

A decorative graphic on the left side of the slide, consisting of several concentric, overlapping circles of purple dots connected by thin purple lines, resembling a molecular or atomic structure. The dots are arranged in a grid-like pattern within each circle.

RADON REMOVAL STUDIES FOR NEXT-GENERATION XENON TPCS

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LUX-ZEPLIN



World-leading WIMP dark matter experiment

Dual-phase xenon **T**ime **P**rojection **C**hamber

7 tonnes active liquid xenon target

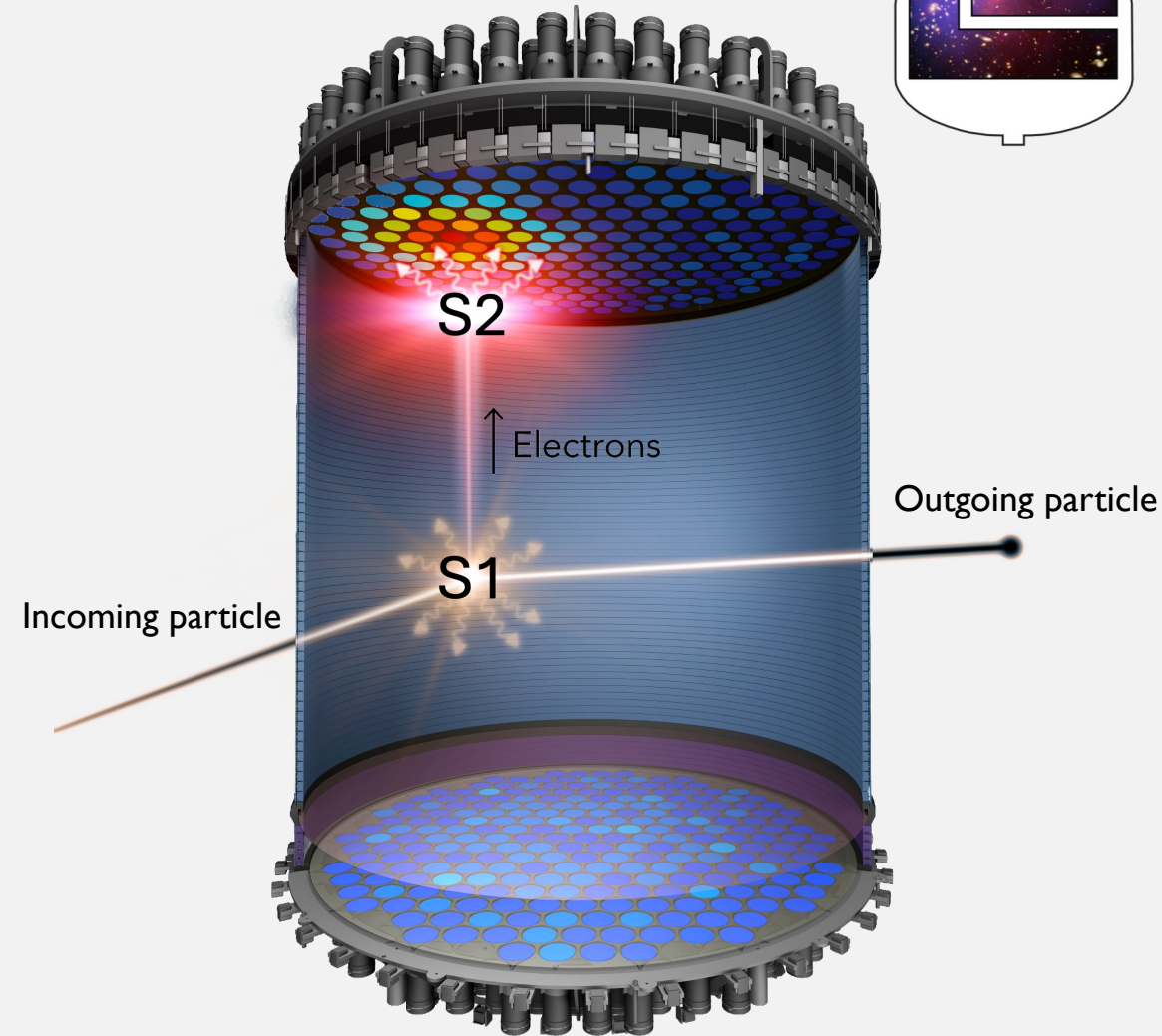
Two signals:

S1 - prompt scintillation photons

S2 - ionisation electrons

$S1/S2$ ratio \rightarrow Electronic (ER) / nuclear recoil discrimination (NR)

