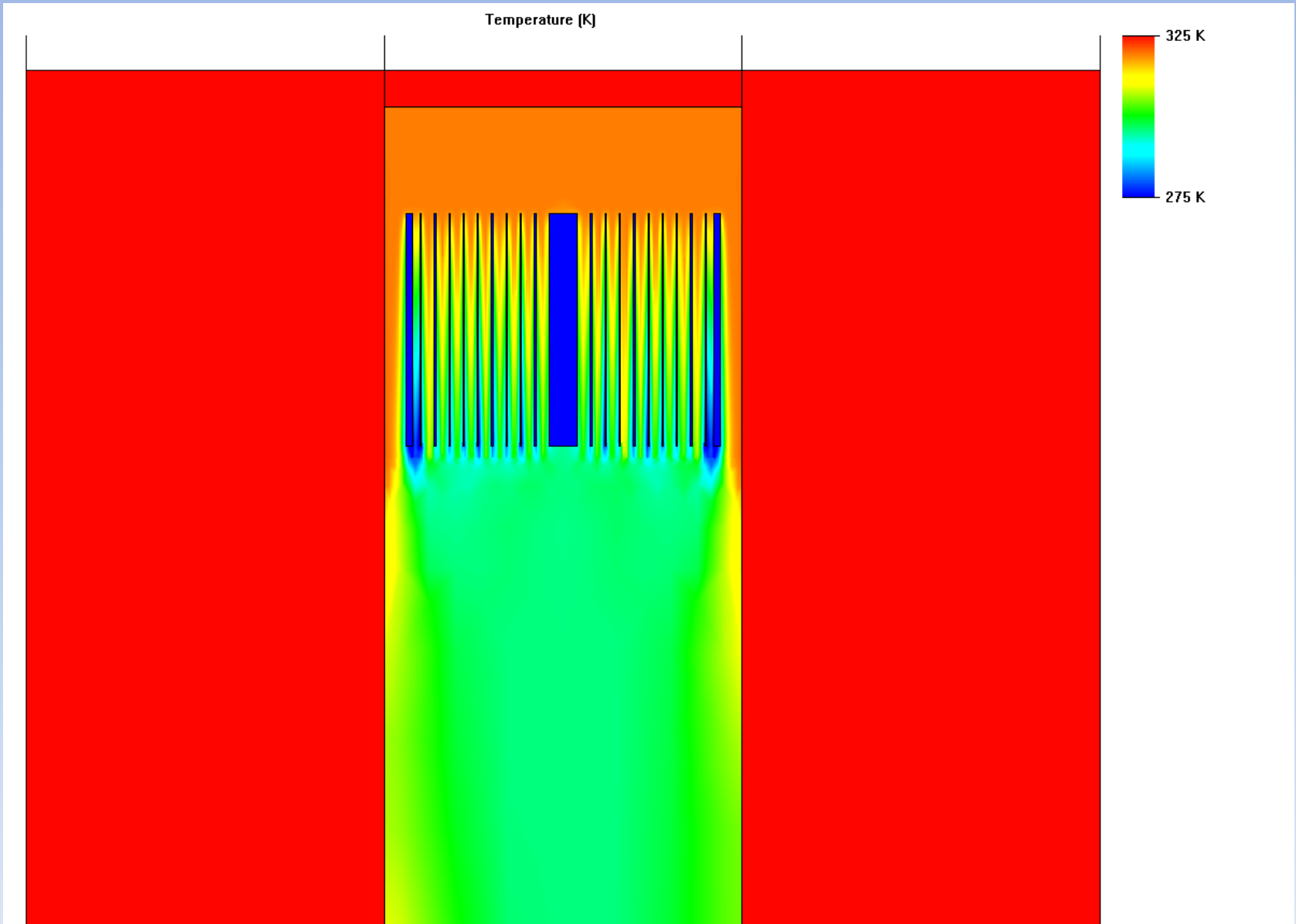
CFD Model for Chilling the Entire Airflow, Similar to a BAC



CFD Model Flow Trajectories



Close up View of Heat Exchanger 1.67 sec



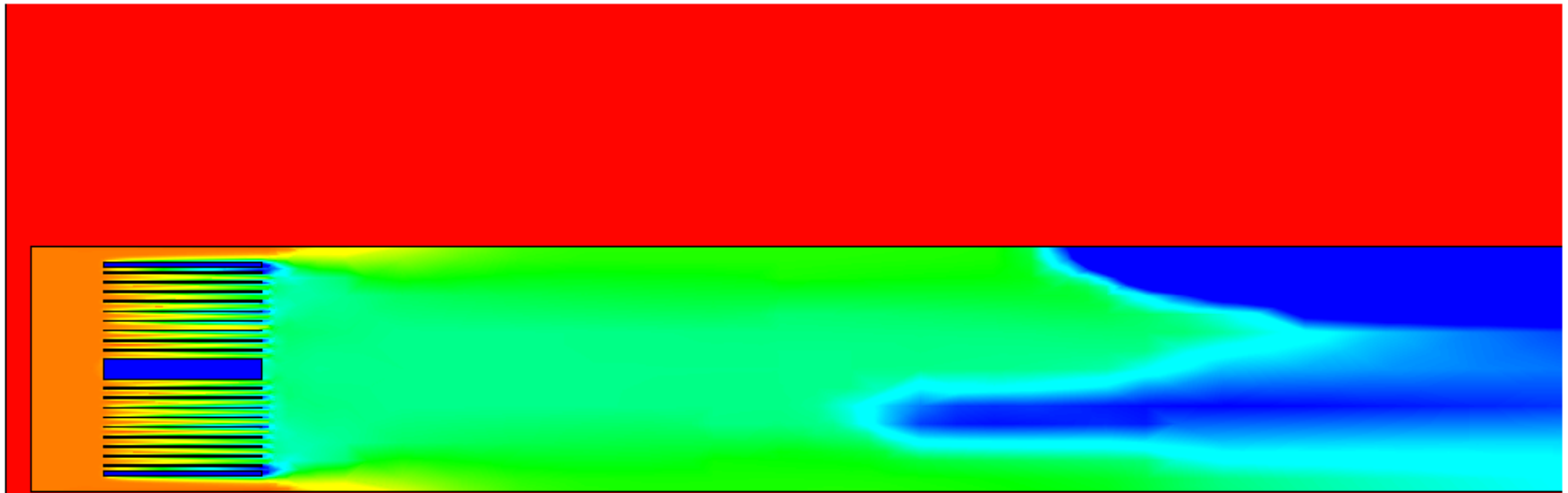
31/05/2017
Min=58.8319 K Max=325.003 K
Time = 1.67701435 s

