

# Water-based Quantum Dots Liquid Scintillator for Neutrino Physics

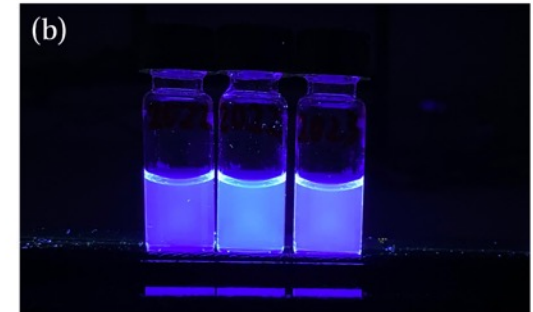
## Outline

1. Quantum Dots
2. Optical measurements
3. Cosmic ray test
4. Next step

## Water-based quantum dots liquid scintillator for particle physics

M. Zhao<sup>1</sup>, M. Taani, J. Cole, B. Crudele,<sup>1</sup> B. Zou, N. Bhuiyan<sup>2</sup>, E. Chowdhury, Y. Duan, S. Fekri,<sup>2</sup> D. Harvey,<sup>3</sup> D. Mitra,<sup>4</sup> O. Raz,<sup>5</sup> A. Thompson,<sup>6</sup> T. Katori<sup>1b</sup> and A. Rakovich<sup>1b\*</sup>

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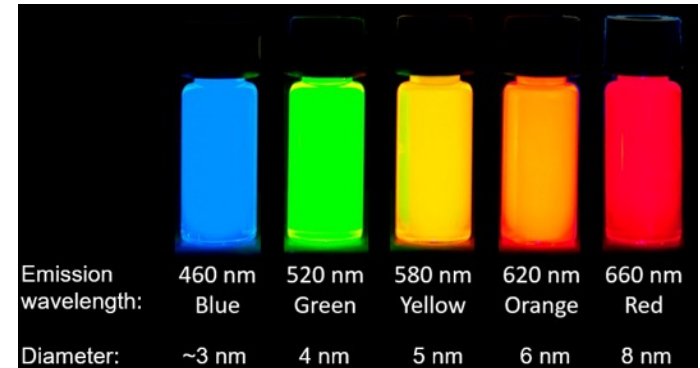
WIN25, Univ. Sussex, Brighton, UK, Jun. 10, 2025

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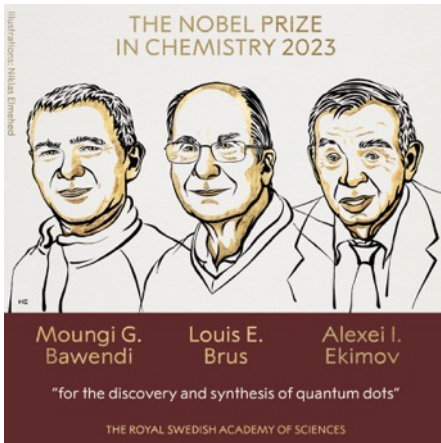
# 1. Quantum Dots

## Semiconductor nano-particle

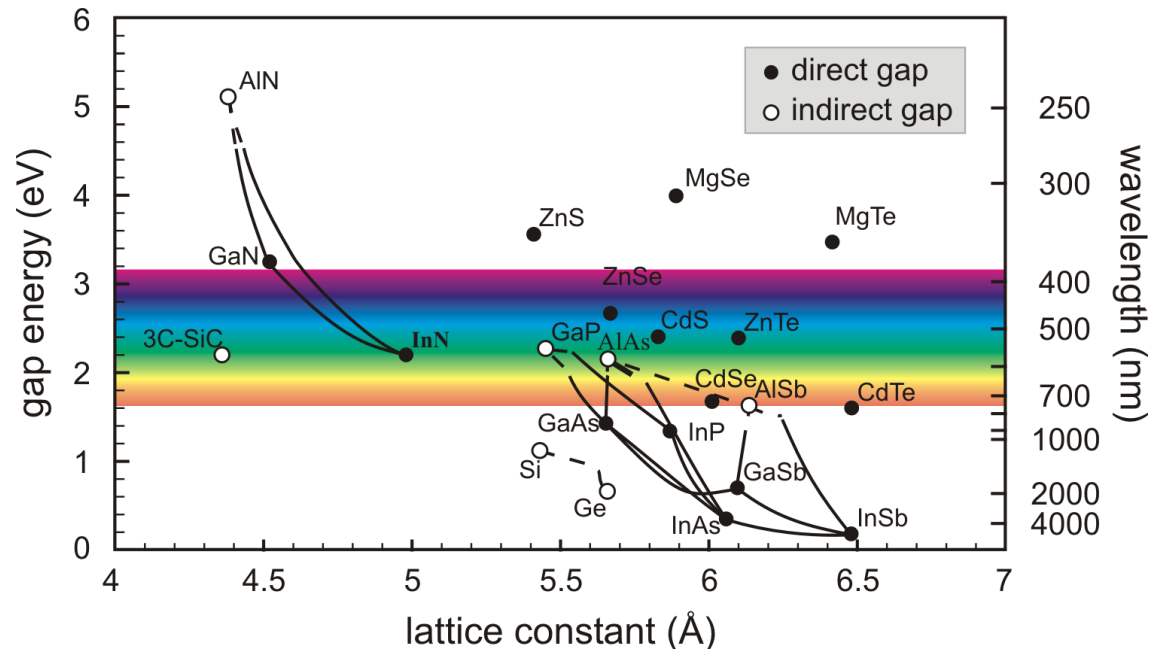
- Smaller than exciton radius (quantum confinement)
- $\sim 2 - 10$  nm, UV to infrared emission
- Tunable sharp emission spectrum
- Large category of choices (core type, size)
- High quantum yield ( $\sim 50\%$ )



CdSe core QDs with CdS/ZnS shell  
<https://www.lateralflows.com/quantum-dots/>



Noble prize in Chemistry (2023)



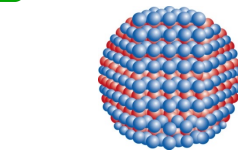
# 1. Quantum Dots

## Semiconductor nano-particle

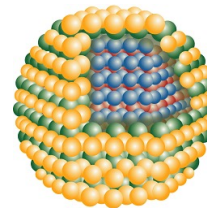
- Smaller than exciton radius (quantum confinement)
- ~2 – 10 nm, UV to infrared emission
- Tunable sharp emission spectrum
- Large category of choices (core type, size)
- High quantum yield (~50%)
- Surface layer engineering to change properties
  - Stability
  - Water solution

## CdS-ZnS Quantum Dots-based water scintillator

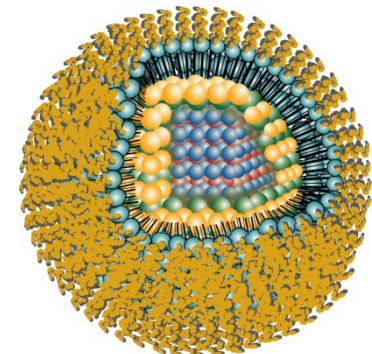
1. High emission (QY ~ 50%) ✓
2. Stability (~ 3 yr) ✓
3. Emission (> 450nm) ⚠
4. Environmental impact ✗
5. Affordable ✗