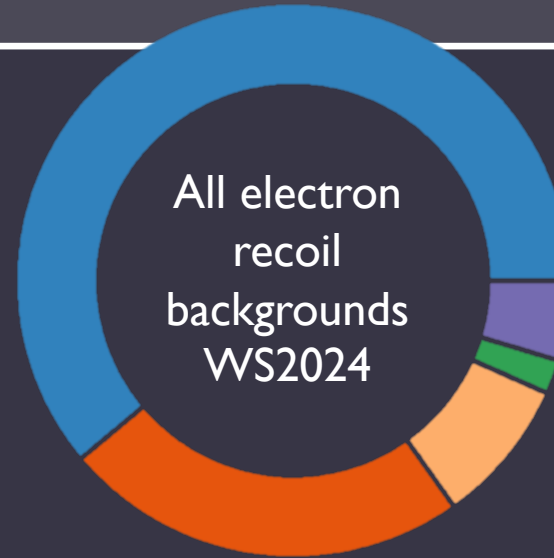
WIMPs expected to interact via NRs



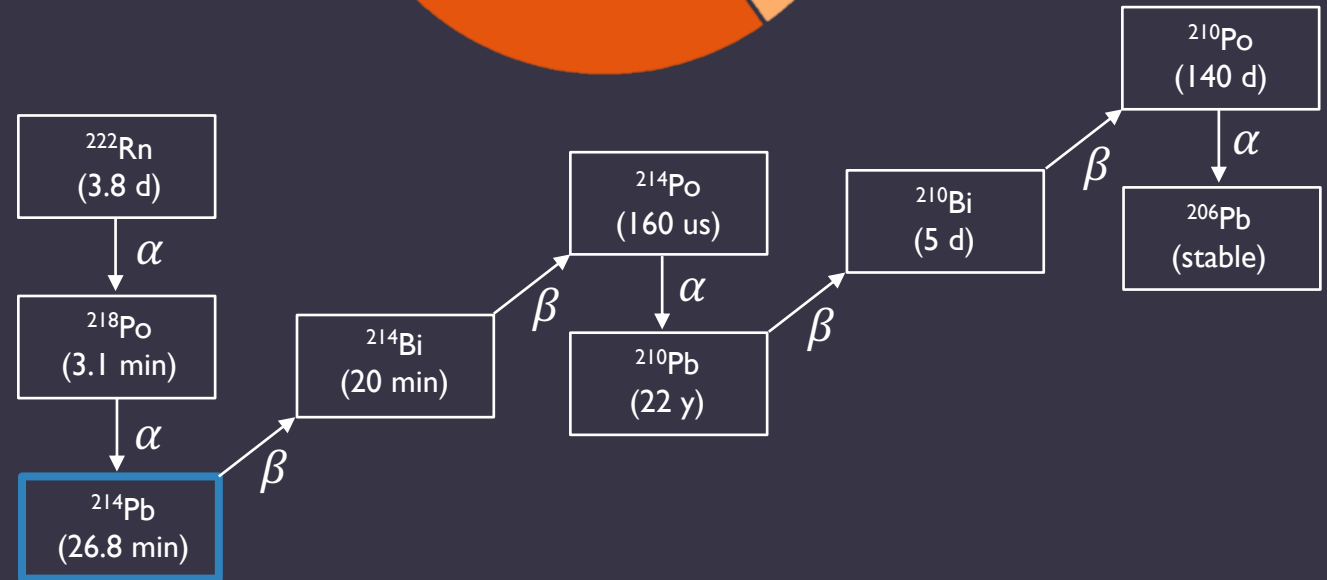
RADON

- ^{238}U naturally exists in most materials
- Decays via a sequence to ^{222}Rn
- Radon is
 - A chemically inert noble gas
 - Hard to shield against
 - Difficult to remove chemically
 - Long half life ($\tau_{1/2} \text{ } ^{222}\text{Rn} = 3.8$ days)

→ Emanates from detector materials into the active volume



^{214}Pb β s
 ^{136}Xe $2\nu\beta\beta$
 Other β s + material γ s
 Solar ν ER
 $^{127}\text{Xe} + ^{125}\text{Xe}$ EC + ^{124}Xe DEC



^{214}Pb : Dominant background (~60% of ER)
 can appear WIMP-like

WHAT DO WE CURRENTLY DO?

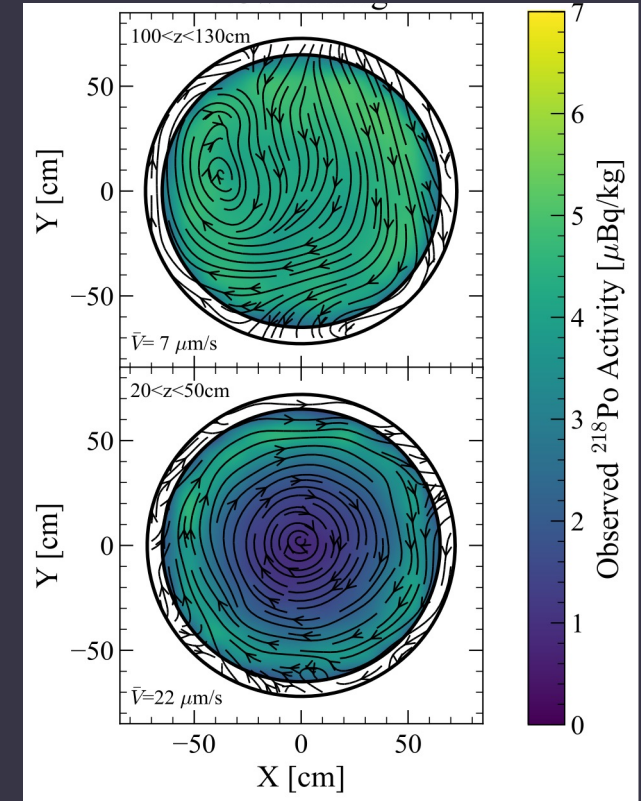
Multiple strategies employed to target radon background

- 5 year materials screening campaign was conducted [1]
- WIMP search 2025 introduced a data-driven technique to our analysis: **flow-based radon tagging** [2]
- **Inline radon removal system** in circulation loop reduces radon concentration in gaseous xenon

Techniques together reach untagged activity levels of $1.8 \mu\text{Bq}/\text{kg}$



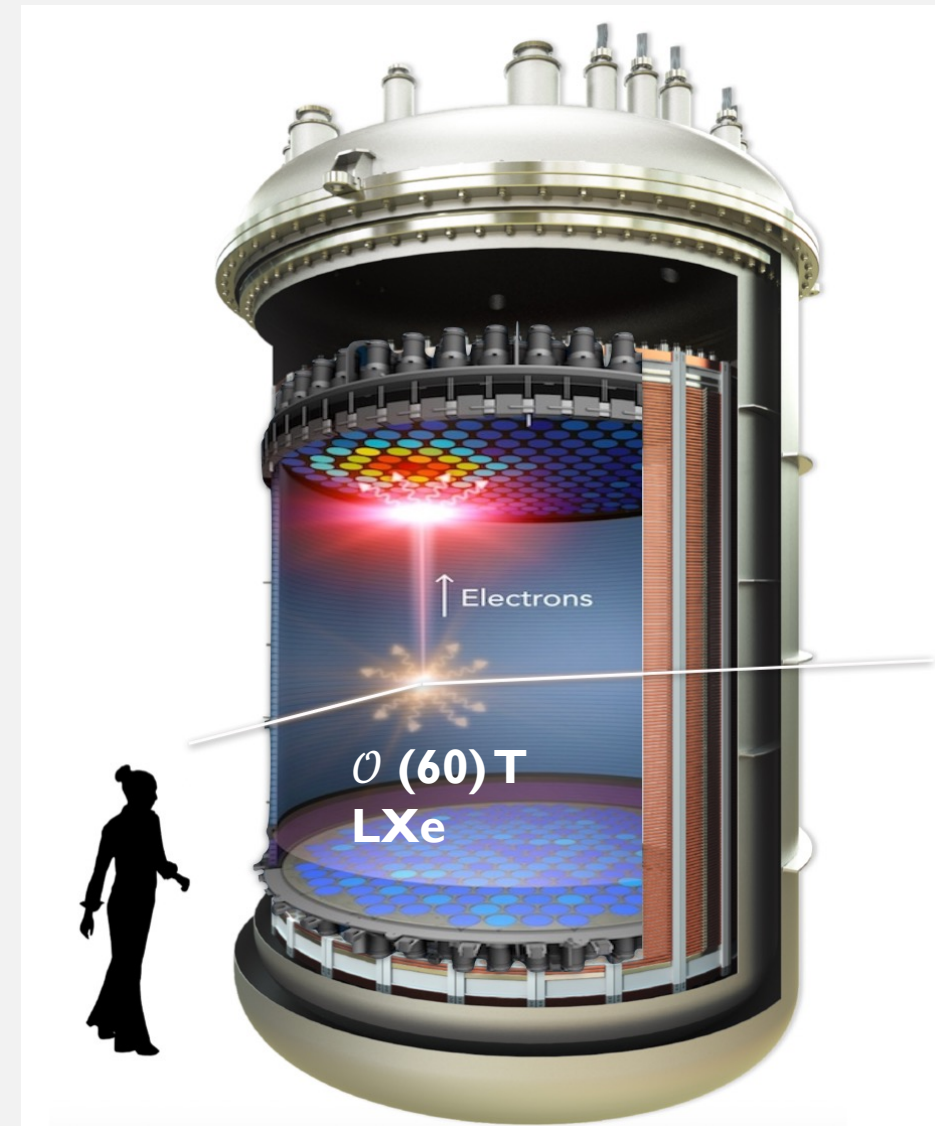
Screening for radon contamination
[1] <https://doi.org/10.1140/epjc/s10052-020-8420-x>