Dr. Daniel L Cluff CAP 2017 Cryogenics

Close up View of Heat Exchanger 8.2 sec

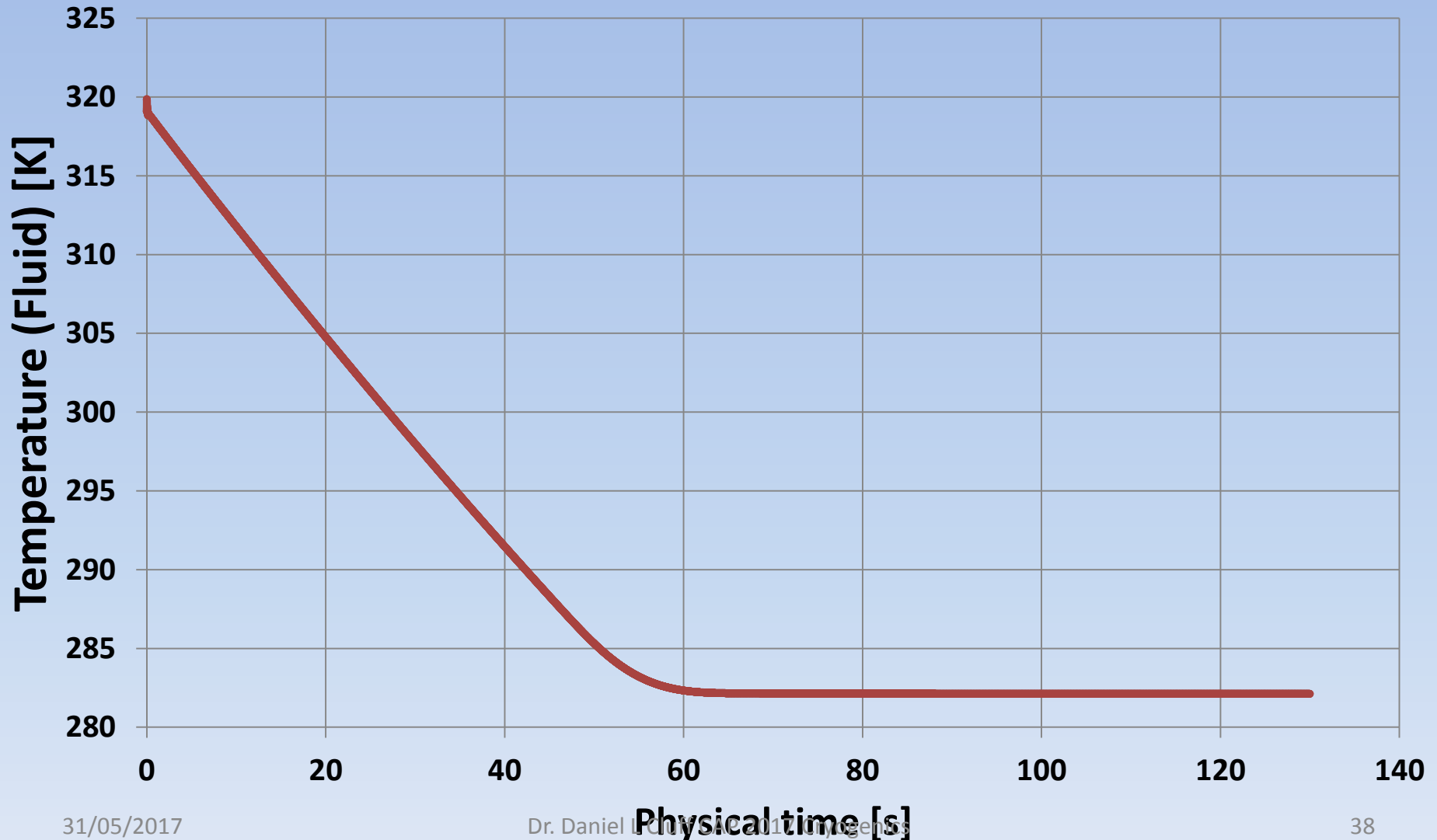


Temperature (K)



Average Temperature of Air in Shaft

500 m deep cyl assemb 3.SLDASM [500 m deep [500 m]]



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Thank you to:

Glencore for Financial Support and Engineering Excellence.

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CEMI for business expertise and project management

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Camborne School of Mines for academic support

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