Quantum dot core (fragile)



QD with shell (stable)



QD with shell and ligand (colloid)

~5nm

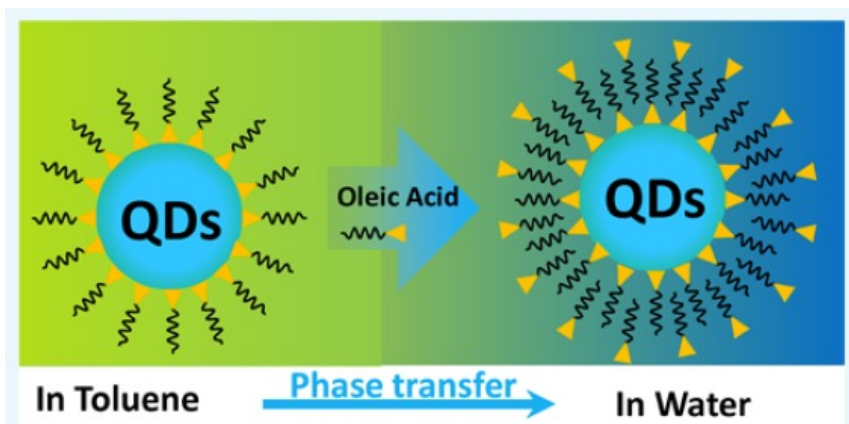
~70nm

Representative surface-capping strategies	Mechanism of interaction	Examples
<b>a</b> Monothiolated caps  $n = 1$ : mercaptoacetic acid $n = 2, 10, 15$ : Benzyl Hydrophilic $\text{HS}(\text{CH}_2)_n(\text{OCH}_2\text{CH}_2)_m\text{OR}$ R = -H, -CH <sub>2</sub> COOH	Dative thiol bond 	Mercaptoacetic acids <sup>4,59</sup> Alkylthiol terminated DNA <sup>61</sup> Thioalkylated oligo-ethylene glycols <sup>62</sup>
<b>b</b> Bidentate thiols  R = -OH R = -(OCH <sub>2</sub> CH <sub>2</sub> ) <sub>n</sub> OH n = 3, 5, -12	Two interactions/ligand 	Dihydrothiolic acid derivatives <sup>26,63</sup>
<b>c</b> Silane shell or box dendrimer  R = -SH, -NH <sub>2</sub> , -PO <sub>3</sub> CH <sub>3</sub>	Hydrophobic Hydrophilic Crosslinked shell 	Mercaptopropyl silanols <sup>3,40</sup> Amine box dendrimers <sup>31</sup>
<b>d</b> Hydrophobic interactions  R = Streptavidin	Hydrophobic Hydrophilic TOP/TOP 	Phosphatidylethanol amine Phosphatidylcholine micelles <sup>39</sup> Modified acrylic acid polymer <sup>33,44,45</sup> Poly(maleic anhydride) alt-1-tetradecene <sup>65</sup>
<b>e</b> Functionalized oligomeric phosphines  R = OH, NH-Streptavidin Hydrophilic	Hydrophobic Hydrophilic 	Oligomeric phosphines <sup>37</sup>
<b>f</b> Amphiphilic triblock copolymer  Hydrophilic Hydrophobic	TOP/TOP 	*Site for EDC-based antibody conjugation Hydrophilic Amphiphilic triblock copolymer <sup>66</sup>

# 1. Phase transfer

## How to make water-based quantum dot solution

1. Dissolve CdS-ZnS QDs (10mg) in toluene
2. Evaporate toluene
3. Add hexane, sonic bath 1min
4. Add oleic acid
5. Add water
6. Sonic bath
7. Shake
8. Sonic bath, shake ... (1hr)
9. Leave in dark 24 hours