Xenon flow plots – used to track the path of radon decay products
[2] <https://doi.org/10.48550/arXiv.2508.19117>

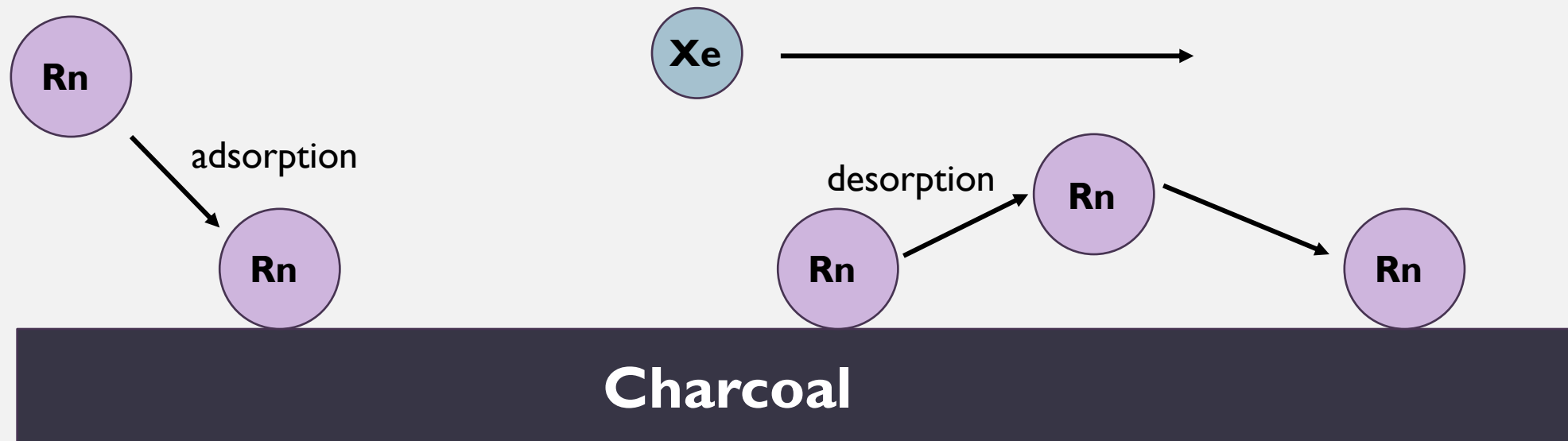
XLZD

- Next-generation dual-phase xenon TPC
 - Combine the expertise built up by XENON and LZ experiments and the DARWIN R&D project
 - $\mathcal{O}(10)$ times the size of current-gen TPCs
 - Observatory for rare events- not just WIMP dark matter
 - Requires ^{222}Rn activity $\sim 0.1 \mu\text{Bq/kg}$ for sensitivity to $0\nu\beta\beta$
- More stringent background controls required. One way of achieving this could be upgrading the inline radon removal system (iRRS).



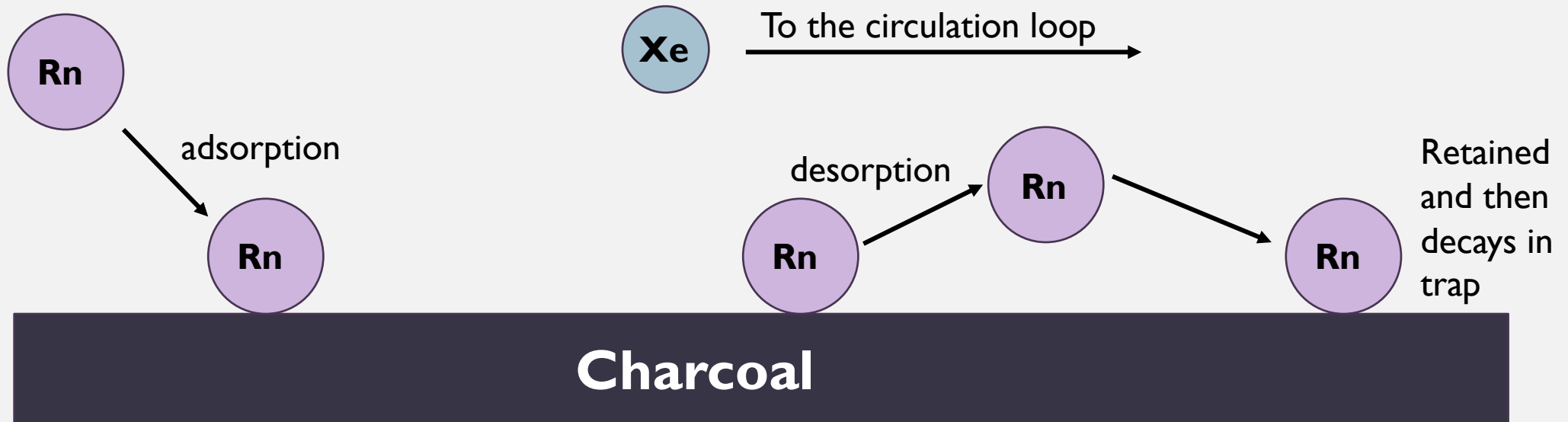
THE IRRS: CHROMATOGRAPHY BASICS

- Chromatography is a technique to separate a mixture into its components
 - The different components can interact with a chromatography medium differently, causing some to adsorb (stick) to the surface more than others.
 - Radon and xenon interact with carbon via the van Der Waals interaction
 - Radon is a larger atom so it is more polarisable → stronger van Der Waals
- **It is adsorbed more easily**



THE IRRS: SCALING UP

- LZ iRRS consists of a single adsorption trap which takes gaseous xenon from the warm regions of the TPC, and passes it through a charcoal chromatography column
 - Radon is retained long enough (~ 3 half-lives) such that it decays while in the trap
 - Purified xenon is returned to main circulation loop
 - Charcoal is also a source of radon! Scaling up the system to XLZD size: charcoal needed would emit more radon than it removes
 - Removing radon purely from gaseous regions would not meet XLZD requirements
- An alternative approach is necessary



SWING SYSTEMS

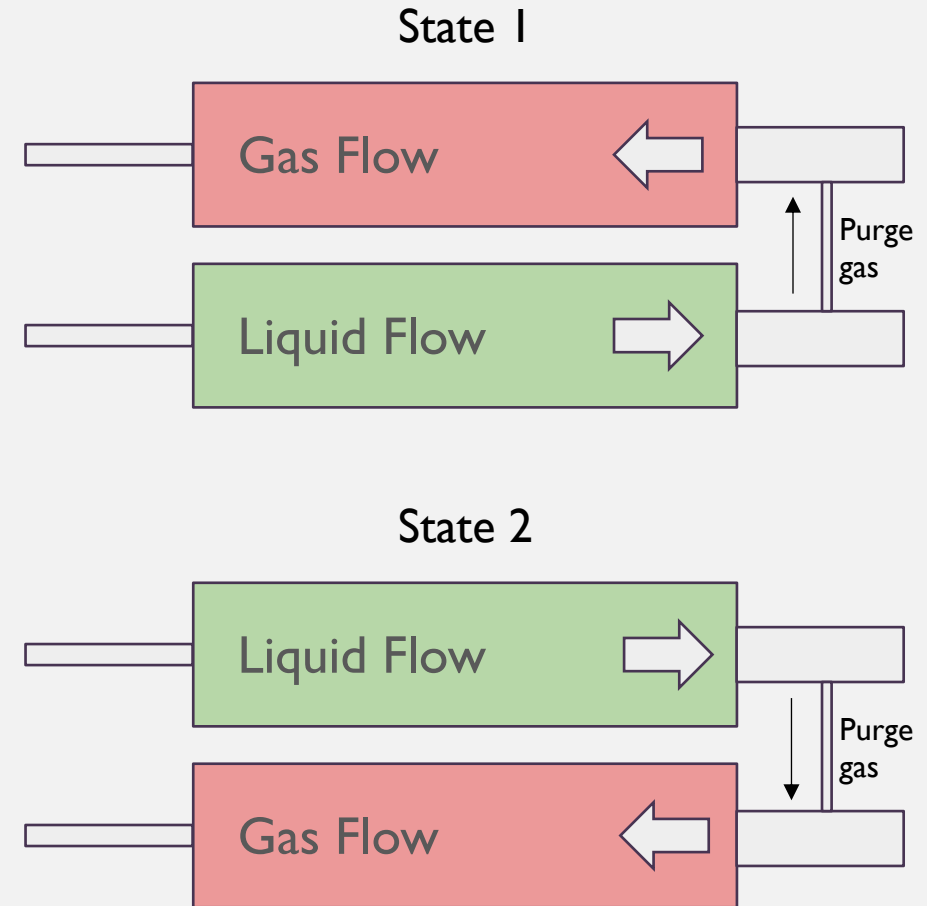
State 1:

- Radon removed from liquid xenon in one column
- Most of the purified xenon returned to circulation loop
- Some is passed through second column to purge it

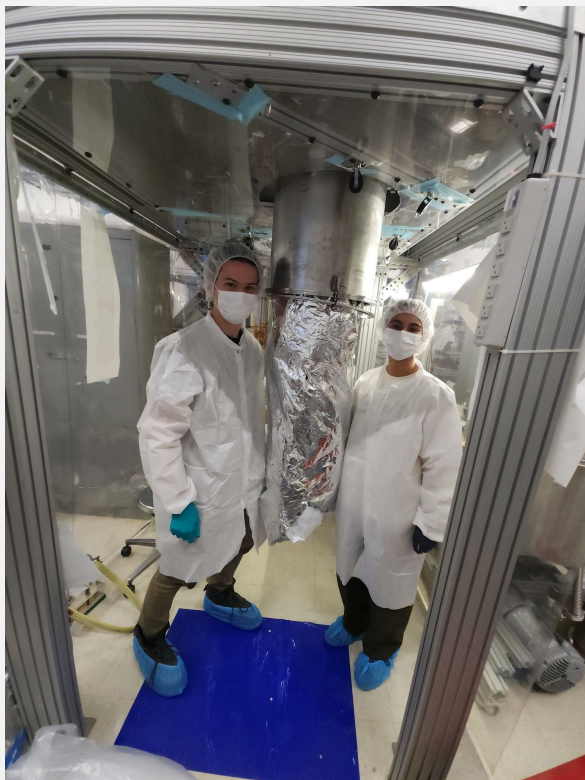
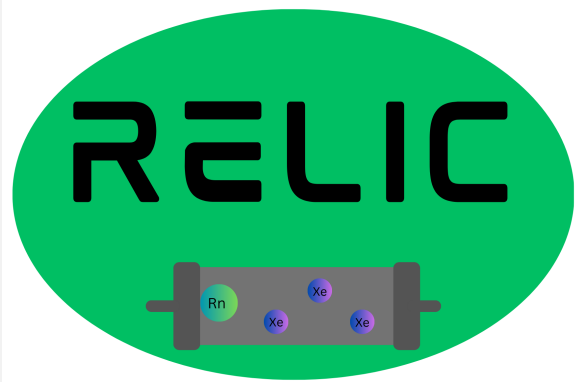
State 2:

- Flow switches: now purging column where radon was removed and using the other column to purify
- ✓ Continuous radon removal
- ✓ Can be part of main circulation loop
- ✓ System can be smaller → less charcoal required!