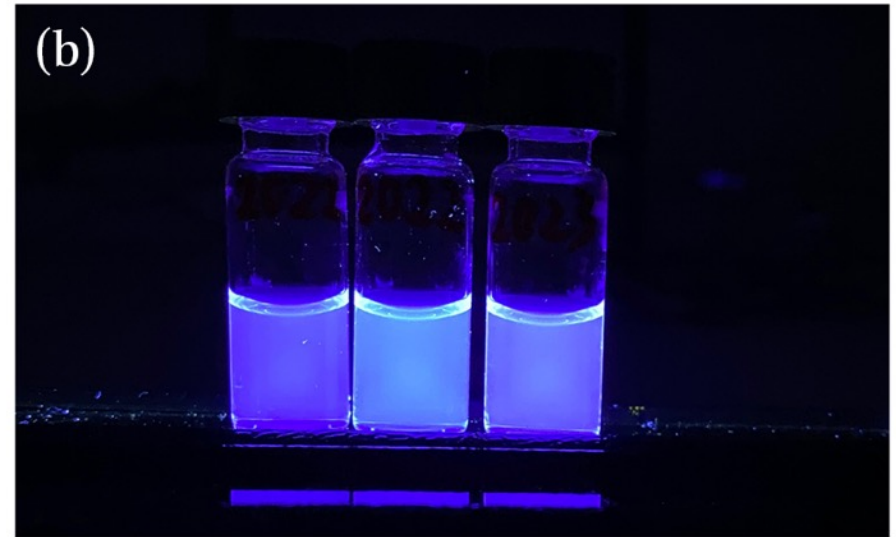


Chemistry is complicated

# 1. Water-based quantum dots liquid scintillator

We made 3 water-based QDs sample

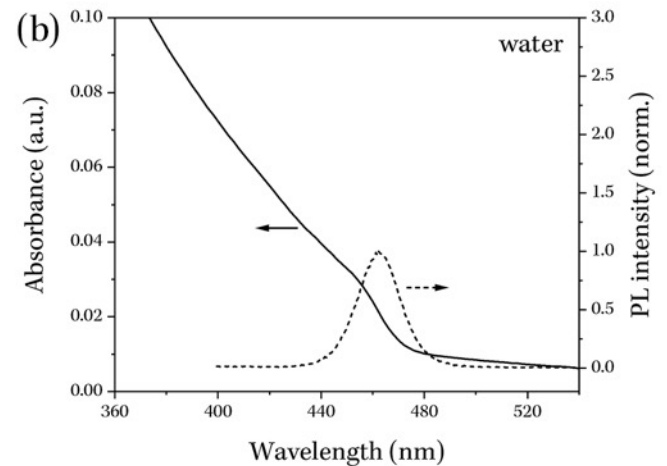
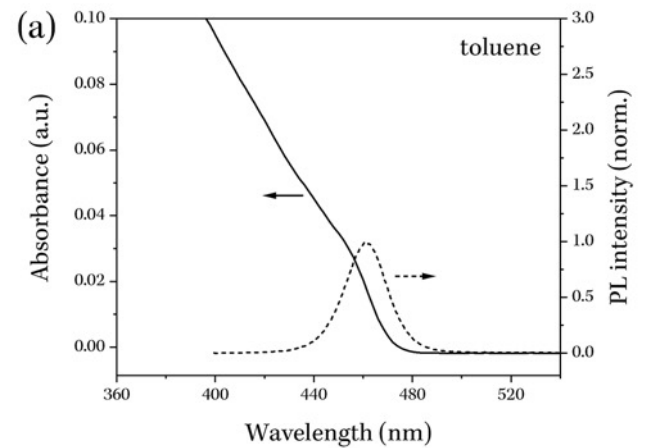
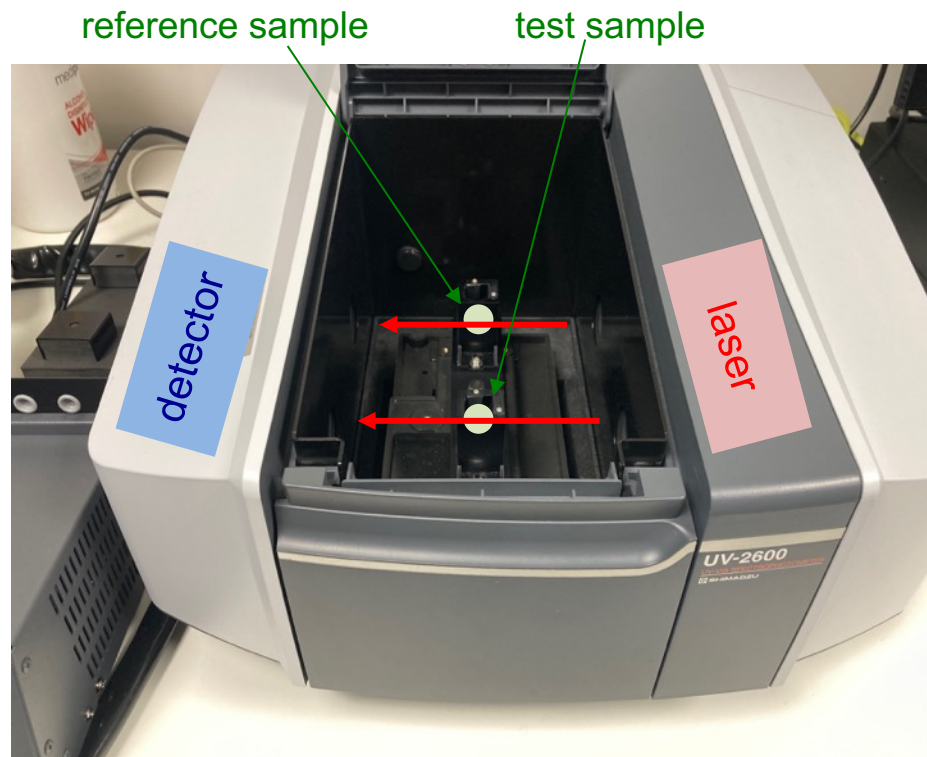
- Ageing tests
  - Quantum dots are stable over 3 years
  - No sign of agglomeration, but surface configuration is not stable
    - Need further study of phase transfer



## 2. Absorbance measurement

### Shimadzu UV-2600i spectrometer

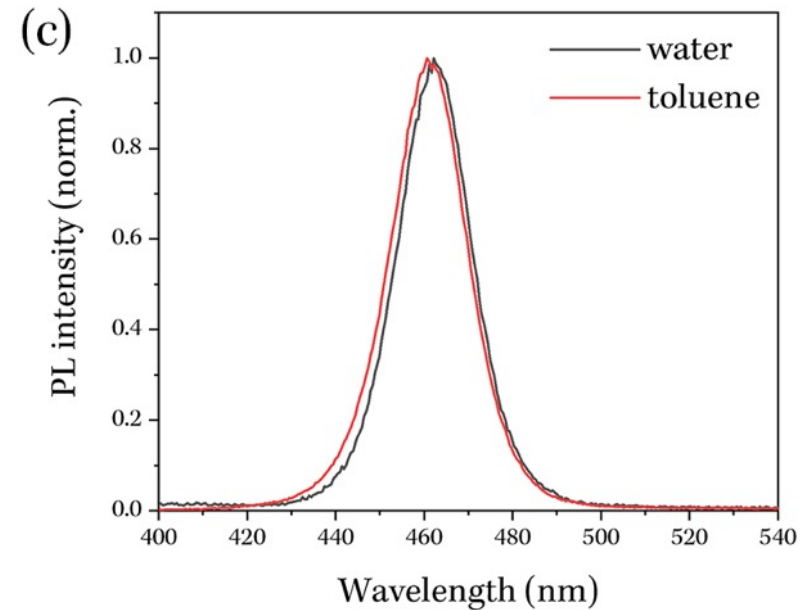
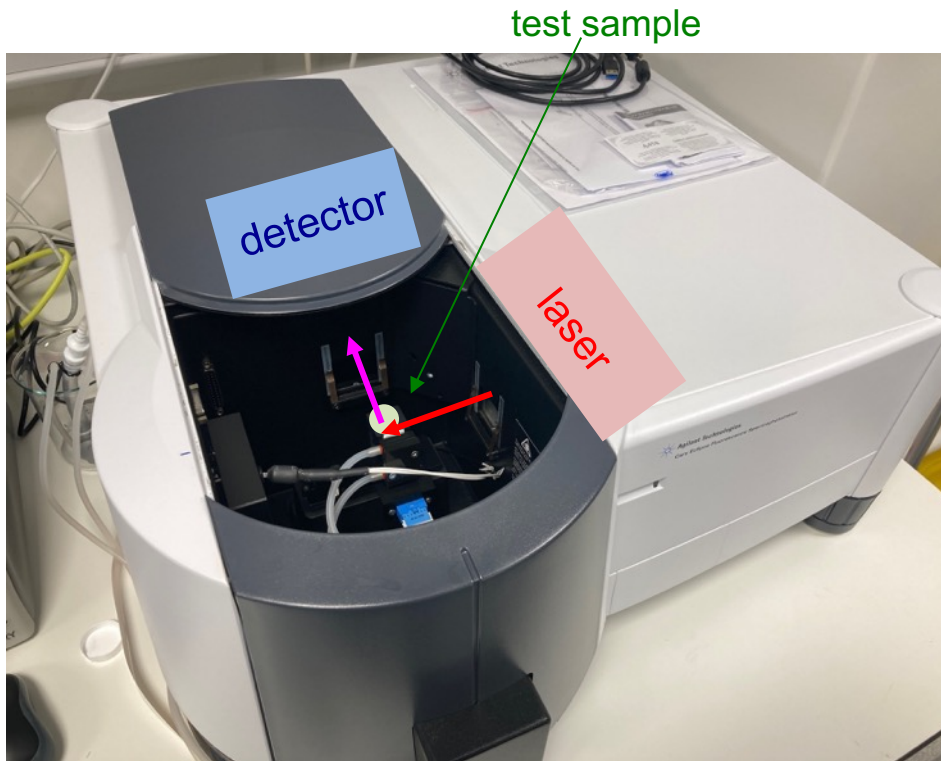
- toluene QDs and water QDs show the same absorption peak ( $\sim 460\text{nm}$ )
- No significance change of QD core by phase transfer



## 2. Fluorescence measurement

### Agilent Technologies Cary Eclipse Fluorescence Spectrometer

- toluene QDs and water QDs show the same emission peak
- No significance change of QD core by phase transfer



## 2. Photoluminescence quantum yield (PLQY)

Measure absorbance and fluorescence with different concentrations

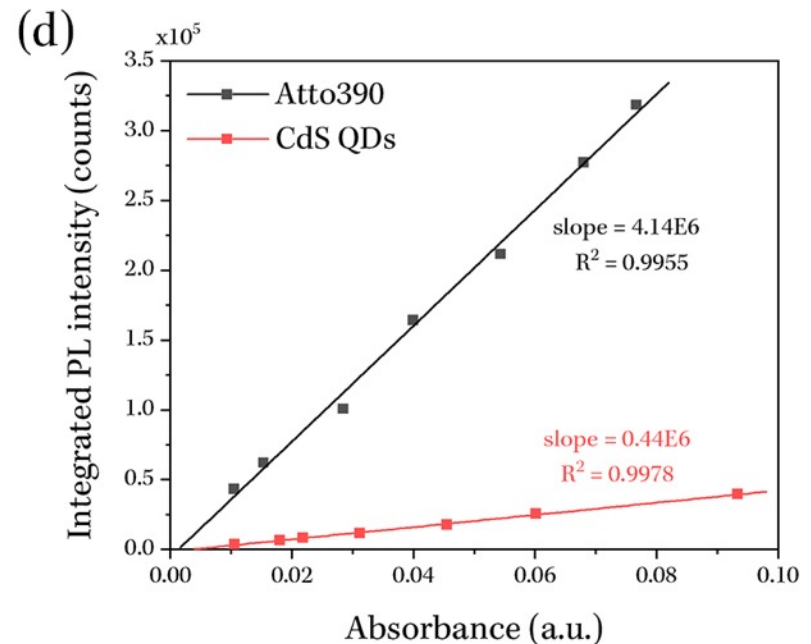
- Integrated PL intensity is a linear function of absorbance. Both integrated PL intensity and absorbance are measured with different concentration of the sample

$$I(A) = I_0 + K \cdot A$$

- Obtained slope “K” is compared with the reference dye Atto390

- Measured PLQY is lower than the company value (~50%) suggesting the phase transfer lose some PLQY

$$\text{PLQY} = 9.5 \pm 0.5\%$$



## 2. Dynamic light scattering (DLS) measurement

Malvern instruments Zetasizer Nano

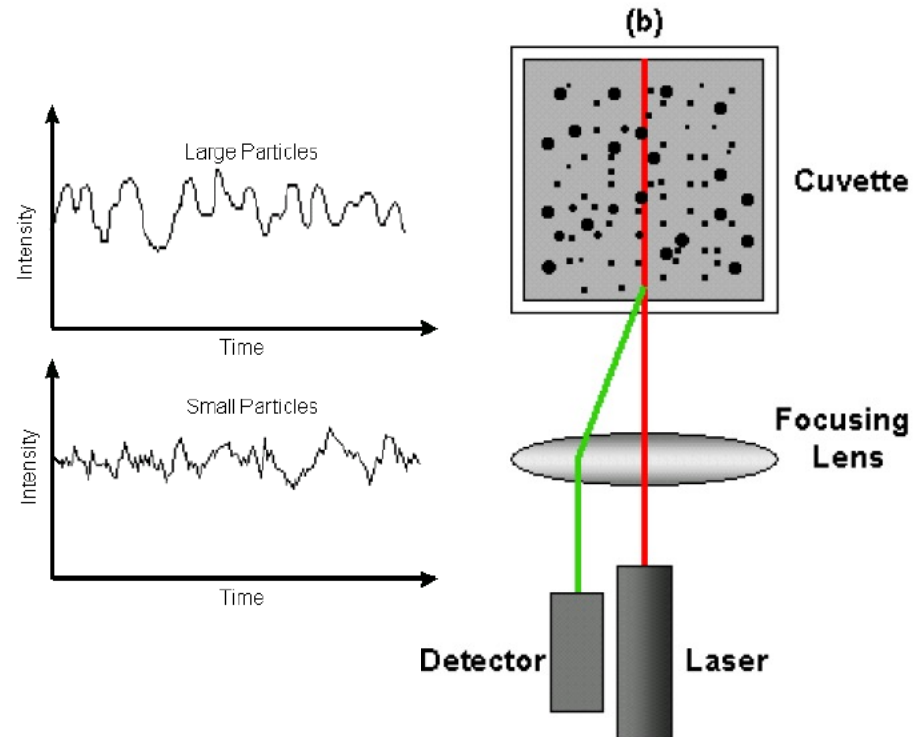
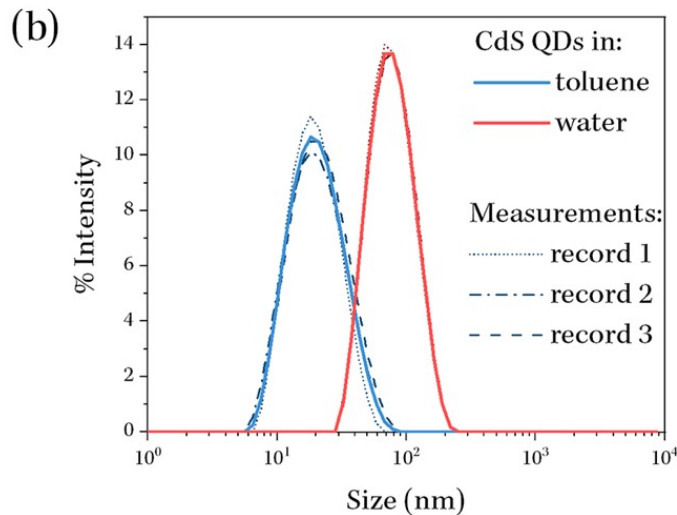
- Measure the size of QDs from the back-scattered light
- Brownian motion gives time-dependent data

Hydrodynamic diameter  $d_H$

$$D_H = \frac{k_B T}{3\pi\eta D}$$

$$D_H^{\text{Toluene}} = 25 \pm 9 \text{ nm}$$

$$D_H^{\text{Water}} = 71.6 \pm 0.3 \text{ nm}$$



<https://www.malvernpanalytical.com/en/products/technology/light-scattering/dynamic-light-scattering>