This all rests on the ability to use chromatography to remove radon from *liquid* xenon - which has never been tested before



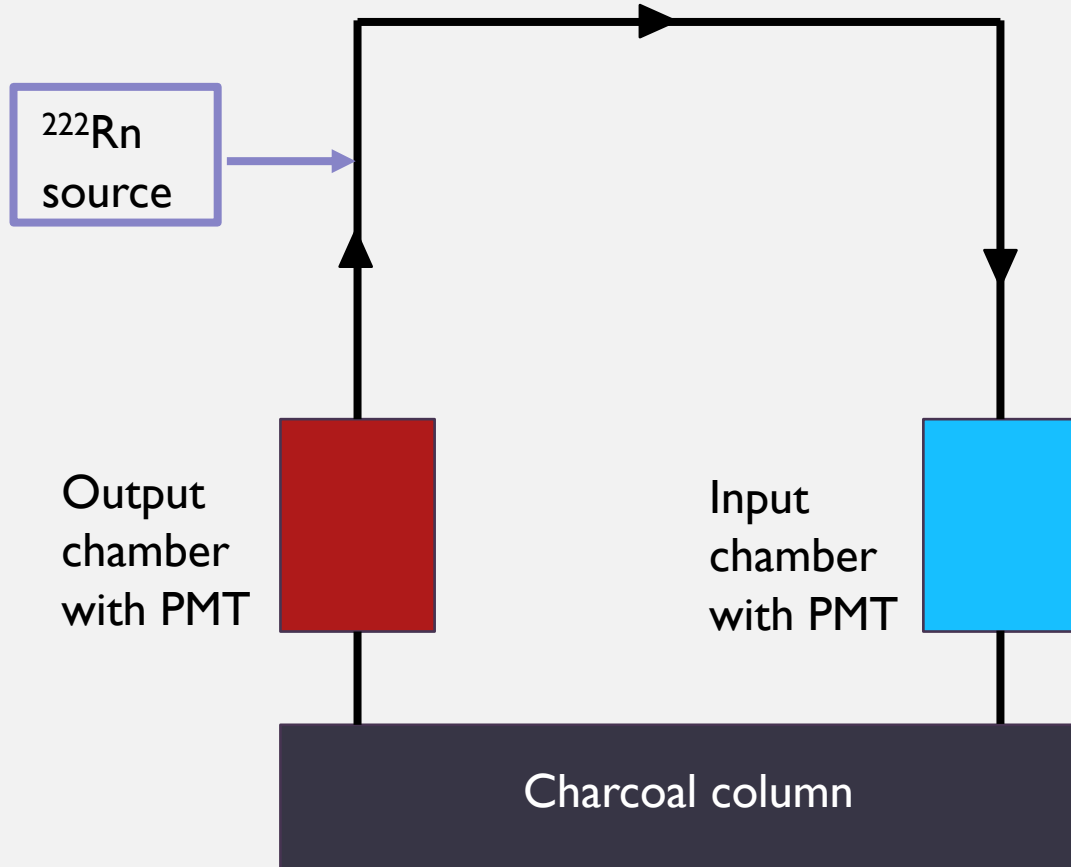
INTRODUCING: RELIC



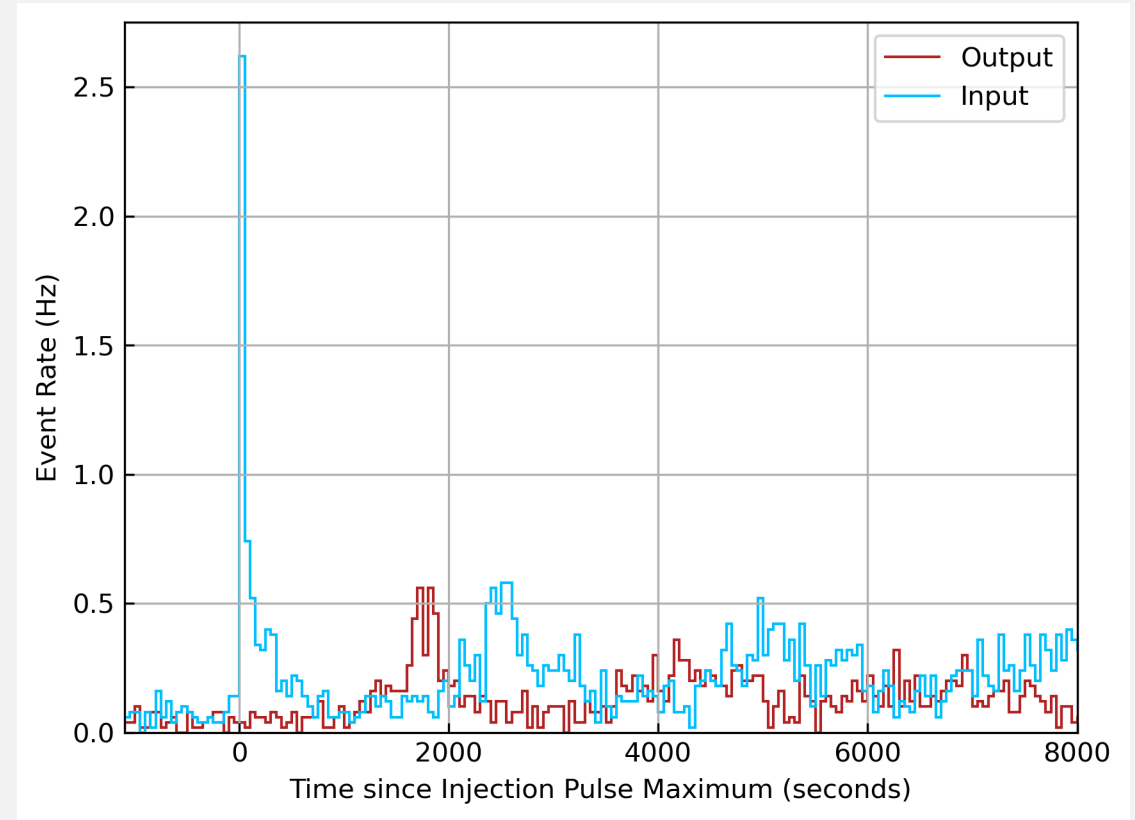
- Efficacy of radon removal is described by the adsorption coefficient, k_a (l/g)
- Measures how well radon binds to charcoal relative to xenon
- ReLIC: **R**emoval of **L**iquid **I**mpurities via **C**hromatography
- Test setup based at SLAC, US National Lab
- Designed to measure k_a