## 2. Transmission electron microscope (TEM)

Malvern instruments Zetasizer Nano

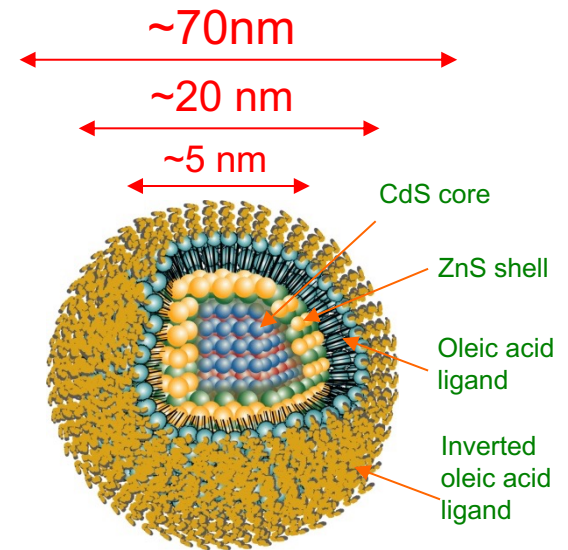
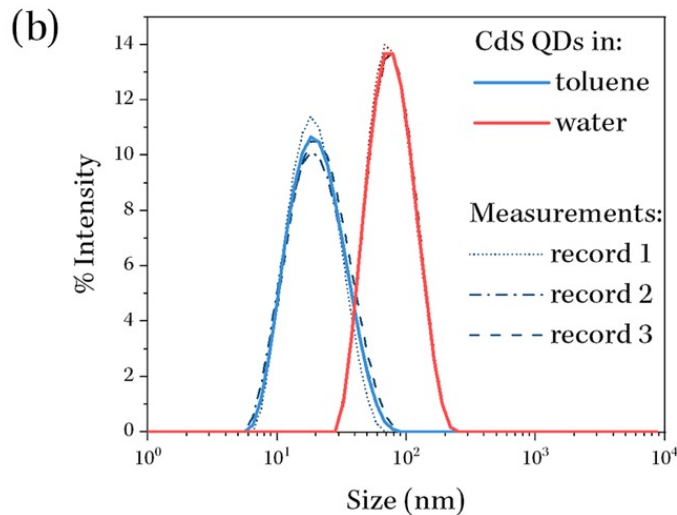
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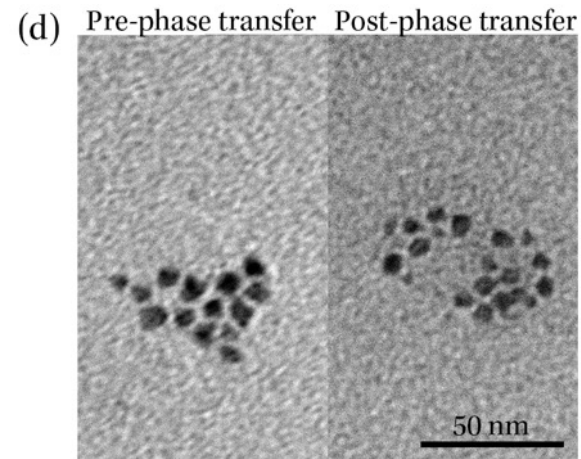


QD size is measured by TEM

- SEM, AFM, didn't work well

$$D_{\text{TEM}}^{\text{Toluene}} = 6.4 \pm 1.0 \text{ nm}$$

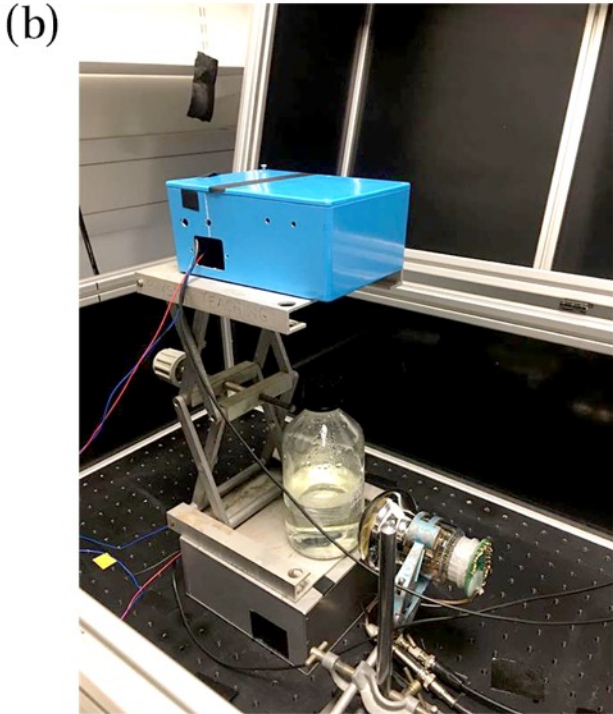
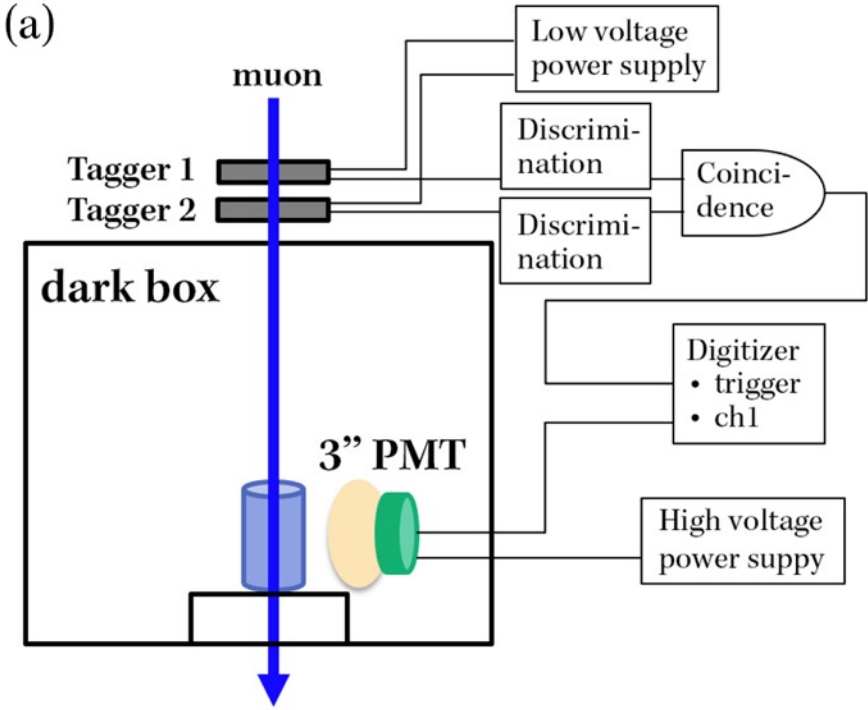
$$D_{\text{TEM}}^{\text{Water}} = 6.4 \pm 1.2 \text{ nm}$$



# 3. Cosmic ray test

## Set up

- Cosmic ray triggers (plastic scintillator + SiPM)
- 3-inch NNVT PMT
- CAEN 250MS/s digitizer



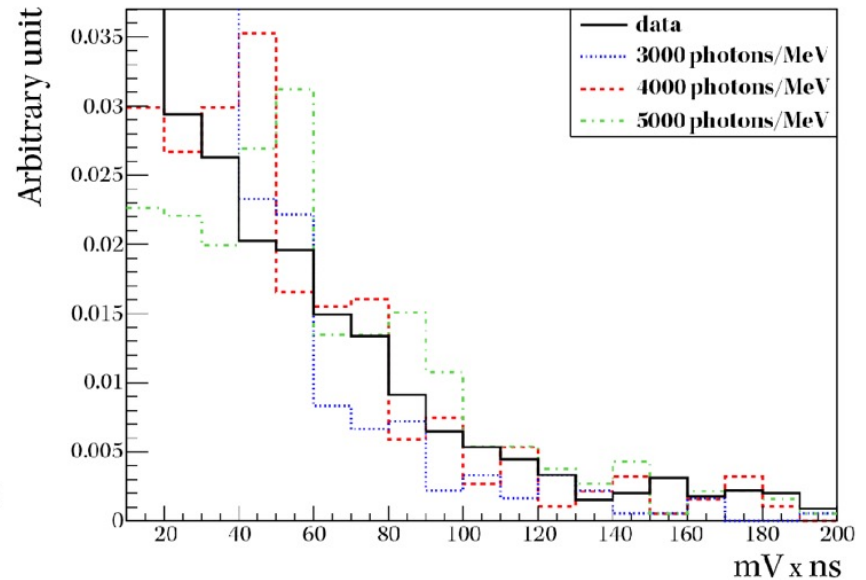
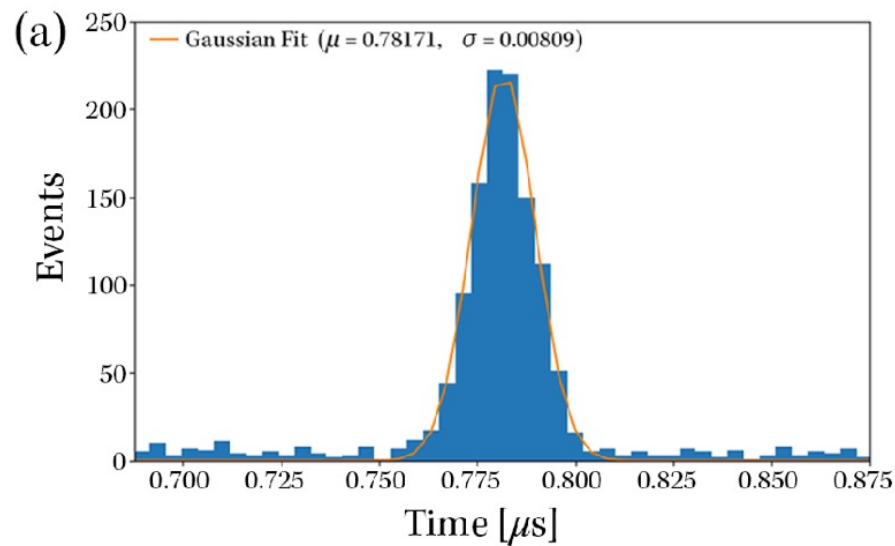
### 3. Cosmic ray test

#### Timing data

- Quantum dots decay time is around 5ns
- Too fast to measure by 250MS/s (4ns bin) digitizer ( $\tau < 8\text{ns}$ )

#### Charge data

- Data seems to have several 1000s photons / MeV
- Comparable emission strength with modern organic scintillator



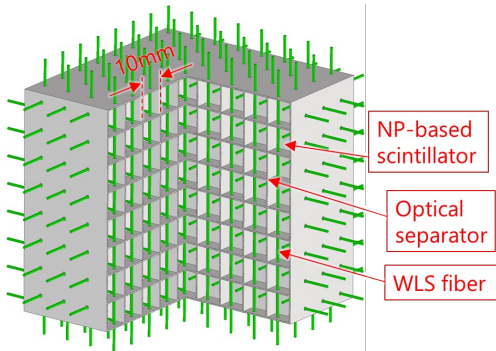
# 4. Next step

Nanocrystal water liquid scintillator works, but...

- 1L WbQDLS ~£3k → ~£3M for 1 ton detector
- water is safe, but many nanocrystal is not safe

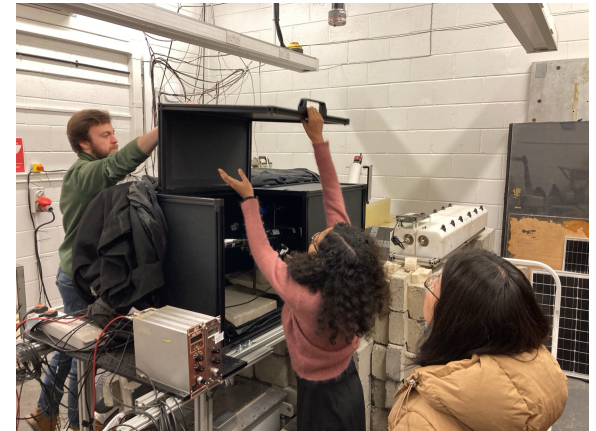
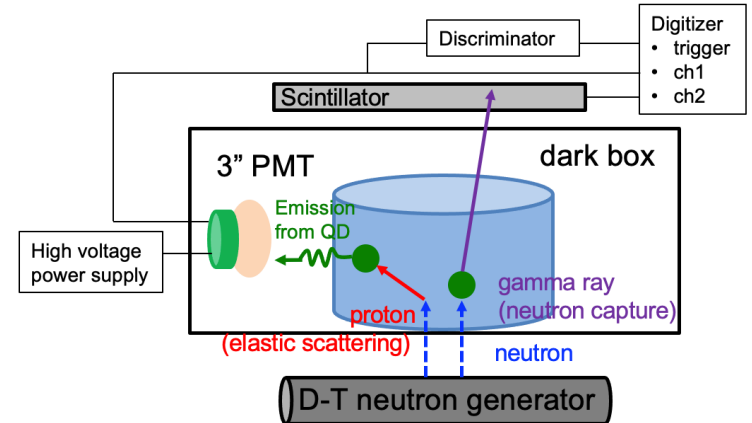
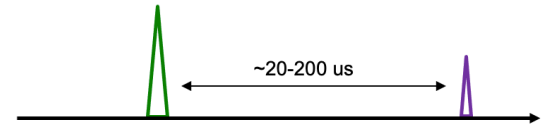
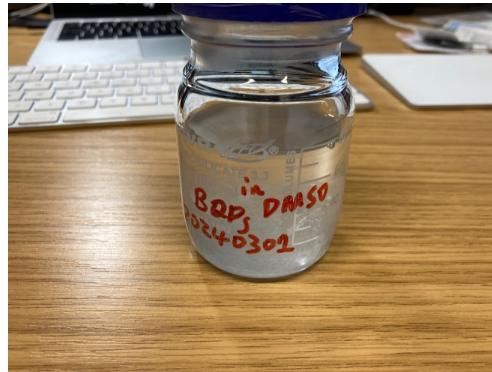
Future: Sustainable nanocrystal water scintillator

- no heavy metal: cheap, safe
- environment-friendly: green synthesis, recyclable
- neutron PID ability



Water-based liquid scintillator for Hyper-Kamiokande new new near detector (Kikawa, Kyoto)

Boron QD sample



QD neutron capture test (Sheffield fast pulse D-T gun)

## 5. Conclusions

Water-based liquid scintillator is successfully made using CdS-ZnS quantum dots.