GAS MEASUREMENTS

Circulation loop: continuously flows Xe and Rn through the system



Simplified diagram of the setup

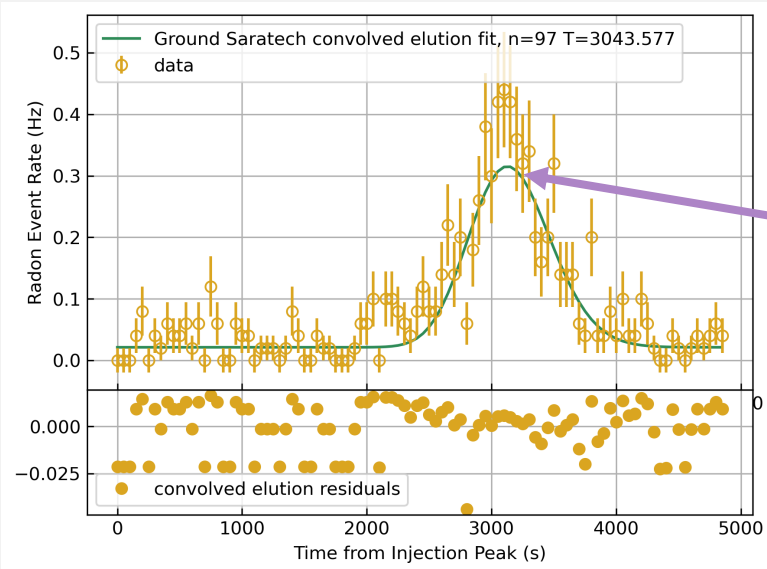
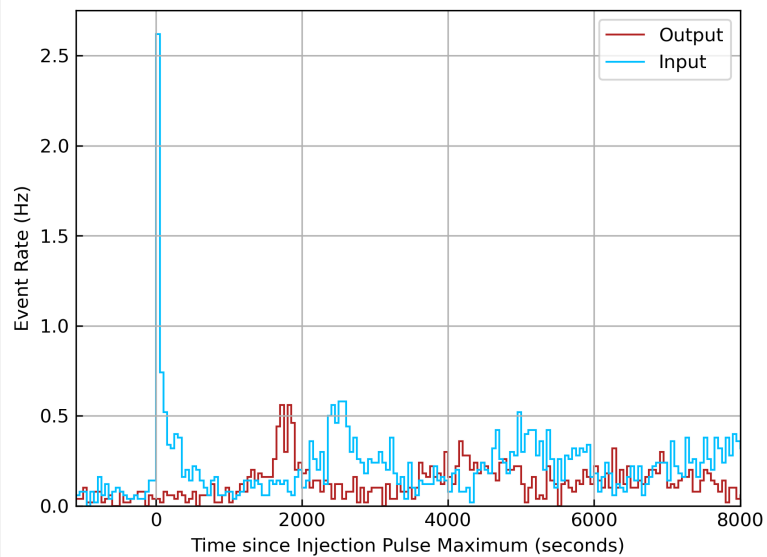


Radon decay rate in the two PMTs

Blue: radon event rate before charcoal column

Red: radon event rate after charcoal column

GAS MEASUREMENTS



- Adsorption coefficient depends on the breakthrough time – how long it takes radon to pass through the column relative to xenon
- This can be extracted from the time difference between the **input** and **output** pulses
- **Input** pulse in gas is sharp- we take this time to be the same as time of injection
- A curve* is fit to determine the time of the **output** pulse. This procedure will be especially useful in the liquid results

* *Elution curve*. A standard curve that material passing through a chromatography column should follow 10

GAS MEASUREMENTS

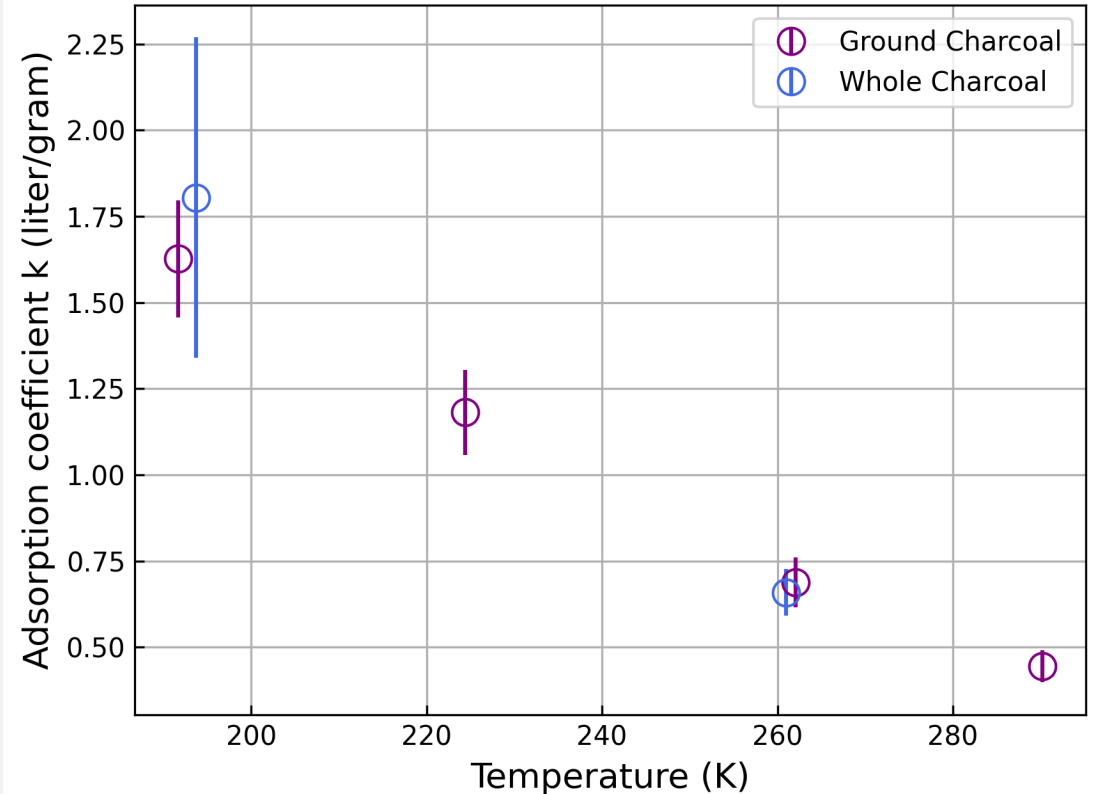
Breakthrough time
from elution curve

Xenon flow rate

$$k_{gas} = \frac{T \times \phi}{M}$$

Our parameter of
interest!