Water QDs photoluminescence quantum yield is  $9.5 \pm 0.5\%$ , lower than the company value ( $\sim 50\%$ ).

Hydrodynamic diameter is measured by DLS, and water QDs ( $\sim 70\text{nm}$ ) is larger than toluene QDs ( $\sim 20\text{nm}$ ) as expected.

Scintillation response is measured from cosmic muons. Time constant seems very fast as expected ( $< 8\text{ns}$ ), and charge response suggests photon yield is comparable as typical organic liquid scintillators.

Sustainable nanocrystal water scintillator is the next step.

# Thank you for your attention!

# Backup

# 1. Quantum Dots

## Display

- Tunable, sharp emission

## Solar cell

- Multiple exciton generation

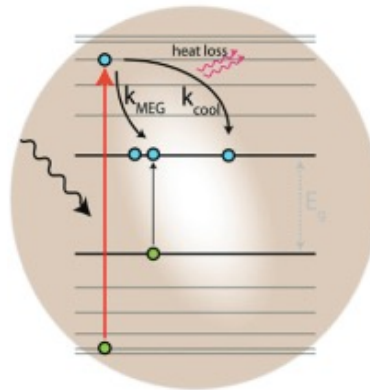
## Biometric marker

- Water soluble
- Infrared
- Single molecular tracking

## Quantum computer

- Spin qubit quantum computer

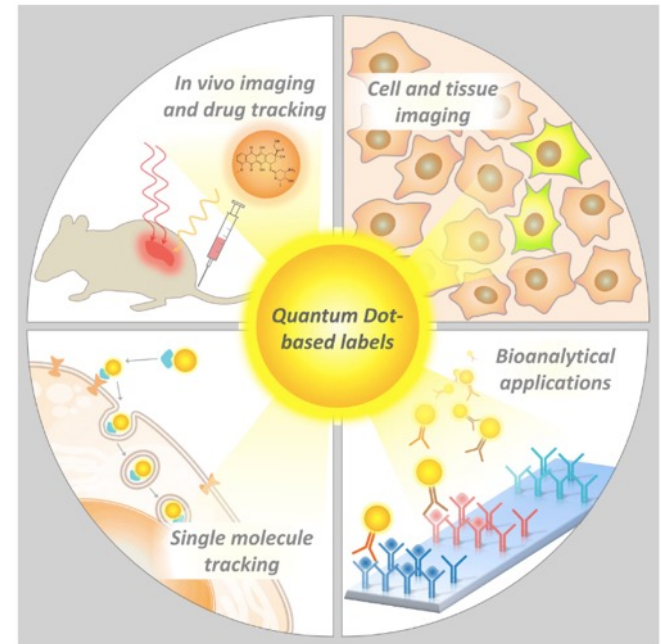
etc



Energ. Environ. Sci. 13 (2020) 1347



Samsung QLED smart TV



ChemBioChem 17 (2016) 2103

# 1. Brus equation

## Masses

- electron mass,  $m_e$
- effective electron mass,  $\sim 0.1m_e$  (CdSe)
- effective hole mass,  $\sim 0.45m_e$
- effective exciton mass,  $\mu \sim 0.08m_e$

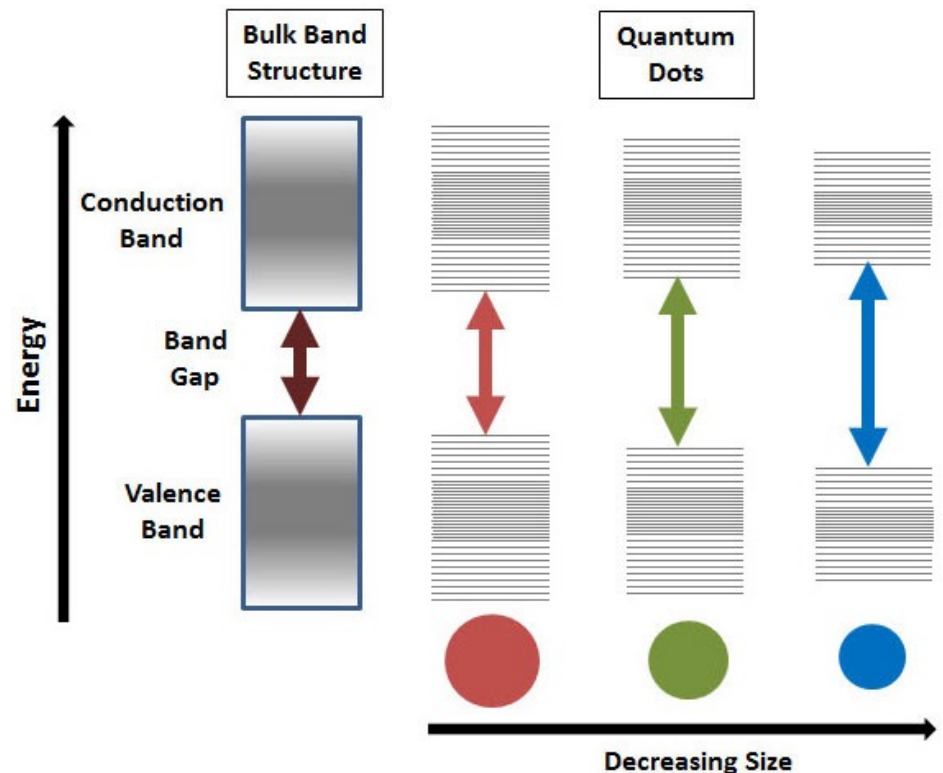
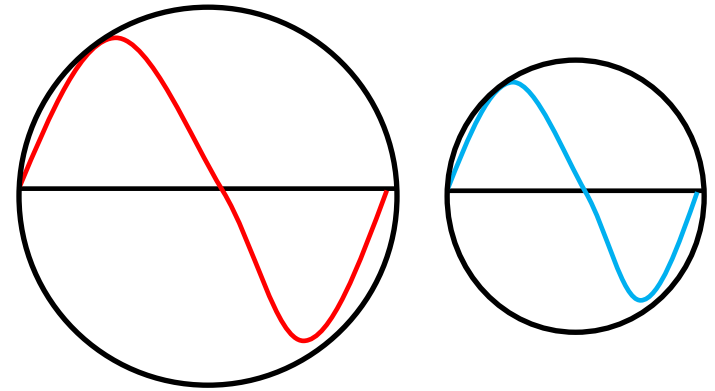
## Size

- Exciton radius  $r \sim 6$  nm
- Exciton diameter  $2r \sim 12$  nm
- Quantum dots diameter: 2-10 nm

## Band gap (Brus equation)

$$E_{QD} = E_{gap} + \frac{\hbar^2\pi^2}{2r^2\mu}$$

- Band gap is wider in quantum dots
- Quantum confinement



## 2. Fluorescence measurement

### Agilent Technologies Cary Eclipse Fluorescence Spectrometer

- toluene QDs and water QDs show the same emission peak
- No significance change of QD core by phase transfer

### Photoluminescence quantum yield (PLQY)

$$I(A) = I_o + K \cdot A$$

- Integrated PL intensity is a linear function of absorbance. Both integrated PL intensity and absorbance are measured with different concentration of the sample
- Obtained slope “K” is compared with the reference dye Atto390
- Measured PLQY is lower than the company value (~50%) suggesting the phase transfer lose some PLQY

$$\text{PLQY} = 9.5 \pm 0.5\%$$

### Extinction peak

$$A(\lambda) = A_0 + A_{max} \cdot \exp \left[ -0.5 \left( \frac{\lambda - \lambda_c}{w} \right)^2 \right]$$

### Empirical law with extinction peak

- mean QD diameter D  
 $D_{\text{Empirical}} = 5.22 \pm 0.01 \text{ nm}$
- concentration C  
 $C = 42 \pm 1 \text{ nM}$

### Typical organic scintillator (20% PPO)

- concentration ~ 4-30 mM
- cross-section area ~ 0.5-1 nm<sup>2</sup>

## 2. SEM, AFM, TEM

### Scanning electron microscope (SEM)

- Sample is dried
- Gold is vacuum evaporated to make it conductive
- Not sensitive to QD size ( $\sim 20\text{nm}$ )

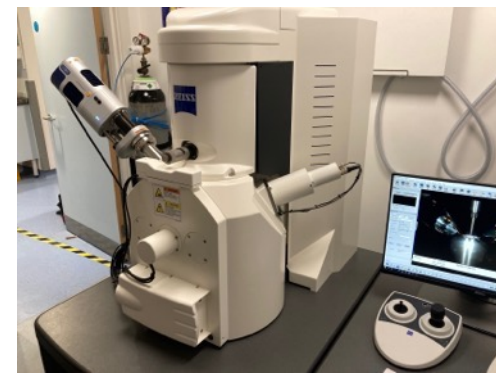
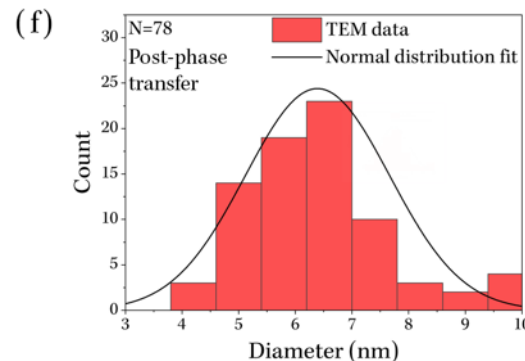
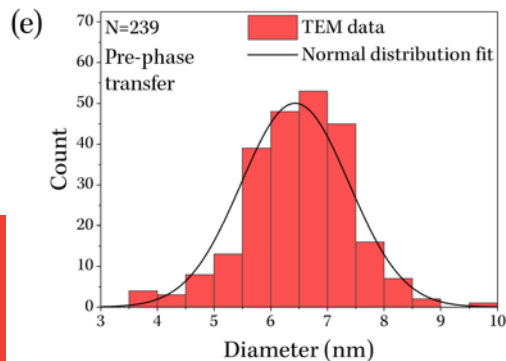
### Atomic force microscope (AFM)

- Sample is dried
- Not sensitive to X-Y because of large tip
- Z measurement may be good?

### Transmission electron microscope (TEM)

$$D_{\text{TEM}}^{\text{Toluene}} = 6.4 \pm 1.0 \text{ nm}$$

$$D_{\text{TEM}}^{\text{Water}} = 6.4 \pm 1.2 \text{ nm}$$

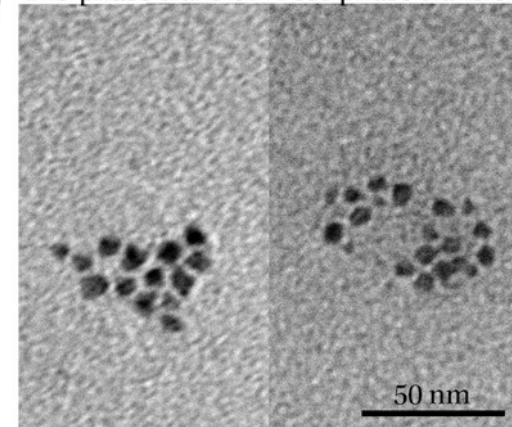


Scanning electron microscope (SEM)



Atomic force microscope (AFM)

(d) Pre-phase transfer Post-phase transfer



TEM image

2025/04/03

19

## 2. Ageing

Sample		PLQY (%)	$D_h$ (nm)	$\zeta$ (mV)
2021 WbQD	as prepared	$15.6 \pm 1.2$	$52.2 \pm 0.3$	not available
	in 2023	$8.8 \pm 0.8$	$48.7 \pm 0.1$	$-13.8 \pm 0.8$
2022 WbQD	as prepared	$8.8 \pm 1.5$	$60 \pm 1$	$-24 \pm 2$
	in 2023	$14.5 \pm 0.7$	$55.7 \pm 0.3$	$-47 \pm 2$
2023 WbQD	as prepared	$9.5 \pm 0.5$	$71.6 \pm 0.3$	$-24.6 \pm 0.7$

### Absorbance and fluorescence

- No change over the period of 3 years

### Size

- Slight decrease of the size (a few nm)
- Likely not losing layers, but re-organization of surface structure
- Zeta potential (surface charge) is changing over time, supporting this interpretation
- No sign of agglomeration

### Photoluminescence quantum yield (PLQY)

- Large sample variation makes difficult to conclude

It seems water QDs are stable, but without ageing information of PLQY it's hard to say if they are suitable for particle detectors (~several years operation)

### 3. Cosmic ray test

#### Timing data

- Quantum dots decay time is around 5ns
- Too fast to measure by 250MS/s (4ns bin) digitizer ( $\tau < 8\text{ns}$ )

#### Charge data

- Need simulation to extract information