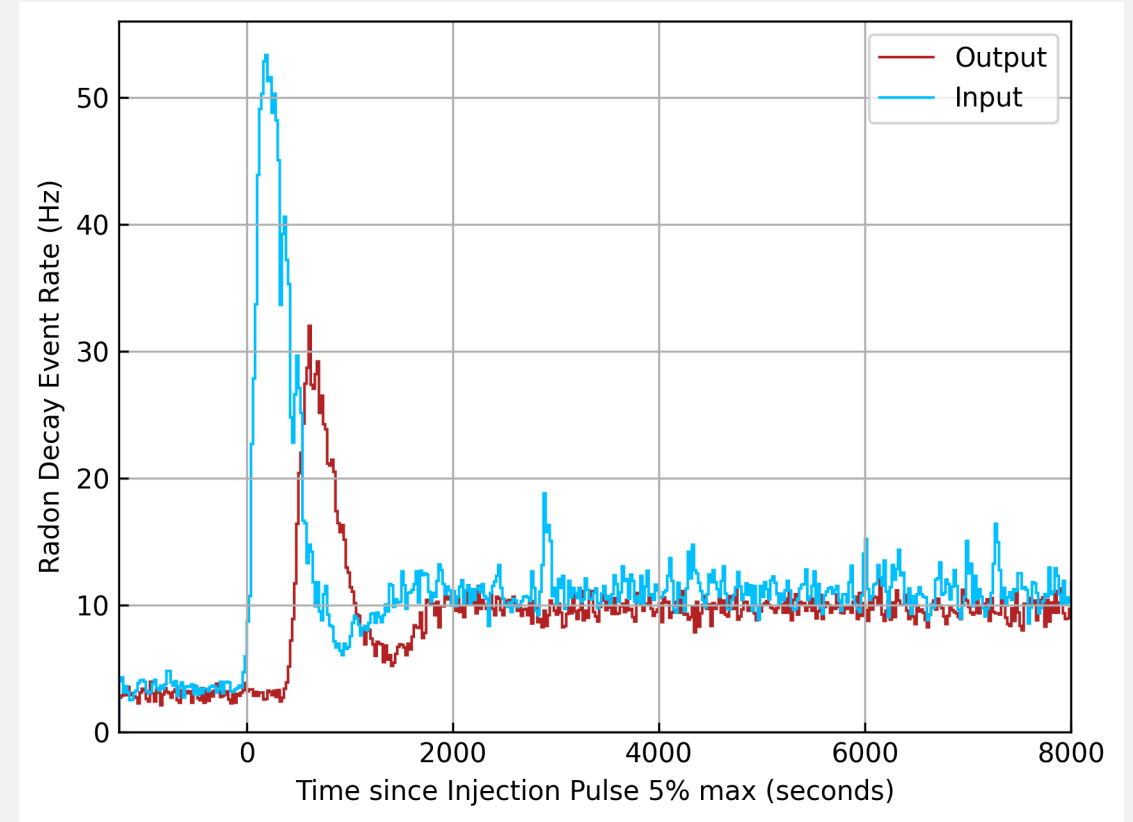
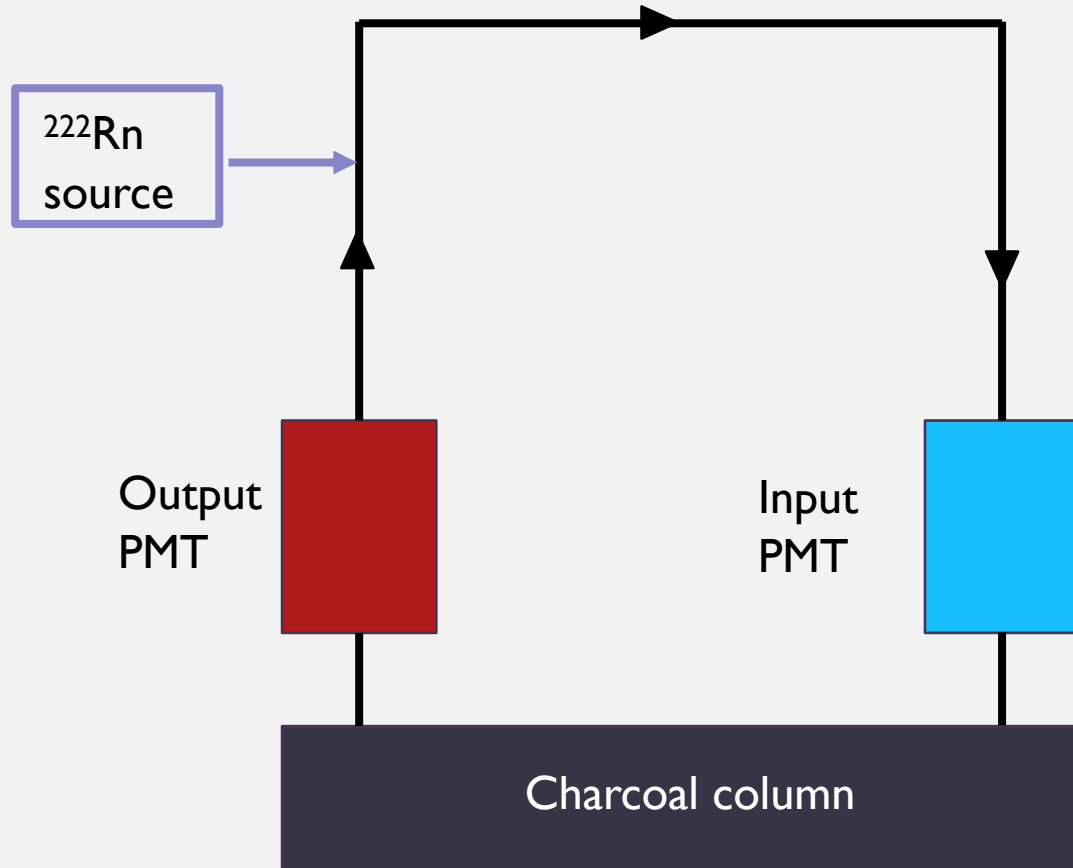
Charcoal mass



Measured adsorption coefficient as a function of temperature
Two runs with different sizes of charcoal beads are shown, to
determine the effect surface area has on k_{gas}

LIQUID MEASUREMENTS

Circulation loop: continuously flows Xe and Rn through the system



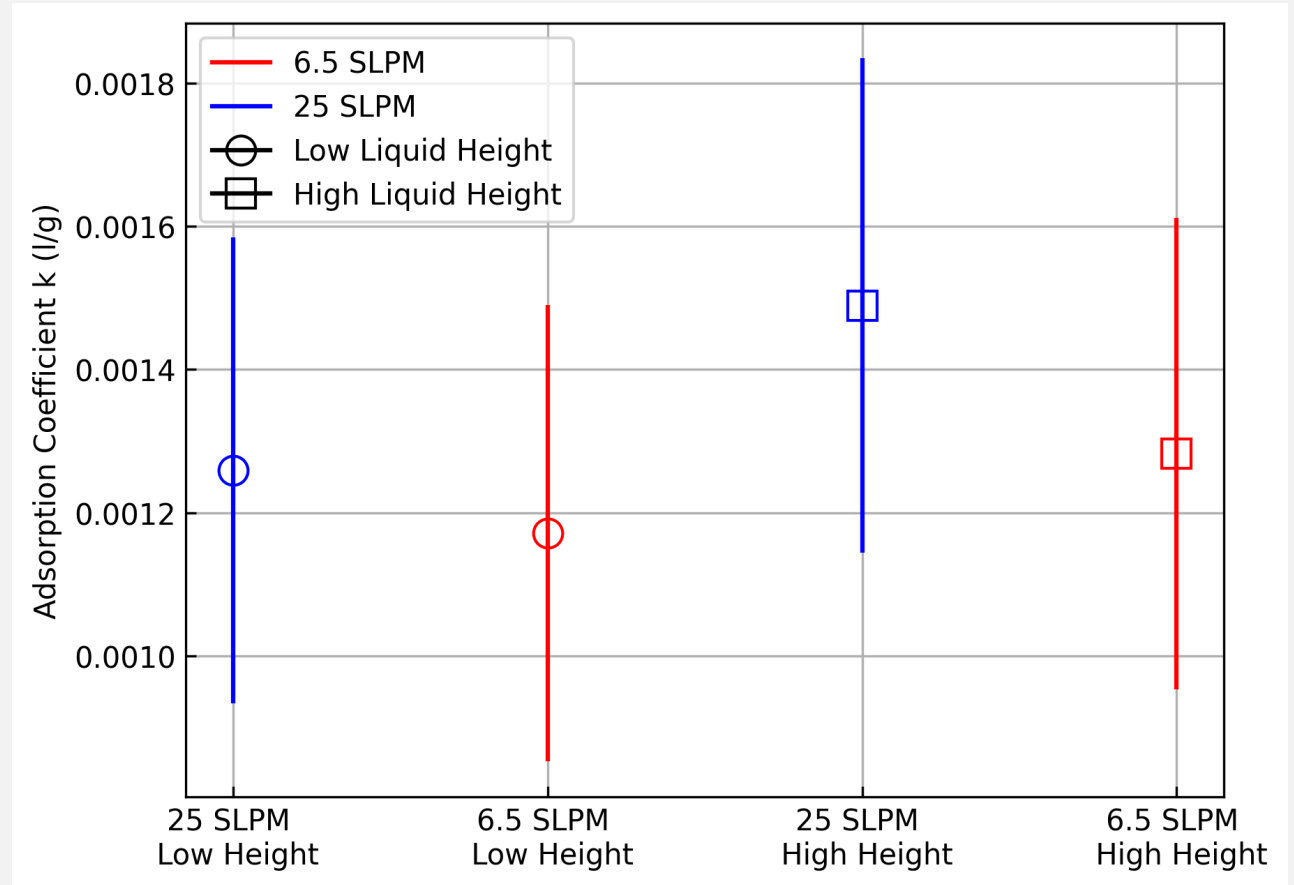
Radon decay rate in the two PMTs

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Red: radon event rate after charcoal column

LIQUID MEASUREMENTS

- Adsorption coefficient 3 orders of magnitude lower than in gas
- No significant effect in changing the flow rate of xenon
- Different liquid heights correspond to different thermodynamic conditions, again no significant effect
- Not enough to make liquid phase radon removal feasible



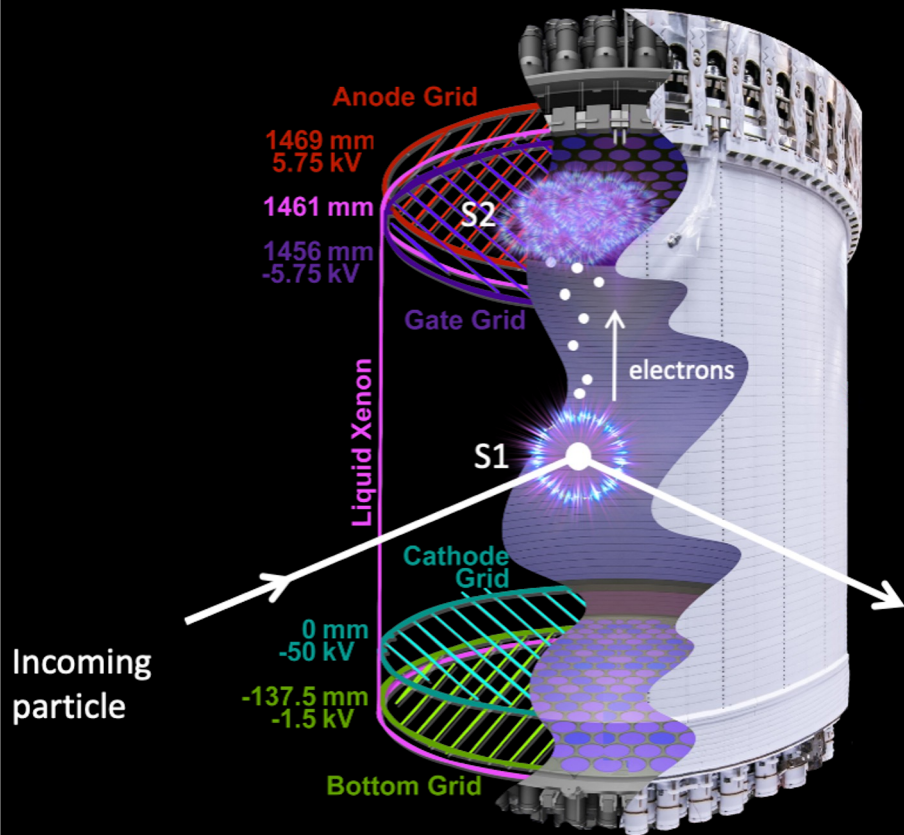
Adsorption coefficient in liquid xenon for a range of flow rates and liquid heights in the inlet liquid Xe reservoir

REFLECTIONS AND FUTURE WORK

- In gas, chromatography performance increases as we lower the temperature
- Previous studies indicated that at these cold gas temperatures, the density of xenon in the column approaches 1/3 of the density of liquid xenon
- This bolstered the idea that liquid phase radon removal would be feasible and effective. This was not supported by our results
- Ongoing simulations to understand these results better
- Further testing of new, synthetic materials as the chromatography medium
- Paper in preparation



Thank you!



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LZ (LUX-ZEPLIN) Collaboration, 38 Institutions

250 scientists, engineers, and technical staff

<https://lz.lbl.gov/>

- Black Hills State University
- Brookhaven National Laboratory
- Brown University
- Center for Underground Physics
- Edinburgh University
- Fermi National Accelerator Lab.
- Imperial College London
- King's College London
- Lawrence Berkeley National Lab.
- Lawrence Livermore National Lab.
- LIP Coimbra
- Northwestern University
- Pennsylvania State University
- Royal Holloway University of London
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